



Carbon dots as multifunctional platform for intracellular pH sensing and bioimaging. *In vitro* and *in vivo* studies

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ABSTRACT

In the present work we have obtained carbon dots (CDs) from glutathione and modified their surface with both β -Cyclodextrin and methyl-3-(4-(2-(5-((methylsulfonyl)oxy)pentyl)-1,3-dioxo-2,3-dihydro-1H-benzo[de]isoquinolin-6-yl)piperazin-1-yl)propanoate (FP), a pH-sensitive molecule. β -Cyclodextrin acted as a glucose source, which facilitated the entrance of the CDs within the cells and, eventually, can be used as drug-carrier to deploy bioactive compounds in the cytoplasm. FP fluorescence varied according to the pH of the medium and can be used to distinguish cancer cells due to the Warburg effect.

Cytotoxicity studies revealed that, contrarily to simple CDs, carbon nanodots modified with β -cyclodextrins had an impact in cell cycle phase distribution, with no alterations of cell morphology. Likewise, CDs modified with β -cyclodextrin and FP can be used to distinguish cancer cells from non-cancerous ones due to an increase in fluorescent emission. Furthermore, this modified CDs can also be used to obtain high contrast *in vitro* images as well as *in vivo* images. These results demonstrate the potential applications of the synthesized nanoparticles both in bioimaging and as theranostic sensor and actuator.

1. Introduction

Research on carbon-based nanoparticles, particularly carbon-dots (CDs), has received a strong boost due to the particular properties of these materials, such as their interesting luminescent characteristics (high quantum yields, photostability, tunable emission wavelength), their high solubility and stability in water, biodegradability and their very low or even null cytotoxicity [1]. In addition to the possibility of their obtaining through easy, low-cost and “green” methods, these materials show the advantage of possessing surface functional groups which can be further modified for particular applications [2,3]. The surface functionalization and/or passivation process during synthesis is one of the main methods used to control the fluorescent properties of CDs, thanks to the modification of the surface emissive sites [4]. Due to their unique photoluminescence properties, carbon dots offer a potential alternative to the traditional fluorescent dyes in applications such as

intra-cell probes, in-vivo labelling and/or bioimaging [5,6]. In fact, over the past ten years scientists have made great strides towards the use of raw native CDs as (bio)sensing materials while the use of hybrid CDs composites are still scarce. So, for example, it has been reported that the fluorescent dye Rhodamine B can be attached to the surface of CDs to develop a ratiometric probes to detect Fe^{3+} in lake and tap waters [7] or histidine [8]. Wang et al. [9] dispersed CDs modified with Rhodamine B into agarose gels to develop a hydrogel sensing film for semiquantitative visual detection of Hg(II) under a UV lamp. Recently, Yu et al. [10] described the “off-on-off” detection of CN⁻, hydroquinone and ascorbic acid using CDs modified with coumarin.

On the other hand, the CDs modification with cyclodextrins (CyDs) for their use in (bio)analytical applications using “turn-off”, “turn-on”, “turn-off-on” and host-guest recognition approaches for selected bio-analytes has also been described [11–14].

In the present work we have designed a novel system based on

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glutathione-based carbon dots (CD) capped with both pH (fluorescent 1,8-naphthylamide-based dye, **FP**) and drug delivery (β -cyclodextrin, β CyD) functions (**FP-CD- β CyD**). As cancer cells excrete lactic acid, cancerous environments have acidic pH values [15]. The fluorescent **FP** dye exhibits pH-dependent fluorescence emission [16] and we leveraged this function for the *in-vitro* determination of intracellular pH in the melanoma-derived murine cell line B16F10 in comparison with noncancerous human mammary fibroblasts (HMF). In addition, *in vivo* analysis of body CD distribution was also performed and analyzed by *in vivo* fluorescence imaging. Our results indicate that, the β -cyclodextrin moiety can act as a potential carrier for drug delivery; and that CDs, may act as a *bone fide* tool as bioimaging reporters. Therefore, the main objective of this work is to characterize **FP-CD- β CyD**s for optical pH sensing and imaging application for *in vitro* and *in vivo* bioanalyses.

2. Experimental

2.1. Reagents and instrumentations

2.1.1. Reagents and materials

N-(3-Dimethylaminopropyl)-N'-ethylcarbodiimide hydrochloride (EDC·HCl), sodium N-hydroxysulfosuccinimide (sulfo-NHS), 4-aminophenylboronic acid (APBA), Tris(hydroxymethyl)aminomethane (TRIS), β -cyclodextrin (β CyD), ethylenediaminetetraacetic acid (EDTA), formaldehyde, 2-[4-(2-hydroxyethyl)-1-piperazinyl-(1)]ethanesulfonic (HEPES), propidium iodide, Triton X-100, NaCl, KCl, CaCl₂, MgSO₄, NaH₂PO₄, ethanol, pyridine and sodium acetate were purchased from Sigma. Phosphate buffer solution 1X pH 7.4 (PBS), trypsin and glucose were purchased from Merck. Milli-Q water (MQW) was used throughout the experimental. 3.5 kDa Membrane Dialysis Tubes were from Merck.

Dulbecco's Modified Eagle Medium (DMEM), fetal bovine serum (FBS), penicillin, streptomycin, glutamine, Sorting buffer (PBS, 5 mM EDTA, 25 mM HEPES, pH 7.0), and ribonuclease A were purchased from Gibco. Xylazine (Rompun) was purchased from Bayer and Ketamine (Imalgene) from Boehringer Ingelheim. Kit Cell Titer 96 Non-Radioactive Cell Proliferation Assay was from Promega. 35 μ m filter labbox, Silicone culture-insert 2 well and μ -Dish 35 were supplied by Ibidi®.

2.1.2. Spectrophotometric instrumentation

FTIR spectra were recorded in a Varian 670 spectrometer, using 4 cm⁻¹ at 2000 cm⁻¹ resolution and 32 scans per sample. The photoluminescence spectra of molecular compounds and nanomaterials were recorded with an Edinburgh Instruments FLSP-920 spectrofluorimeter. Fluorescence quantum yield was measured on Edinburgh Instruments FS-5 with an integrating sphere using solutions in such concentrations that their absorbance at the excitation wavelength was kept below 0.1. Absorbance measurements were carried out in a Biotek PowerWave XS multiplayer VIS / UV absorption spectrophotometer using 96-well plates.

Fluorescence intensity of nanomaterials in cultured cells was measured using a Biotek Hybrid Synergy H4 multiplate fluorimeter by exciting at 350 nm and emission at 430 nm. *In vivo* fluorescence bio-images were acquired on a Xenogen IVIS Imaging System (MA, USA) at 520 nm (GFP filter) using 430 nm as excitation wavelength, exposure of 1 min and pixel binning set at 4. Images from tissue samples were obtained in a Leica TCS-SP8X confocal fluorescence microscope (Heidelberg, Germany) using a 405 nm laser diode and the emitted fluorescence detected with a photomultiplier tube detector.

Detection of fluorescent nanoparticles within living cells was analysed on a Beckman Coulter Cytotflex S cytometer and their effect on cell cycle phase distribution was carried out on a Beckman Coulter Cytomics FC500 cytometer (CA, USA). Images from cultured cells were obtained in a Zeiss Axiovert 200 m inverted microscope using 3 excitation wavelengths 365, 436 and 546 nm.

2.2. Synthesis of the new nanomaterial

Raw carbon dots (CDs) were synthesized as described by Díaz-Faes et al. [17]. Functionalization of CDs with β -cyclodextrin was carried out as reported by S. Mondal et al. [18] with some modifications.

a Functionalization of CDs with 4-aminophenylboronic acid, **CD-APBA**

10 mL of Milli-Q water were added to a Schlenk flask containing 0.4 g of CDs and the mixture was sonicated in an ultrasound bath for 18 min. Then, the flask was connected to a stream of N₂ and sonication continued for 2 more min. 38.4 mg (0.2 mmol) of N-(3-dimethylaminopropyl)-N'-ethylcarbodiimide hydrochloride (EDC·HCl) and 43.4 mg (0.2 mmol) of sodium N-hydroxysulfosuccinimide (sulfo-NHS) were added and allowed to react 1 h with constant stirring at 360 rpm, at room temperature under a N₂ atmosphere. Finally, 31.6 mg (0.2 mmol) of 4-aminophenylboronic acid were introduced into the foregoing reaction. The mixture was continuously stirred at room temperature for 48 h. Lastly, the solvent was evaporated on a rotatory evaporator under reduced pressure. The visible orange solid was purified by dispersion in 5 mL of ethanol, centrifugation at 5000 r.p.m for 10 min and removal of the supernatant. The process was repeated 3 times to obtain the **CD-APBA**. 0.13 g of **CD-APBA** were obtained.

b Functionalization of **CD-APBA** with β cyclodextrin, **CD- β CyD**

0.13 g of **CD-APBA** were dispersed in 5 mL of 0.1 M tris(hydroxymethyl)aminomethane (TRIS) buffer at pH 10.4. Then, 1 mL of a saturated solution of β CyD in 0.1 M TRIS buffer pH 10.4 was added and the mixture was continuously stirred at 360 rpm for 12 h. Thereafter, the solvent was removed using a rotatory evaporator under reduced pressure. The obtained solid was dispersed in 5 mL of ethanol in an ultrasonic bath, centrifuged at 5000 rpm and the supernatant discarded. This sequence was repeated three times and 0.08 g of **CD- β CyD**, a light brown solid, was finally obtained.

c Functionalization of **CD- β CyD** with methyl-3-(4-(2-(5-((methylsulfonyl)oxy)pentyl)-1,3-dioxo-2,3-dihydro-1H-benzo[de]isoquinolin-6-yl)piperazin-1-yl)propanoate, **FP-CD- β CyD**

The surface functionalization of CD in the ensemble **CD- β CyD** with a pH-sensitive fluorescent probe (**FP**, methyl-3-(4-(2-(5-((methylsulfonyl)oxy)pentyl)-1,3-dioxo-2,3-dihydro-1H-benzo[de]isoquinolin-6-yl)piperazin-1-yl)propanoate) was carried out following the steps described by Espina et al. [16] (See supplementary Information), obtaining 0.03 g of a dark orange solid, **FP-CD- β CyD**.

