

A multi-faceted, environmental forensic characterization of a paradigmatic brownfield polluted by hazardous waste containing Hg, As, PAHs and dioxins

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Abstract

Hg and As mining-metallurgy plants have severely impacted environmental compartments. In this context, La Soterraña site (northern Spain) is a paradigmatic site that has been studied in terms of soil, water and vegetation pollution. However, here we used a novel multi-purpose forensic approach to determine the characteristics of the main source of pollution that is still active at the site, namely the waste deposits that built up over decades. This approach included advanced analytics focused not only on the main mining-metallurgical waste but also construction and demolition (C&D) waste as a potential repository of hazardous pollutants. In addition, we used GC-MS techniques to perform a detailed assessment of organic pollution. Our results indicate that most of the C&D waste should be considered hazardous, although, as expected, waste from the metallurgy industry raises maximum concern. The presence of extremely high concentrations of Hg and As in very fine grain-size fractions of the waste, the extent of As mobility found, and the co-occurrence of organic contaminants (PAHs, dioxins, etc.) and unexpected toxic organometallics confirm that La Soterraña is one of the most polluted sites in Europe. This scenario therefore calls for an urgent clean-up campaign.

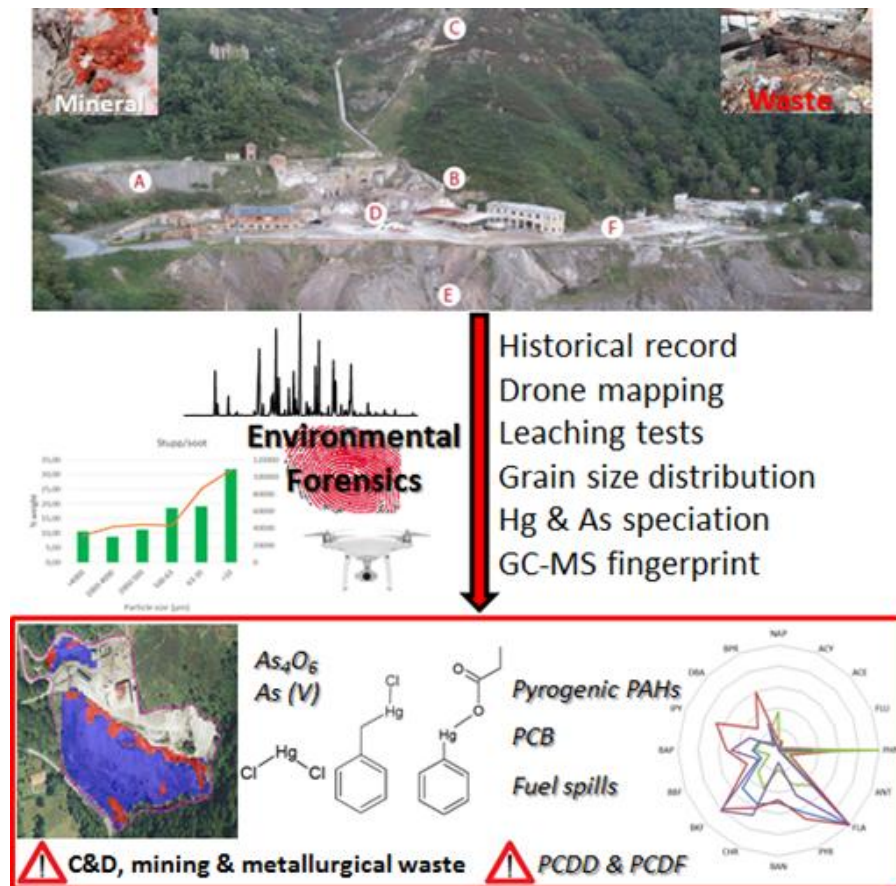
Keywords

Hazardous waste; Arsenic; Mercury; PAHs; Dioxins.

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(Fernández et al., 2020)

Graphical abstract



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Highlights

A multifaceted forensic approach revealed severe pollution in a brownfield.