2.3. *In vitro* studies

Three cell lines were used to analyze CDs-cell interaction: the murine melanoma cell line B16F10, the human osteosarcoma cell line U2OS and human mammary fibroblasts HMF. B16F10 and U2OS were kindly provided by Dr Carlos López-Otín (Universidad de Oviedo) and the HMF cell line was purchased from Innoprot. All cells were routinely cultured in an incubator at 37 °C, in a 5 % CO₂ atmosphere and in DMEM medium containing 10 % heat-inactivated fetal bovine serum, supplemented with 100 U/mL penicillin and 50 μ g/mL streptomycin from Gibco Life Technologies (Complete Cell Medium). These incubation conditions and cell medium composition are described as Standard Conditions throughout the text.

2.3.1. Cell culture and MTT assay

To quantify the effect of CDs on cell proliferation, B16F10 and U2OS were employed. To this end, the kit Cell Proliferation Assay (Promega) was used following manufacturer's instructions. In brief, 5000 cells/well were seeded in 96-well plates and exposed to free CDs and **CD- β CyD**s at

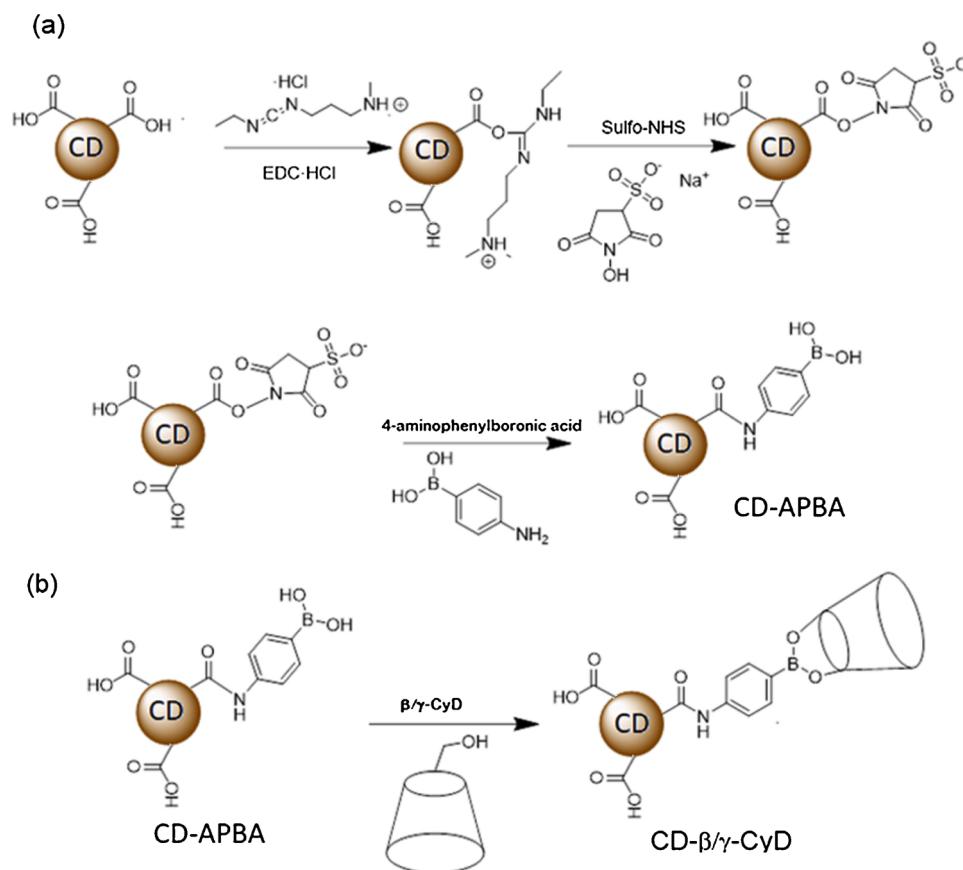


Fig. 1. Surface modification of CDs with β/γ -CyDs.

concentrations of 0.0, 0.01, 0.1, 0.5, 1.0 and 2.0. mg mL⁻¹ resuspended in Complete Cell Medium from a 10 mg mL⁻¹ stock solution in PBS during 5 days at 37 °C and 5 % CO₂. Cell proliferation was measured by the conversion of 3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulfophenyl)-2H-tetrazolium (MTT) into formazan water soluble compound. Absorbance can be then measured at 570 nm using a Synergy H4 Hybrid reader. Each condition was performed in sextuplets.

2.3.2. Fluorescence imaging

5×10^5 B16F10 and HMF cells were cultured on cover glasses within 24-well plates (Menzel-Gläser, Germany) and allowed to attach for 24 h (Standard Conditions). Cells were then incubated in the presence of 1 mg mL⁻¹ solution of free CDs or CD- β CyDs in Complete Cell Medium for 60 min under the same conditions. Then, media was removed, and cells washed 3 times with PBS pH 7.4. Finally, cells were fixed with 4% formaldehyde and mounted with Vectashield Antifade Mounting Medium.

2.3.3. Flow cytometry

B16F10, U2OS and HMF cells were cultured in 6-well plates in the presence of 1 mg mL⁻¹ of CDs for 1 h in Complete Cell Medium. After the incubation, cells were washed three times with PBS and detached from the well with a solution containing 0.25 % trypsin, 0.5 mM EDTA in DMEM. Following centrifugation (433 \times g), cells were re-suspended at a concentration of 1×10^6 cells/mL in 500 μ L of Sorting Buffer enriched with 2 % fetal bovine serum. Samples were filtered through a 35 μ m filter before each cytometer measurement to remove cell aggregates.

For cell cycle phase distribution studies, after 1 h incubation with CDs in the above conditions, trypsinised U2OS and HMF cells were washed three times with PBS, fixed with 80 % ethanol and kept at 4 °C until analysis was performed. For cytometric analysis, cells were centrifuged to remove ethanol and re-suspended in 250 μ L PBS. Finally,

2 mg of ribonuclease A and 0.2 mL of 1 mg mL⁻¹ propidium iodide were added. Cell cycle distribution was analysed in a Beckman Coulter Cytomics FC500 cytometer.

2.3.4. Intracellular pH by bioimaging

U2OS cells were cultured in the presence of 2×10^{-5} M FP-CD- β CyDs during 15 min in Standard Conditions. Then, cells were washed three times with PBS and fixed with 4 % v/v formaldehyde. Finally, images were taken in a Leica SP8 confocal microscope using a 405 nm laser and after 30 min incubation with high potassium cation regulators (30 mM NaCl, 120 mM KCl, 1 mM CaCl₂, 0.5 mM MgSO₄, 1 mM NaH₂PO₄, 5 mM glucose, 20 mM HEPES, 20 mM sodium acetate) at pH 6.0 and at pH 7.4.

2.4. In vivo studies

To perform *in vivo* studies, six 24-week-old C57BI/6J mice were used, housed under SPF (Specific Pathogen Free) conditions and according to the guidelines of the Animal Experimentation Committee of the University of Oviedo. They were distributed into two groups (n = 3), with one of them used as a control. All experiments and procedures with animals were carried out following the European Directive 2012/63/EU and following the indications of the Animal Experimentation facility of Universidad de Oviedo.