C&D waste contains As and Hg.

Complex speciation, including unforeseen Hg-aromatic compounds, examined.

PAH ratios and distribution revealed pyrogenic sources.

Hg-pyrometallurgy processes produced dioxins and furans.

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2 **brownfield polluted by hazardous waste containing Hg, As, PAHs and dioxins**

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15 used a novel multi-purpose forensic approach to determine the characteristics of the
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21 that most of the C&D waste should be considered hazardous, although, as expected,
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24 extent of As mobility found, and the co-occurrence of organic contaminants (PAHs,
25 dioxins, etc.) and unexpected toxic organometallics confirm that La Soterraña is one of
26 the most polluted sites in Europe. This scenario therefore calls for an urgent clean-up
27 campaign.

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30

31

32 1. Introduction

33 Former industrial and mining sites (brownfields) are usually affected by mixtures of
34 pollutants that have been released into the environment or have been disposed of
35 inappropriately. The co-occurrence of such complex mixtures poses a challenge for site
36 remediation. In this context, a forensic approach (Alker et al., 2000; Gallego et al.,
37 2016; Haggmann et al., 2019), including advanced analytical methods, is often required
38 to identify the distinctive features of source-specific contamination (Morrison and
39 Murphy, 2006; Uhler et al., 2010). Balancing the attributes of different target and non-
40 target analytical methods is therefore advantageous for the comprehensive chemical
41 fingerprinting of these sites (Lara-Gonzalo et al., 2015).

42 With respect to the environmental risks associated with brownfields, the mobilization of
43 potentially toxic elements (PTE) in abandoned mining areas has been widely studied
44 (Ahn et al., 2005; Álvarez-Ayuso et al., 2012; Bindler et al., 2012; Chopin and Alloway,
45 2007; Gray et al., 2002; Moore and Luoma, 1990; Riewerts et al., 2014). Hg, for
46 instance, is the focus of the Minamata convention and has received considerable
47 attention (Evers et al., 2016; Liu et al., 2012; UNEP, 2013). This metal has a complex
48 biogeochemical cycle, and ionic Hg can be converted into alkyl-Hg compounds, which
49 are highly toxic (He et al., 2019). In turn, it is widely recognized that As poses a threat
50 to public health (Haffert and Craw, 2008; Tarvainen et al., 2013). Arsenic toxicity
51 causes environmental problems worldwide, in particular in relation to natural
52 groundwater (Wang et al., 2018; Wilkin et al., 2018). Indeed, mine tailings and their
53 effluents usually contain high concentrations of As and are thus one of the main
54 anthropogenic sources of this contaminant (Wang and Mulligan, 2006).

55 Until the 1970s, the Asturian mining region in northern Spain was one of the largest
56 Hg-producing regions in the world. Most of the ores exploited contained both As and
57 Hg (Loredo et al., 1988). The environmental effects of such mining activity have been
58 extensively addressed in recent decades (Higueras et al., 2015; Loredo et al., 1999),
59 together with approaches for the treatment of polluted soils and waters (Boente et al.,
60 2018; Fernández et al., 2017; Gil-Díaz et al., 2019; Sierra et al., 2013). Of the former
61 Hg-sites in the region, two processing plants were particularly relevant because of the
62 concurrence of mining activity and metallurgical processes. The former metallurgical
63 activity of the first plant, known as El Terronal, and its dramatic effects on the
64 environmental compartments has been widely studied (Gallego et al., 2015; González-

65 [Fernández et al., 2018](#)). The second plant, known as La Soterraña, presents similar
66 problems ([Loredo et al., 2006](#)) and has been evaluated from various perspectives with
67 the aim to propose potential remediation measures ([Ayala and Fernández, 2019](#);
68 [Matanzas et al. 2017](#); [Sierra et al., 2011](#)). However, a detailed forensic study of the
69 main sources of pollution (different types of waste) and their leachability has not been
70 done to date. Furthermore, the abandoned buildings and installations have not been fully
71 described, nor have combined studies taken into consideration the potential presence of
72 organic contaminants.

73 To properly manage environmental hazards in brownfields, analyses should address not
74 only stock-piles of distinct industrial or mining waste but also construction &
75 demolition (C&D) waste ([Somasundaram et al., 2015](#); [Staunton et al., 2014](#)). Therefore,
76 in sites with a severe metal and metalloid pollutant load, C&D waste should be taken
77 into account ([Saca et al., 2017](#)). In turn, Hg-sites are commonly characterized by the co-
78 occurrence of organic contaminants ([Gallego et al., 2015](#)). This feature deserves further
79 attention given that brownfields commonly house installations related to organic
80 pollution, such as underground storage tanks, transformers, and high-temperature ovens
81 ([Lara-Gonzalo et al., 2015](#)). C&D waste can also include organic pollutants ([Butera et
82 al., 2014](#); [Jang and Townsend, 2001](#)). In this regard, it should be noted that the
83 methodical evaluation of the co-occurrence of organic and heavy metal(loid)s in
84 brownfield sites is uncommon ([Gallego et al., 2016](#); [Thavamani et al., 2011](#)). Notably,
85 gas chromatography-mass spectrometry (GC-MS) techniques have proved powerful
86 tools for environmental studies and for the chemical fingerprinting of organic
87 compounds, especially in the case of complex mixtures and processes such as
88 weathering ([Douglas et al., 2007](#); [Kruge et al., 2018](#); [Uhler et al., 2010](#)).

89 In the context of organic and PTE pollution in brownfield sites, here we conducted a
90 multifaceted, environmental, forensic evaluation of La Soterraña, including the
91 consideration of C&D waste and a detailed analysis of potential organic contaminants
92 (PAHs, PCBs, dioxins and others). This integral and exemplifying study: i) provides a
93 comprehensive classification of the waste found, including volumes and surfaces
94 affected; ii) addresses the mechanisms of pollutant dispersal in the surroundings of the
95 brownfield; iii) identifies sources of organic pollutants on the site; and iv) sets out a
96 comprehensive approach to evaluate the study site, including toxicological features and
97 potential remediation measures.