2.4.1. Fluorescence bioimaging

Mice anesthetized with ketamine (50 mg kg⁻¹ body weight) and xylazine (10 mg kg⁻¹ body weight) were used for *in vivo* fluorescence bioimaging. Once anesthetized, a dose of 2 mg kg⁻¹ animal of FP-CD- β CyD in PBS was injected through the lateral tail veins. Fluorescent distribution throughout the body was observed with the Xenogen IVIS Imaging System over a period of 1 h. To analyse specific organs, one hour after the intravenous injection of the nanomaterials, mice were

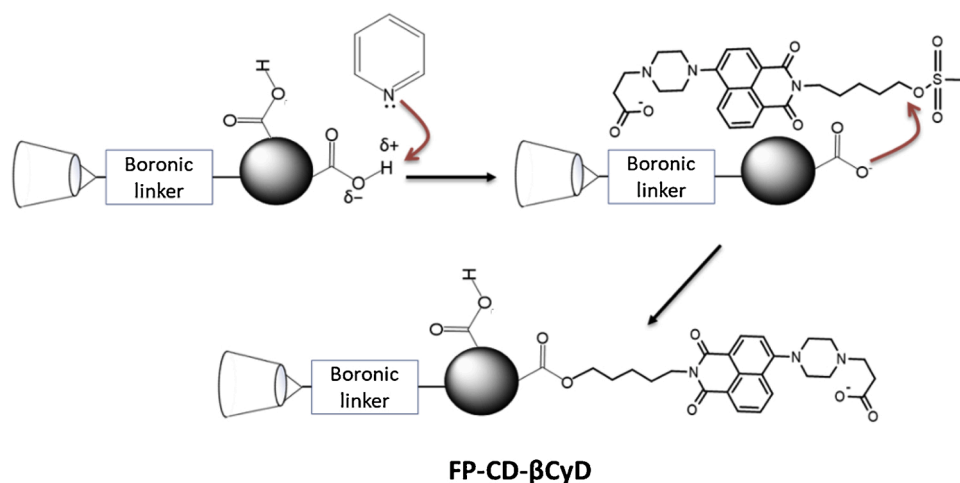


Fig. 2. Reaction pathway to obtain the ditopic assembly FP-CD-βCyD.

sacrificed in a CO₂ chamber. Liver, lung, kidney and heart were removed and immediately fixed in a 4 % formaldehyde saline solution for 24 h. Organs were then washed with 70 % ethanol and embedded in paraffin blocks. Before measuring fluorescence, samples were deparaffinized and treated with 0.1 % Triton X-100 to permeabilize and 0.1 M glycine to decrease autofluorescence.

3. Results and discussion

3.1. Synthesis and functionalization of carbon dots

Raw CDs obtained by a hydrothermal synthesis from glutathione and citric acid [17] have been selected in this work due to their interesting characteristics, mainly, a) high luminescence quantum yield with no photobleaching drawbacks [19], an interesting characteristics for bio-imaging applications; b) low cytotoxicity [20], c) antioxidant activity [21], and d) an easily modifiable surface due to the presence of the glutathione and citric acid functional groups (—COOH, NH₂ and S—H). Due to the presence of these surface groups, it was possible to sequentially modify glutathione-citric acid based CDs with different functions. So, in a first step, the covalent bond of βCyD was achieved by activation of surface C—OOH groups using a carbodiimide (EDC·HCl) reaction forming an amine-reactive O-acyl isourea intermediate which was then stabilized by the addition of sulfo-NHS, forming a sulfo-NHS ester more stable and efficient for conjugation to primary amines, APBA in the present case. As a result, the CDs were grafted with a thin organic layer of APBA (CD-APBA) (Fig. 1a), which further reacted with hydroxyl groups of the βCyD forming stable cyclic boronate esters [22], to render the CD-βCyD (Fig. 1b).

The ATR-FTIR spectrum of CD-βCyD (Fig. S1) shows the characteristic bands from their precursor compounds, which suggest an efficient functionalization. Thus, sharp band between 3300–3500 cm⁻¹, assignable to stretching vibrations of O—H bonds, can be clearly observed in CyD as well in the CD-βCyD spectra, while in raw CDs the band is unsharpened broad and weak. On the other hand, characteristic bands of βCyD can be observed in CD-βCyD such as the band at 2917 cm⁻¹, assigned to stretching vibrations of groups with C in sp³ hybridization (—CH and C—H₂) and the band at 1152 cm⁻¹ the vibration of the COC— (cyclic ether) group. The peak appearing at 1700 cm⁻¹ in CD and CD-βCyD spectra can be attributed to carboxylic groups on CD surface (from glutathione/citrate), although in the CD-βCyD the intensity of this band decreased significantly indicating a possible involvement of the APBA linker. The peak appearing at 1577 cm⁻¹ was assigned to the flexion of the NH group and stretching of the CN— group (amide, glutathione). Peaks at 1384 cm⁻¹ and 996 cm⁻¹ were assigned due to the asymmetric stretching of the B—O group and the

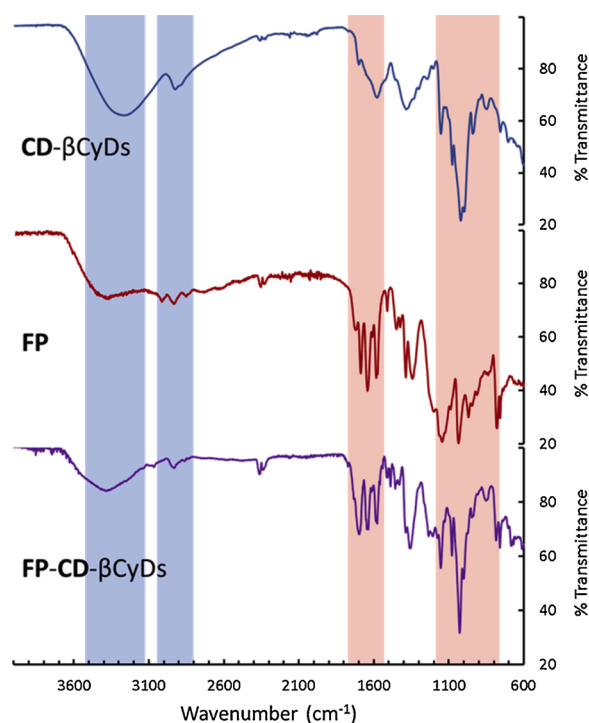


Fig. 3. ATR-FTIR spectra for FD-CD-βCyD and its main precursors.

deformation of the BOH group, respectively. These data, therefore, proved the participation of all precursors used in the synthesis of CD—βCyD. The comparative FTIR band assignment of precursors and final functionalized CDs are shown in Table S1. On the other side, HRTEM images demonstrated that the functionalization did not affect significantly the spherical morphology nor affected the original size of the raw CDs, being the nanoparticle diameter in the 3 ± 2 nm range both for CD and CD-βCyD (Fig. S2). In a second step, we performed a further functionalization of CD-βCyD with the fluorescent probe FP. This reaction was carried out in pyridine as solvent. Pyridine also acted as reaction activator, that is, the basic pyridine-like nitrogen ability to accept protons from the —COOH groups on the CD surface which, in turn favored the carboxylates nucleophilic attack of carboxylates to the carbon bound to the mesylate group of the dye. As a final result, the bi-functional ensemble FP-CD-βCyD was obtained (Fig. 2).

ATR-FTIR spectra of the bi-functional assembly exhibited characteristic peaks of the fluorescent probe FP (Fig. 3). The peaks at 1684 and