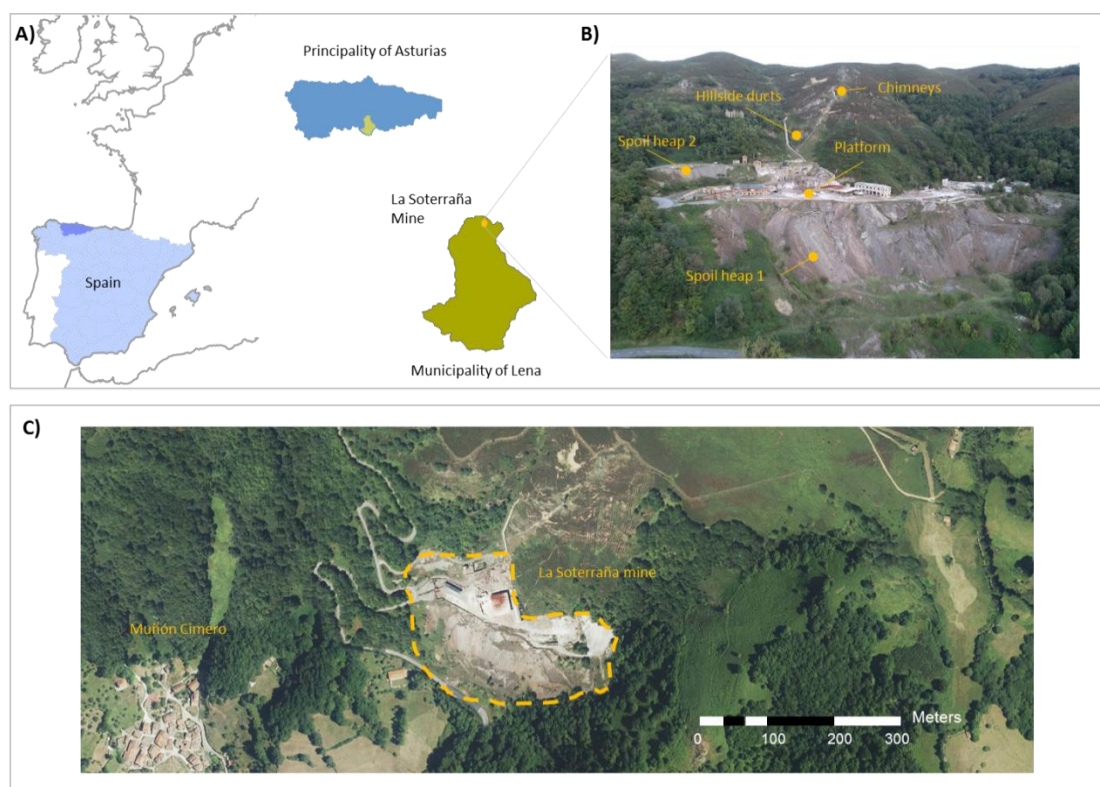
98 **2. Materials and methods**

99

100 *2.1 Historical data, site description and samples*

101

102 The site of La Soterraña (Fig. 1A) is located in Asturias (NW Spain). It was used for
103 decades—until its closure in 1974—for the exploitation of an underground Hg mine and
104 for the metallurgical processing of Hg ores from that mine and others in the vicinity.
105 The main characteristics of the site (ore geology, geomorphology, vegetation, weather,
106 etc.) have been described previously (see Matanzas et al., 2017 and references therein).
107 As a general overview (see results for details), the site currently consists of a wide
108 platform where the main installations and ruins of buildings are located. The first soil-
109 heap and the old chimneys are located uphill of this area, while a larger spoil heap is
110 located downhill (Fig. 1B).



111

112

113 Figure 1: A) Location of the study site in NW Spain (43°11'34" N, 5°50'36" W). B)
114 Current front-view of La Soterraña site and the main installations. C) Aerial view of La
115 Soterraña and distance (below 500 m) to the closest populated areas.

116

117 We studied the historical information available, including aerial views from different
118 dates, an initial detailed in situ reconnaissance, and a detailed cartography of the site
119 (drone flight carried out with a Phantom IV Pro at 70 m altitude and the subsequent GIS
120 processing), thereby identifying the buildings on the main platform of the
121 pyrometallurgy plant, the surrounding spoil heaps, and the duct/chimney area. The
122 volumes and surfaces of the aforementioned elements within the site were estimated by
123 the “volume calculations tools” of Arcmap 10.3 on the digital surface model previously
124 obtained from high-resolution drone imagery (processed with Pix4Dmapper 4.3.31
125 photogrammetry software).

126 Subsequently, we designed and performed a scrupulous sampling of 27 distinct waste
127 materials (see Fig 2. in Results for specific location of the samples). Composite samples
128 (ranging from 200 g to several kg depending on their typology) were collected and
129 stored in plastic bags. The sampled materials were brushed lightly in situ to remove
130 recent dust. When needed, a clean stainless steel tool was used to scrape fragments and
131 fine material off the first centimetres of the C&D waste found (wall coverings, brick
132 surfaces, etc.).

133

134

135 2.2 *Elemental analysis*

136 Composite samples were initially screened and sorted (different types of waste) by
137 means of semi quantitative x-ray fluorescence using dispersed energy (FRX-ED) in a
138 Shimadzu spectrometer (EDX-720HS) equipped with a Si(Li) detector.

139 After the initial classification, representative waste subsamples were dried at room
140 temperature in order to minimize the loss of volatile contaminants. They were then
141 disaggregated by a roller in order to facilitate sieving through a 4-mm screen. Materials
142 with a grain-size greater than 4 mm were vigorously washed and rubbed to recover fine
143 particles adhered to the gravels and pebbles, which, once cleaned, were excluded from
144 the study. The final material was mechanically split (riffle) to obtain 50-g representative
145 subsamples and then pulverized to a particle size of less than 100 μm . All utensils were
146 thoroughly cleaned with ethanol between samples. For multi-element analysis, 0.250-g
147 representative subsamples were leached by means of an ‘Aqua regia’ digestion
148 ($\text{HCl}+\text{HNO}_3$) in an Anton Paar 3000 microwave. The samples were diluted and filtered.