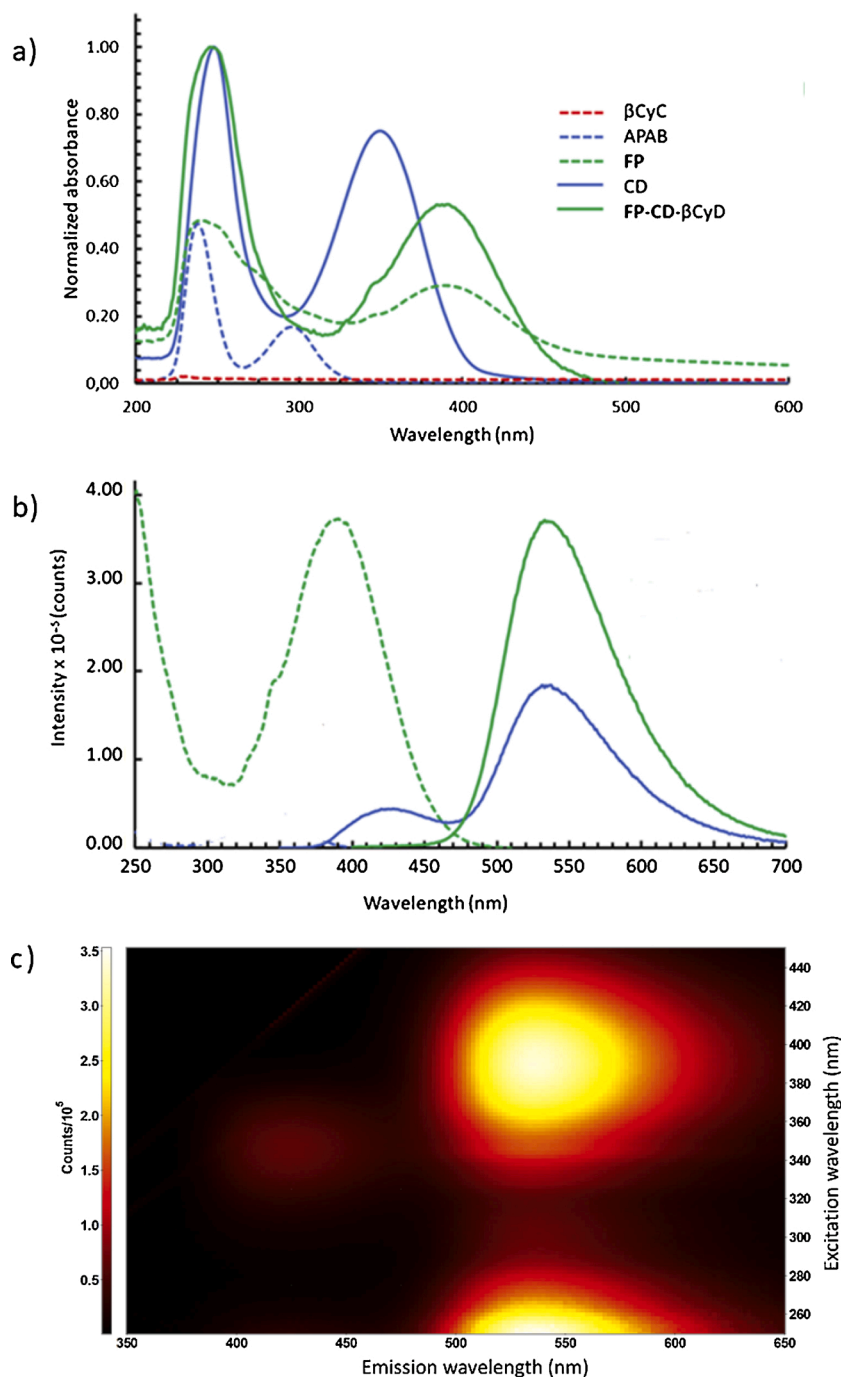


Fig. 4. a) Normalized absorption UV-Vis spectra of FP-CD-βCyD and their precursors: beta-cyclodextrin (βCyD), 4-aminophenylboronic acid (APAB), fluorescent probe (FP) and CDs. b) Excitation and photoemission spectra of FP-CD-βCyD: λ_{exc} 395 nm (green line) and λ_{exc} 345 nm (blue line). c) Contour plot of an excitation-emission matrix for the FP-CD-βCyD ensemble.

1645 cm^{-1} due to CO= stretching of amide groups, peaks at 1350 or 1151 cm^{-1} ascribed to ester acyl groups and peaks at 750–780 cm^{-1} assigned to aromatic rings. Besides, it is worth to note that in FP-CD-βCyD the peaks at 1199 cm^{-1} and 1145 cm^{-1} (S=O symmetric stretching) and that at 1089 cm^{-1} (from S–O flexion) characteristics of the methanesulfonate leaving group are absent, thus suggesting an efficient functionalization (Fig. 3, Table S1).

3.2. Spectroscopic characterization

Fig. 4a shows the absorption spectra of CDs and FP-CD-βCyD together with those of the chemicals used for functionalization. Raw CDs

spectrum exhibits a dominating absorption band at 248 nm assigned to π - π^* transitions in sp² dominions of CC and NC= bonds in CD [23] and a less intense band at 356 nm ascribed to π - π^* transitions of surface plasmon states corresponding to CO= bonds [24,25]. The FP-CD-βCyD shows an intense band centered at 248 nm somewhat broader than that of the raw CDs, due to the overlapping bands from APAB at 237 nm (π - π^* transitions of CC= phenyl bond) and at 294 nm from the CN n- σ^* transitions. This suggests that APAB was successfully attached to the CD surface and supports the FTIR data. As can be seen in the Fig. 4a, the absorption band of FP could also contribute to the broadening due to its band at 240 nm. FP-CD-βCyD also presented an absorption band at 393 nm, characteristic of the FP, which supported that the dye has been

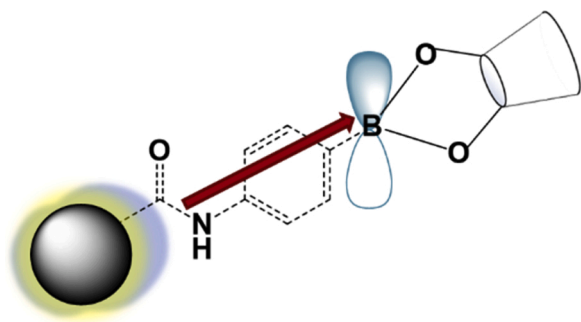


Fig. 5. Boron electron-withdrawing effects on the CDs surface states.

Table 1
Absolute quantum yields bands of the different synthesized nanomaterials.

Nanomaterial	λ_{exc} , nm	λ_{em} , nm	Absolute quantum yield, %
CDs	350	440	49
CD- β CyD	356	430	35
FP-CD	350	445	51
	395	535	16
FP-CD- β CyD	345–350	425	21
	395	533	60

successfully attached to the CDs.

The fluorescence excitation spectra of FP-CD- β CyD (Fig. 4b) showed an intense excitation band with a maximum intensity at 395 nm and a weak shoulder at 345 nm consistent with the relevant Vis-UV absorption bands. The fluorescent emission spectra of FP-CD- β CyD in Fig. 4b show that excitation at 395 nm (long wavelength absorption band of FP) shows a single emission band centred at 535 nm, whereas excitation at 345 nm leads to a very weak emission band centred at 446 nm, due to the excitation in-between the absorption bands of CD- β CyD (see Fig. S3) and a more intense band from FP centred at 536 nm. In accordance with this, the simultaneous presence of two luminescent centres in FP-CD- β CyD does not seem to influence the position of the main emission band. This can be also seen in the 2D excitation-emission contour map for the FP-CD- β CyD ensemble (Fig. 4c).

Upon excitation at 356 nm, the quantum yield of CD- β CyD resulted to be 35 %, lower than that from neat CDs, (49 % when excited at 350 nm) that could be ascribed to the presence of the boronic group close to the CD surface. As β CyDs were not directly bound to the CDs surface but to peripheral boronic groups, their influence on the CDs luminescent properties should be marginal. However, the proximity to the CDs surface of boron atoms, archetypal Lewis acids due to an empty *p* orbital, makes them act as electron-withdrawing inductive groups over neighbouring donating groups on the CDs surface, such -NH₂, -OH and -SH. The surface electronic states of the CDs may change as result of those inductive and resonance effects (Fig. 5) which, in turn, can affect the

electronic band structure of the CDs lowering significantly (approximately 30 %) the quantum yield of CD- β CyD in comparison to neat CDs.

On the other hand, the photoluminescence quantum yield of the bifunctional ensemble FP-CD- β CyD resulted to be 21 % when excited at 345–350 nm and increased to 60 % when excited at 395 nm. According to these data (Table 1), we conclude that when FP-CD- β CyD were excited in the range 345–350 nm (excitation maxima of neat CDs) the influence of the boron effect manifested itself as a low quantum yield. However, the high quantum yield upon excitation at 395 nm (excitation maximum of FP) could be ascribed to the presence of β CyD moieties that provided a protective restricted environment to the FP molecules from collisional quenching. In fact, in absence of β CyD, the quantum yield of the FP-CD system decreased to 16 % upon excitation at 395 nm (Table 1).

3.3. pH response and in vitro studies

In aqueous solution, the FP-CD- β CyD ensemble showed a similar response to pH to that of FP-CD, thus suggesting that the presence of the cyclodextrin moiety did not modify its ability to recognize H⁺ ions through a PET process [16]. The profile of the FP-CD- β CyD fluorescence intensity versus pH (excitation at 395 nm) showed a linear response in the interval pH 2–9 with a correlation coefficient 0.998 and a high sensitivity, around 100 fluorescence units/pH unit (Fig. S4). To determine the fitness of FP-CD- β CyD as fluorescent intracellular pH probe or as a bio-imaging dye, any potentially harmful cytotoxic effect must be evaluated. For that purpose, cytotoxicity was checked using B16F10 cells incubated in the presence of CDs, CD- β CyDs or FP-CD- β CyDs at 1 mg mL⁻¹ for 1 h. A cell culture without nanoparticles was employed as control. The optical microscope images (Fig. 6) did not show significant differences in cell morphology, which suggest that CDs do not alter cell integrity under the assayed conditions.

Cell proliferation studies [26], evaluated through the mitochondrial activity of cells using the MTT assay, were carried out by exposing B16F10 and U2OS cell lines to different concentrations of nanoparticles from 0.01 to 2.0 mg mL⁻¹ for 96 h. Presence of CD- β CyDs and FP-CD- β CyDs induced low toxicity at lower concentrations (0.01–0.1 mg mL⁻¹) towards B16F10 cells which was comparable to that of free CDs. At higher concentrations and larger exposure times, CD- β CyDs and FP-CD- β CyDs induced higher toxicity than that of free CDs (Fig. 7). Similar response profiles were observed for U2OS cells (Fig. S5), even though U2OS were more sensitive to both, CD- β CyDs and FP-CD- β CyDs, than B16F10 cells. In fact, B16F10 showed a cell viability of 90 % with concentrations as high as 0.1 mg mL⁻¹, whereas U2OS show this cell viability till the range of 0.01 mg mL⁻¹. On the other hand, while free CDs caused no apparent damage on any of the two cell lines within the concentration range studied, no differences in cell-induced toxicity between CD- β CyDs and FP-CD- β CyDs were observed. These results seem to suggest that the presence of the β CyD moiety on the FP-CD- β CyD surface is responsible for the observed toxic effect, rather than the fluorescent

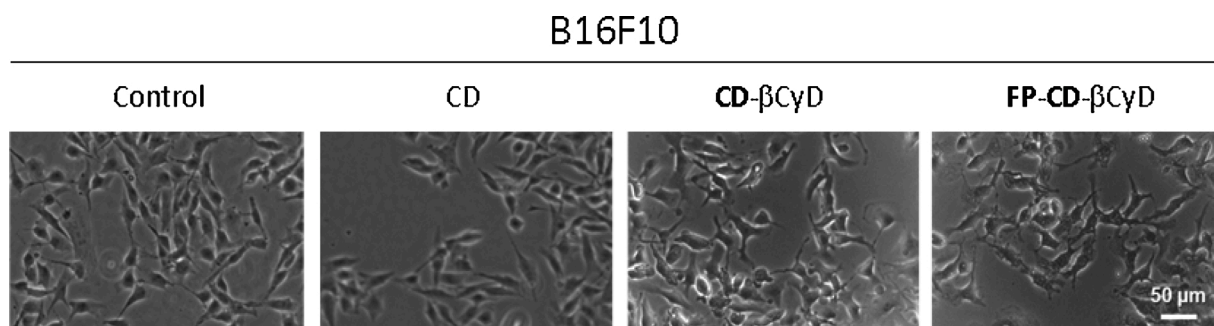


Fig. 6. Phase contrast microscopy images of control B16F10 cells in the absence (control) and in the presence of 1 mg/mL of CDs, CD- β CyD and FP-CD- β CyD for 1 h at 37 °C. Images were taken with a 20 \times objective.

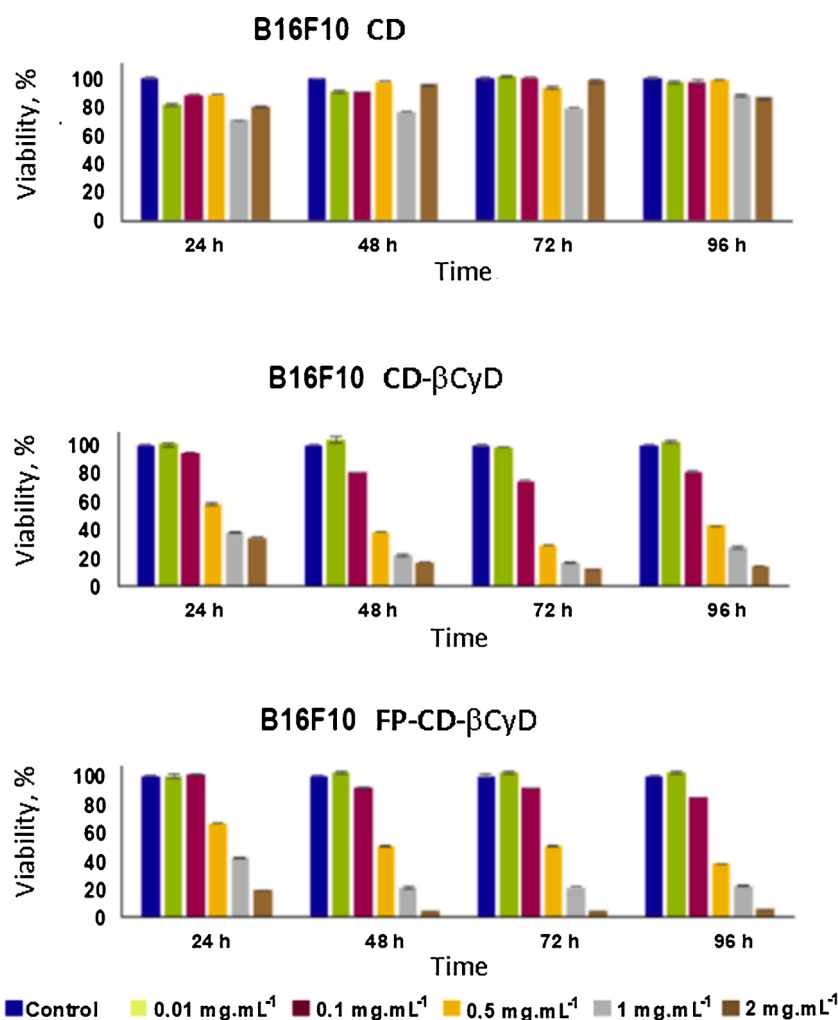


Fig. 7. Cytotoxicity evaluation of B16F10 cells after incubation with different concentrations of CDs, CD-βCyDs and FP-CD-βCyDs and during different times.

probe itself. In this regard, several reports describe the toxic effects of βCyD to living organisms due to its affinity to cholesterol and to its capacity to solubilize cell membrane lipids [27–29].

Fluorescence microscope images, obtained at 365 nm, 436 nm and 546 nm, showed that FP-CD-βCyDs were located at the cytoplasm of treated cells, likely internalized by an endocytosis mechanism [30] (Fig. 8). Furthermore, after cell internalization, a clear contrasted fluorescence signal was observed within each cell with discrete spots indicating specific compartmentalization of the nanoparticles. This images clearly suggest the potentiality of these nanoparticles to be used in cellular bioimaging. These results were confirmed by cell sorting in a cytometer (Fig. S6) in which different cell populations can be distinguished. Thus, the signal of the intensity in the cell population containing FP-CD-βCyDs clearly differs from that of the control population and is characterized by higher fluorescence signal within the cells. However, cells containing CD-βCyDs cannot be as clearly differentiated from control cells. The fluorescence observed upon excitation at 546 nm may be due not only to the presence of fluorescent proteins in the cells but also to the CDs. In fact, it is well known that the CDs photoluminescence spectrum is excitation-wavelength dependent, undergoing bathochromic shift as the excitation red shifts. Besides, the presence of surface functional groups in CDs such as C=O and CN= groups can tune the fluorescence of CDs. So, O-related groups (C=O) and N-related groups (C=N) may introduce two new energy levels (HOMO-n_O and HOMO-n_N), thus producing new electron transitions and the red-shift emission occurs when the electrons in C=O or CN= groups related defect states return to the HOMO [31–34].

Effects of CD-βCyDs and FP-CD-βCyDs in cell cycle phase distribution of the tumor cell line U2OS in comparison with HMF cells was studied after incubation with 1 mg.mL⁻¹ of the corresponding nanoparticles. Fig. 9 shows that HMF cells slightly accumulate in S (DNA synthesis) and G2 (mitosis preparation) phases after incubation with both modified CDs and when compared with control cells. Thus, presence of the fluorescence probes in the CDs may compromise among other things cellular division. On the other hand, cell cycle distribution of U2OS cells was not affected after incubation with CD-βCyD or FP-CD-βCyD, suggesting the biocompatibility of these nanoparticles with this cell line.

3.3.1. Intracellular pH

In cancer cells, cytoplasmic pH may oscillate between 7.2 and 7.5 during cell growth as consequence of cell metabolism, while intracellular pH can oscillate between 6.2 and 6.9 (Warburg effect) [35]. We recently demonstrated [16] that the FP-CD fluorescent system was sensitive to pH in aqueous media through a PET (photoinduced electron transfer) process.

In Fig. 10, confocal microscopy images for U2OS cells shows the inactivation of the PET process in the probe at pH 7.4 in comparison with pH 6.0 after incubation with the FP molecule for 30 min. In order to evaluate the potential of FP-CD-βCyD as a fluorescent probe for the recognition of cancer cells based on the Warburg effect, we carried out a comparative study with healthy HMF cells and cancerous B16F10 cells treated with 1 mg mL⁻¹ CDs, CD-βCyD and FP-CD-βCyD. The relative intensity data showed that fluorescence after incubation with CD and CD-βCyD did not seem to be pH sensitive as marginal differences were