149 As and Hg were quantified by an Inductively Coupled Plasma Mass Spectrometer (ICP-
150 MS 7700, Agilent Technologies) using IDA (Isotope Dilution Analysis).

151

152 2.3 *Leaching tests*

153 Compliance testing was conducted to check whether the samples satisfied widely
154 accepted waste regulations. To this end, the waste materials were classified using the
155 UNE-EN 12457-3:2004 test and USEPA Method 1311 (TCLP, Toxicity Characteristic
156 Leaching Procedure).

157 The UNE-EN 12457-3:2004 test was applied with a liquid/solid (L/S) ratio of 10 l/kg,
158 and deionized water as a leaching liquid. This method involved stirring the solution for
159 24 h and then leaving the samples to decant. The solution was then passed through a
160 membrane filter (0.45 μm), and a subsample of the leachate was taken for each material
161 for further analysis by ICP-MS.

162 The TCLP test was performed to evaluate the potential mobility and availability of As
163 and Hg in the waste samples. An acetic acid-based TCLP extraction fluid was prepared
164 with 2.572 g of NaOH and 5.72 mL of glacial acetic acid ($\text{pH } 4.93 \pm 0.05$). This test
165 involved stirring 1 g of sample with 20 ml of leaching liquid. After 18 h of mixing, the
166 leachate was passed through a Millipore 0.4- μm filter. The pH of the filtrate was then
167 measured, and the leachate was acidified with a small amount of nitric acid to a pH less
168 than 2, followed by ICP-MS analysis.

169

170 2.4 *Chemical speciation*

171 As and Hg species were measured in order to obtain more information about the
172 potential toxicity of the wastes. In brief, to measure the former, 0.1 g of waste and an
173 extracting agent (1 M H_3PO_4 + 0.1 M ascorbic acid) were placed in a microwave vessel.
174 The extracts were diluted and filtered. The As species were split with a mobile phase of
175 2 M PBS (Phosphate Buffered Saline)/0.2 M EDTA ($\text{pH} = 6.0$) in a separation column
176 with a 1260 Infinity HPLC coupled to the ICP-MS described above. In addition, Hg
177 species were extracted using a solution of 7.6% HCl and 10% 2-mercaptoethanol in an
178 ultrasonic bath. The extract was then centrifuged and diluted. The determination was
179 carried out in a HPLC apparatus coupled to ICP-MS, using a ZORBAX Eclipse XDB
180 C18 (2.1 mm i.d. \times 50 mm, 5 μm) column and 0.06 M ammonium acetate, 5% methanol
181 and 0.1% 2-mercaptoethanol ($\text{pH} = 6.8$) as mobile phase.

182

183 2.5 *Grain-size distribution*

184

185 Particle size distribution tests were performed by wet sieving 1 kg of another set of
186 representative subsamples (only for mining and metallurgical waste) to obtain particle-
187 size fractions of <63, 63-500, 500-2000, 2000-4000 and >4000 μm . The grain-size
188 distribution of the <63 μm fraction was obtained in a Beckman Coulter particle sizing
189 analyzer, model LS 13 320, using the Aqueous Liquid Module (ALM) with an auto-
190 sampler. In addition, to obtain sub-samples in the size intervals of 10-63 μm and <10
191 μm , classification tests were performed in a hydrocyclone Mozley C124 rig (currently
192 provided by Salter Cyclones). The separation conditions used were established after
193 some preliminary tests and consisted of a cyclone diameter of 50.8 mm, underflow
194 diameter of 9.5 mm, overflow diameter of 14.6 mm, 0.2 bar of pressure and a solids
195 feed of 15% wt.