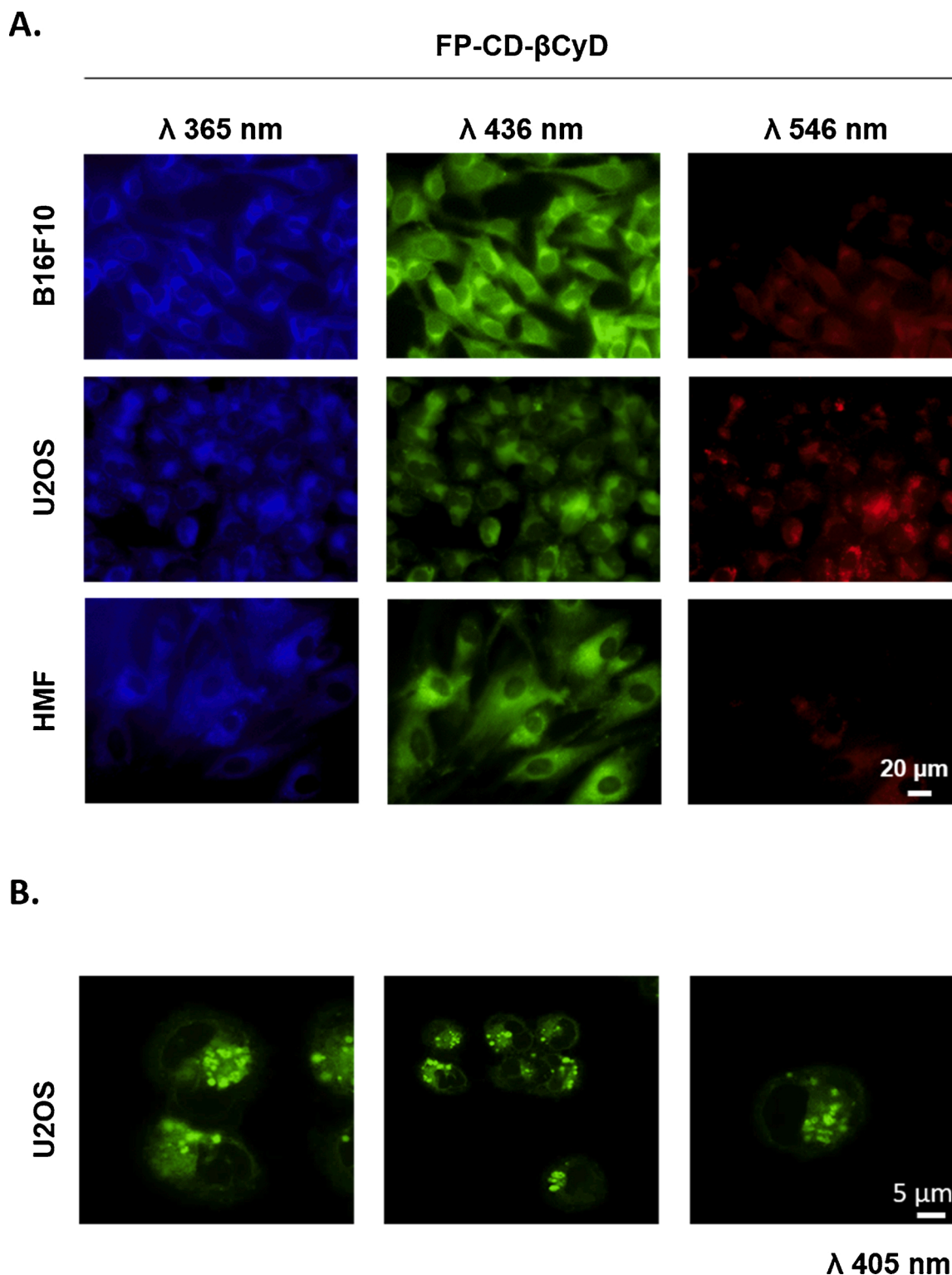


Fig. 8. A. Cell localization of FP-CD- β CyDs after incubation with the indicated cell line. Images were taken at 63x from fixed cells after 1 h incubation at 37 °C with 1 mg mL⁻¹ of FP-CD- β CyDs. B. Detailed images of U2OS cells were taken by confocal fluorescence at 63x augmentation showing discrete distribution of FP-CD- β CyDs. Scale is indicated in white.

observed between HMF and B16F10 cells (Fig. 11a). However, there was a significant difference in the fluorescence emission in HMF culture compared to that of B16F10 cells when FP-CD- β CyDs were used (2-fold). The mechanism of FP-CD- β CyD interaction with cell surface is suggested in Fig. 11b in which we take into consideration the fact that cyclodextrins are known to interact with components of cell membranes [36]. In

this scenario, a possible mechanism that could explain the difference between both cell lines may be based on the interaction of FP-CD- β CyDs with the cell membrane. While the β CyD unit of FP-CD- β CyDs may penetrate the cell membrane interface forming hydrogen bonds with phosphate and glycerol ester-groups of lipids [37], the FP unit can remain exposed towards the external membrane surroundings

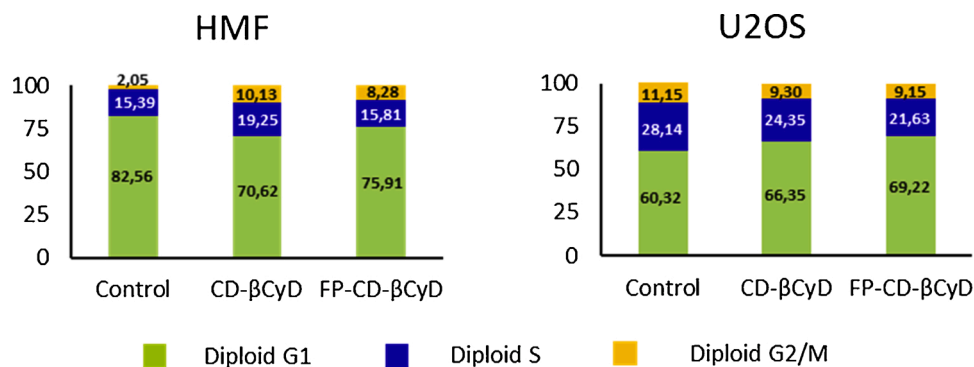


Fig. 9. Cell cycle phase distribution after incubation with 1 mg mL⁻¹ of CD-βCyD or FP-CD-βCyD and in the absence of any CD. Numbers indicate cell percentages at the corresponding cell cycle phase.

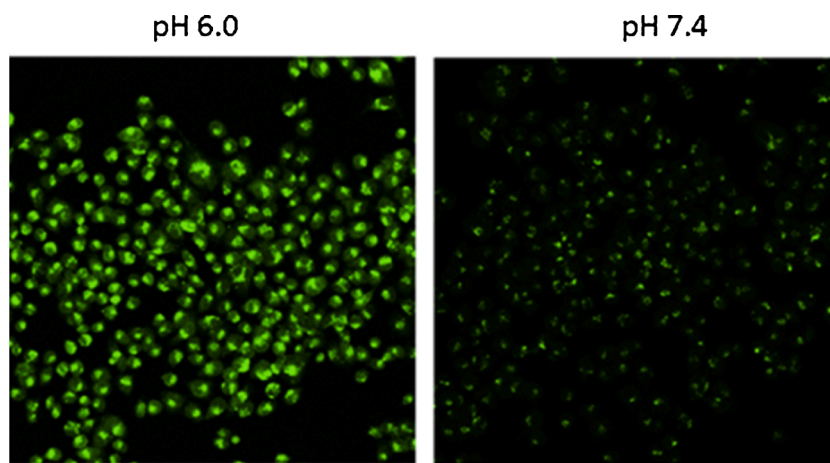


Fig. 10. Fluorescence confocal microscopy images for U2OS cells (63×, laser 405 nm), at pH 6.0 and pH 7.4 in the presence of the FP molecule.

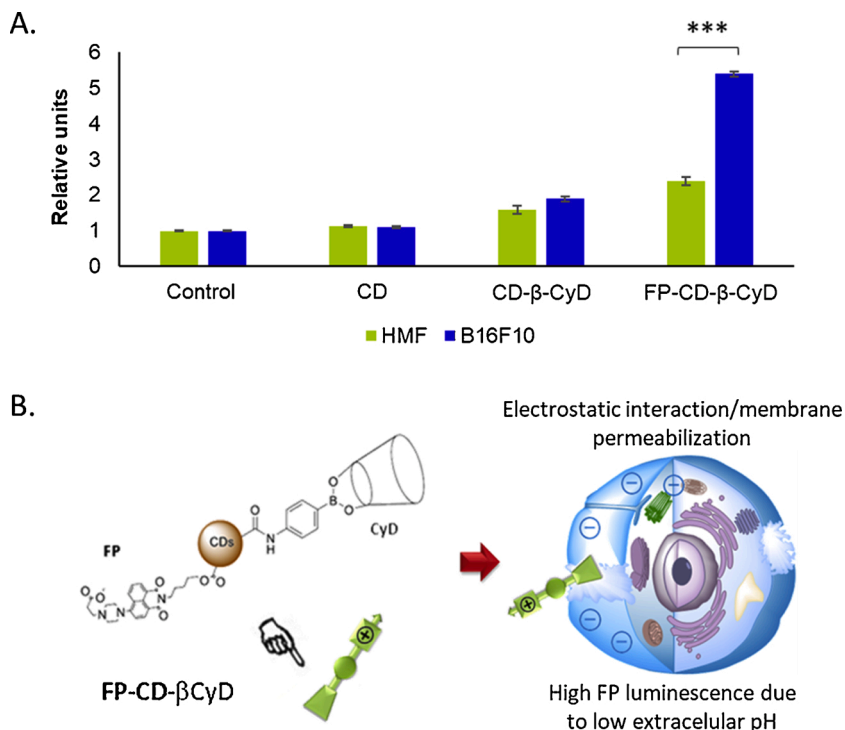


Fig. 11. A. Fluorescence relative intensities of HMF and B16F10 cells after 1 h exposition to 1 mg mL⁻¹ of nanomaterials (control: untreated cells). B. Suggested mechanism of FP moiety exposure to the cell surface. (***) $p < 0.005$, t Student).

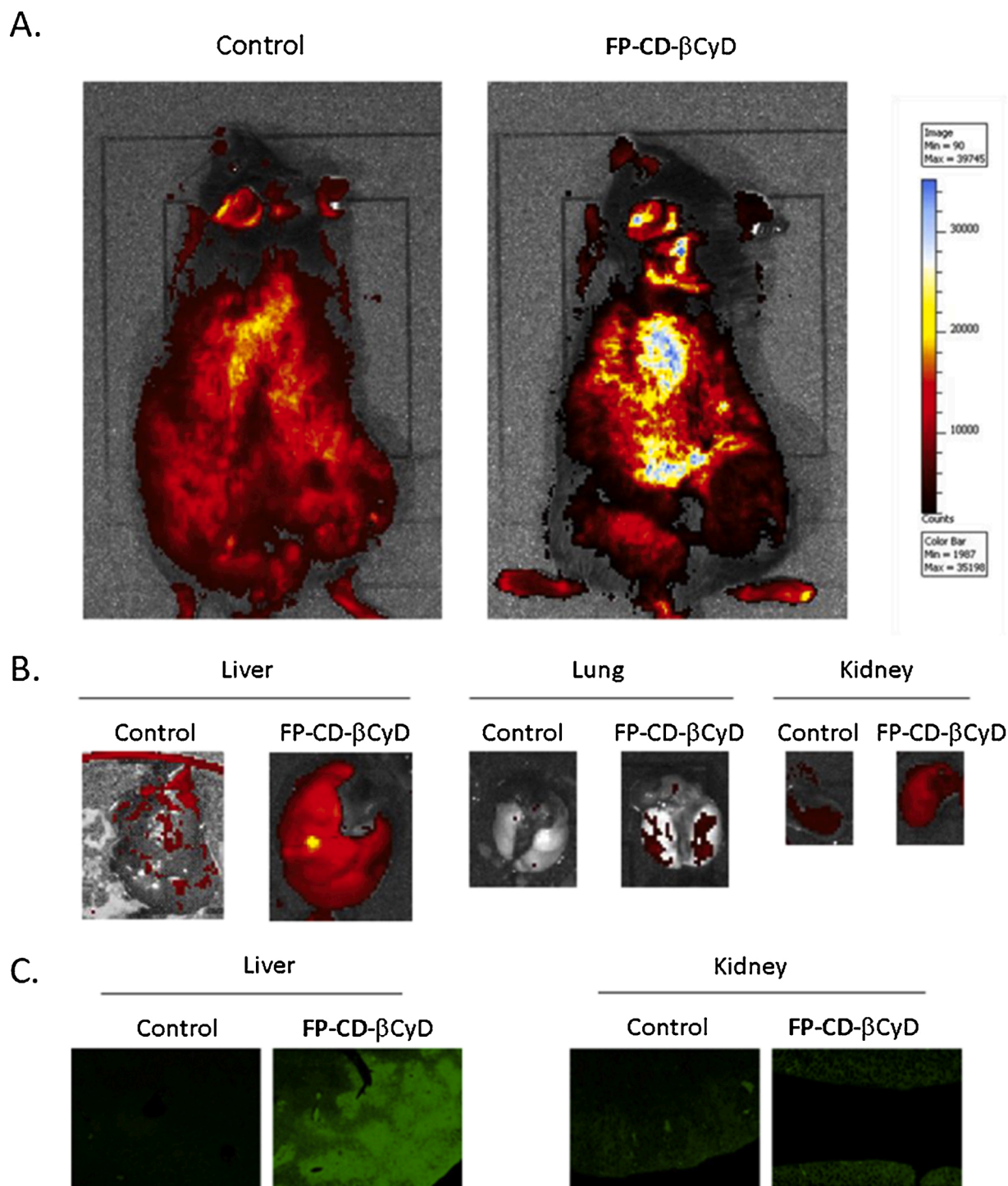


Fig. 12. A. *In vivo* bioimages of mice 5 min after intravenous injection of FP-CD-βCyDs. Left: control mouse with no injection showing autofluorescence. Right: FP-CD-βCyD injected mouse. B. *Ex vivo* images of organs removed 1 h after FP-CD-βCyD injection. C. Confocal fluorescence images of liver and kidney tissue slices.

(Fig. 11b). In the case of HMF cells, the FP unit was in contact with an intracellular pH 7.2–7.5 and its fluorescence was quenched due to the electron transfer from the nitrogen atoms of the H⁺ recognition unit (the piperazine group, pKa = 7.45) to the excited 1,8-naphthylamide signaling subunit (PET process) [16]. On the contrary, when the FP unit was in contact with the low pH (6.2–6.4) of B16F10 cells the PET process was blocked and the fluorescence revived when the lone nitrogen atoms of the recognition units were protonated.

3.4. *In vivo* studies

In vivo studies were carried out in mice in order to study the distribution of FP-CD-βCyDs in whole organisms after blood injection. Results showed that nanoparticles tend to accumulate in several organs in which higher fluorescence was detected (Fig. 12a). To characterize these organs, mice were sacrificed one hour after FP-CD-βCyD injection and then, liver, kidneys and lungs were analyzed. Fluorescence of these organs are shown in Fig. 12b. Results indicated that liver and kidney accumulate more FP-CD-βCyDs than lung attending to the level of the

fluorescence intensity. In addition, tissue samples were microtomed and analyzed using confocal fluorescence microscopy, confirming that liver was the organ with the highest amount of FP-CD- β CyD (Fig. 12c), probably due to its unique vasculature among the other organs and its role in the transformation of xenobiotics to compounds that the body can eliminate.

4. Conclusions

In this article, glutathione-based carbon nanoparticles have been successfully bi-functionalized. In the first place, with the incorporation of a fluorescent FP label which can be used as a tool to differentiate healthy cells from cells with an acid external pH (typically cancer cells due to Warburg effect). On the other hand, cyclodextrins have also been incorporated as a glucose source, which in turn, eases glycolysis in cancer cells and favors its use as a controlled transport and delivery system for different drugs.

The cytotoxicity studies of these materials in three different cell lines, revealed that, although there is not any visual substantial modification in cell morphology, the β CyD functionalized CDs diminished cell viability and induced alterations in cell cycle phase distribution. In fact, HMF cells seem to be arrested around G2/M phase, something that did not happen with the unfunctionalized CDs.

Likewise, CDs are not only able to get inside the cells and accumulate within discrete points of the cytoplasm, but also to emit fluorescence allowing *in vitro* images with high contrast, especially when using FP-CD- β CyDs, whose fluorescence in cancer cell lines is significantly higher than in the noncancerous cell line. In addition, these materials have been used for obtaining *in vivo* fluorescence images after their injection in mice, with an important incidence in those organs with the highest vasculature (mainly liver). These results suggest that these materials are a promising alternative for bioimaging applications.

Finally, from our results we can stand out that the bi-functionalized nanoparticles show a high potential for their use in theranostics as sensor and actuator. Thus, they can be used either as cancer cell marker, because of their ability to identify cells with a slightly acidic environment, as well as to facilitate drug administration during therapy.

Author contribution

Rosana Badía-Lafño and Marta Elena Díaz García conceived and supervised the research; Jorge Espina-Casado and Tania Fontanil developed the experimental work with the same degree of responsibility; Alfonso Fernández-González performed FTIR characterization and data interpretation; Alfonso Fernández-González, R. Badía and Marta Elena Díaz-García wrote the original draft preparation, the revision and edition; S. Cal and A.J. Obaya supervised and interpreted the *in vitro* and *in vivo* toxicity experiments; Rosana Badía-Lafño was responsible for funding acquisition and project administration. All authors gave final approval for publication.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary material related to this article can be found, in the online version, at doi:<https://doi.org/10.1016/j.snb.2021.130555>.

References

- [1] D. Qu, X. Wang, Y. Bao, Z. Sun, Recent advance of carbon dots in bio-related applications, *J. Phys. Mater.* 3 (2020), 022003, <https://doi.org/10.1088/2515-7639/ab7cb9>.
- [2] G. Romero, S.E. Moya, Chapter 4 - synthesis of organic nanoparticles, in: J.M. de la Fuente, V. Grazu (Eds.), *Frontiers of Nanoscience*, Elsevier, 2012, pp. 115–141.
- [3] C. Xia, S. Zhu, T. Feng, M. Yang, B. Yang, Evolution and synthesis of carbon dots: from carbon dots to carbonized polymer dots, *Adv. Sci.* 6 (23) (2019) 4564–4565, <https://doi.org/10.1002/advs.201901316>.
- [4] K. Dimos, Carbon quantum dots: surface passivation and functionalization, *Curr. Org. Chem.* 20 (2016) 682–695, <https://doi.org/10.2174/1385272819666150730220948>.
- [5] B. Yao, H. Huang, Y. Liu, Z. Kang, Carbon dots: a small conundrum, *Trends Chem.* 1 (2019) 235–246, <https://doi.org/10.1016/j.trechm.2019.02.003>.
- [6] S. Das, L. Ngashangva, P. Goswami, Carbon dots: an emerging smart material for analytical applications, *Micromachines* 12 (2021) 84, <https://doi.org/10.3390/mi12010084>.
- [7] L. Liu, L. Chen, J. Liang, L. Liu, H. Han, A novel ratiometric probe based on nitrogen-doped carbon dots and rhodamine isothiocyanate for detection of Fe³⁺ in aqueous solution, *J. Anal. Meth. Chem.* (2016), 4939582, <https://doi.org/10.1155/2016/4939582>.
- [8] H. Sing, J.S. Sidhu, D.K. Mahajan, N. Singh, A carbon quantum dot and rhodamine-based ratiometric fluorescent complex for the recognition of histidine in aqueous systems, *Chem. Front.* 3 (2019) 476–483, <https://doi.org/10.1039/c8qm00554k>.
- [9] L.L. Guo, Y.H. Song, K.Y. Cai, L. Wang, 'On-off' ratiometric fluorescent detection of Hg²⁺ based on N-doped carbon dots rhodamine B@TAPT-DHTA-COF, *Spectrochim. Acta Mol. Biomol. Spectrosc.* 227 (2020) 1–8, <https://doi.org/10.1016/j.saa.2019.117703>.
- [10] Z. Yu, W. Ma, T. Wu, J. Wen, Y. Zhang, L. Wang, Y. He, H. Chu, M. Hu, Coumarin-modified graphene quantum dots as a sensing platform for multicomponent detection and its applications in fruits and living cells, *ACS Omega* 5 (2020), <https://doi.org/10.1021/acsomega.9b04387>, 7369–7368.
- [11] Q. Xiao, S. Lu, C. Huang, W. Su, S. Huang, Novel N-doped carbon dots/ β -cyclodextrin nanocomposites for enantioselective recognition of tryptophan enantiomers, *Sensors* 16 (2016) 1874, <https://doi.org/10.3390/s16111874>.
- [12] Z.Y. Lin, Y.C. Kuo, C.J. Chang, Y.S. Lin, T.C. Chiu, C.C. Hu, Highly sensitive sensing of hydroquinone and catechol based on β -cyclodextrin-modified carbon dots, *RSC Adv.* 8 (2018) 19381–19388, <https://doi.org/10.1039/c8ra02813c>.
- [13] E. de los Reyes-Berbel, I. Ortíz-Gómez, M. Ortega-Muñoz, A. Salinas-Castillo, L. F. Capitan-Vallvey, F. Hernández-Mateo, F.J. López-Jaramillo, F. Santoyo-González, Carbon dots-inspired fluorescent cyclodextrins: competitive supramolecular "off-on" (bio)sensors, *Nanoscale* 12 (2020) 9178–9185, <https://doi.org/10.1039/d0nr01004a>.
- [14] A.O. Alqarni, S.A. Alkahtani, A.M. Mahmoud, M.M. E-Wekil, Design of "Turn On" fluorimetric nanoprobe based on nitrogen doped graphene quantum dots modified with β -cyclodextrin and vitamin B6 cofactor for selective sensing of dopamine in human serum, *Spectrochim. Acta Part A: Mol. Biomol. Spectrosc.* 248 (2021), 119180, <https://doi.org/10.1016/j.saa.2020.119180>.
- [15] R.M. Pascale, D.F. Clavisi, M.M. Simile, C.F. Feo, F. Feo, The Warburg effect 97 years after its discovery, *Cancers* 12 (2020) 2819, <https://doi.org/10.3390/cancers12102819>.
- [16] J. Espina Casado, A. Fernández González, M.E. Díaz García, R. Badía Lafño, Smart carbon dots as chemosensor for control of water contamination in organic media, *Sens. Actuators B Chem.* 329 (2021), 129262, <https://doi.org/10.1016/j.snb.2020.129262>.
- [17] T. Díaz-Faes López, A. Fernández González, M.E. Díaz-García, R. Badía-Lafño, Highly efficient Förster resonance energy transfer between carbon nanoparticles and europium-tetracycline complex, *Carbon* 94 (2015) 142–151, <https://doi.org/10.1016/j.carbon.2015.06.066>.
- [18] S. Mondal, P. Purkayastha, A-cyclodextrin functionalized carbon dots: pronounced photoinduced electron transfer by aggregated nanostructures, *J. Phys. Chem. C* 120 (26) (2016) 14365–14371, <https://doi.org/10.1021/acs.jpcc.6b03145>.
- [19] H. Li, S. Ye, J. Guo, H. Wang, W. Yan, J. Song, J. Qu, Biocompatible carbon dots with low-saturation-intensity and high-photobleaching-resistance for STED nanoscopy imaging of the nucleolus and tunneling nanotubes in living cells, *Nano Res.* 12 (2019) 3075–3084, <https://doi.org/10.1007/s12274-019-2554-x>.
- [20] N. Liu, M. Tang, Toxicity of different types of quantum dots to mammalian cells in vitro: an update review, *J. Hazard. Mater.* 399 (2020), 122606, <https://doi.org/10.1016/j.jhazmat.2020.122606>.
- [21] C. Murru, R. Badía Lafño, M.E. Díaz García, Synthesis and characterization of green carbon dots for scavenging radical oxygen species in aqueous and oil samples, *Antioxidants* 9 (11) (2020) 1147, <https://doi.org/10.3390/antiox9111147>.
- [22] C. Shimpuku, R. Ozawa, A. Sasaki, F. Sato, T. Hashimoto, A. Yamauchi, I. Suzuki, T. Hayashita, Selective glucose recognition by boronic acid azoprobe/ γ -cyclodextrin complexes in water, *Chem. Commun.* (13) (2009) 1709–1711, <https://doi.org/10.1039/B819938H>.

- [23] H. Wang, C. Sun, X. Chen, Y. Zhang, V.L. Colvin, Q. Rice, J. Seo, S. Feng, S. Wang, W.W. Yu, Excitation wavelength independent visible color emission of carbon dots, *Nanoscale* 9 (5) (2017) 1909–1915, <https://doi.org/10.1039/C6NR09200D>.
- [24] S. Fatimah, S.M. Bilqis, I. Tahir, D. Tahir, Luminescence properties of carbon dots synthesis from sugar for enhancing glows in paints, *Mater. Res. Express* 6 (9) (2019), 095006, <https://doi.org/10.1088/2053-1591/ab2bb7>.
- [25] D.Y. Guo, C.X. Shan, K.K. Liu, Q. Lou, D.Z. Shen, Surface plasmon effect of carbon nanodots, *Nanoscale* 7 (2015) 18908, <https://doi.org/10.1039/c5nr05918f>.
- [26] T. Mosmann, Rapid colorimetric assay for cellular growth and survival: application to proliferation and cytotoxicity assays, *J. Immunol. Methods* 65 (1) (1983) 55–63, [https://doi.org/10.1016/0022-1759\(83\)90303-4](https://doi.org/10.1016/0022-1759(83)90303-4).
- [27] T. Kiss, F. Fenyvesi, I. Bácskay, J. Váradi, E. Fenyvesi, R. Iványi, L. Szente, A. Tószaki, M. Vecsernyés, Evaluation of the cytotoxicity of β -cyclodextrin derivatives: evidence for the role of cholesterol extraction, *Eur. J. Pharm. Sci.* 40 (2010) 376–380, <https://doi.org/10.1016/j.ejps.2010.04.014>.
- [28] V.J. Stella, Q. He, Cyclodextrins, *Toxicol. Pathol.* 36 (2008) 30–42, <https://doi.org/10.1177/0192623307310945>.
- [29] S. Shityakov, R.E. Salmas, E. Salvador, N. Roewer, J. Broscheit, C. Förster, Evaluation of the potential toxicity of unmodified and modified cyclodextrins on murine blood-brain barrier endothelial cells, *J. Toxicol. Sci.* 41 (2) (2016) 175–184, <https://doi.org/10.2131/jts.41.175>.
- [30] B. Unnikrishnan, R.S. Wu, S.C. Wei, C.C. Huang, H.T. Chang, Fluorescent carbon dots for selective labeling of subcellular organelles, *ACS Omega* 5 (20) (2020) 11248–11261, <https://doi.org/10.1021/acsomega.9b04301>.
- [31] M. Liu, Optical properties of carbon dots: a review, *Nanoarchitectonic* 1 (1) (2020) 1–12, <https://doi.org/10.37256/nat.112020124.1-12>.
- [32] K. Jiang, S. Sun, L. Zhang, Y. Lu, A. Wu, C. Cai, H. Lin, Red, green, and blue luminescence by carbon dots: full-color emission tuning and multicolor cellular imaging, *Angew. Chem. Int. Ed.* 54 (2015) 5360–5363, <https://doi.org/10.1002/anie.201501193>.
- [33] S. Sarkar, M. Sudolská, M. Dubecký, C.J. Reckmeier, A.L. Rogach, R. Zboril, M. Otyepka, Graphitic nitrogen doping in carbon dots causes red-shifted absorption, *J. Phys. Chem. C* 120 (2) (2016) 1303–1308, <https://doi.org/10.1021/acs.jpcc.5b10186>.
- [34] K. Holá, M. Sudolská, S. Kalytchuk, D. Nachtigallová, A.L. Rogach, M. Otyepka, R. Zboril, Graphitic nitrogen triggers red fluorescence in carbon dots, *ACS Nano* 11 (12) (2017) 12402–12410, <https://doi.org/10.1021/acsnano.7b06399>.
- [35] L. Schwartz, T. Seyfried, K.O. Alfarouk, J.V. Moreira, S. Fais, Out of Warburg effect: an effective cancer treatment targeting the tumor specific metabolism and dysregulated pH, *Semin. Cancer Biol.* 43 (2017) 134–138, <https://doi.org/10.1016/j.semcancer.2017.01.005>.
- [36] Z. Hammoud, N. Khreich, L. Auezova, S. Fourmentin, A. Elaissari, H. Greige-Gerges, Cyclodextrin-membrane interaction in drug delivery and membrane structure maintenance, *Int. J. Pharm.* 564 (2019) 59, <https://doi.org/10.1016/j.ijpharm.2019.03.063>.
- [37] W. Khuntawee, P. Wolschann, T. Rungrotmongkol, J. Wong-Ekkabut, S. Hannongbua, Molecular dynamics simulations of the interaction of beta cyclodextrin with a lipid bilayer, *J. Chem. Inf. Model.* 55 (9) (2015) 1894–1902, <https://doi.org/10.1021/acs.jcim.5b00152>.

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