196

197 2.6 *Organic analysis*

198 Aliquots of the same 27 samples used for inorganic analysis were considered for
199 organic pollutant assessment. Also, two additional samples (see section 3.1) were taken,
200 both showing evident organoleptic properties, thereby suggesting hydrocarbon and PCB
201 content, respectively.

202 *GC-MS screening*

203 1-g representative subsamples were extracted with hexane:dichloromethane (1:1, v/v) in
204 a Soxtherm system (Gerhardt). The extract was concentrated by rotary evaporation.
205 Aliquots of the Soxtherm extract were fractionated and gravimetrically quantified by
206 LC (Liquid Chromatography) steps in the case of pure products (the two additional
207 samples described above), whereas the extracts of the other 27 samples were directly
208 injected into the chromatograph after rotary evaporation. For LC, maltenes and
209 asphaltenes were separated by passing the extract through 0.45- μm filters using hexane
210 and dichloromethane, respectively. Maltenes were then split into saturated, aromatic,
211 and polar fractions by LC in columns filled with silica gel and alumina (saturates were
212 eluted with hexane, aromatics with a mix of dichloromethane:hexane (4:1, v/v) and the
213 polar fraction with methanol).

214 Direct extracts and LC fractions were analyzed by GC/MS (Gas Chromatography –
215 Mass Spectrometry). The extracts were injected into a 7890A GC System coupled to a

216 5975C Inert XL MSD with Triple-Axis Detector (Agilent Technologies). A capillary
217 column DB-5ms (5% phenyl 95% dimethylpolysiloxane) 30 m × 0.25 mm i.d. × 0.25
218 μm film (Agilent Technologies) was used, with helium as carrier gas at 1 mL/min. The
219 initial oven temperature was 40°C (held for 5 min) and was ramped up at 5°C/min to
220 300°C (held for 20 min). The mass spectrometer was operated in electron ionization
221 mode (EI) at 70 eV and was calibrated daily by auto-tuning with perfluorotributylamine
222 (PFTBA). The chromatograms were acquired in full-scan mode (mass range acquisition
223 from 45 to 500 m/z).

224

225 *PAH and PCB quantification*

226 After Soxhterm extraction (see above) with Dichloromethane:Acetone (1:1), the 16
227 priority PAHs were quantified in the same GC/MS described above. In this case, the
228 initial oven temperature was 70°C (held for 2 min) and was ramped up at 20°C/min to
229 220°C, raised to 270°C at 10°C/min (held for 1 min), then raised to 290°C at 10°C/min
230 (held for 1 min), and finally ramped up at 10°C/min to 300°C (held for 7 min). The GC
231 injector was operated in splitless mode for 2 min, and its temperature was maintained at
232 260°C. The mass spectrometer was operated in selected ion monitoring mode (SIM),
233 and the quantification of m/z relations was as follows: 128 (Naphthalene); 152, 153 and
234 154 (Acenaphthylene and Acenaphthene); 165 and 166 (Fluorene); 178 (Phenanthrene
235 and Anthracene); 202 (Fluoranthene and Pyrene); 228 (Benz[a]anthracene and
236 Chrysene); 252 (Benzo[b]fluoranthene, Benzo[k]fluoranthene and Benzo[a]pyrene); and
237 276 and 278 (Dibenz[a,h]anthracene, Benzo[ghi]perylene and Indene[1,2,3-cd]pyrene).
238 We also measured the PCB Congener Content Evaluation Revised Mix 1 in the same
239 GC/MS device. To do so, the initial oven temperature was 50°C (held for 1 min) and
240 was ramped up at 30°C/min to 200°C (held for 2 min), and finally ramped up to
241 10°C/min to 300°C (held for 15 min). The GC injector was operated in splitless mode,
242 and its temperature was maintained at 280°C. The mass spectrometer was operated in
243 SIM, and the quantification of m/z relations was as follows: 186, 256, 258 (2,4,4'-
244 Trichlorobiphenyl); 220, 290, 292 (2,2',5,5'- Tetrachlorobiphenyl); 256, 326, 328
245 (2,2',4,5,5'- Pentachlorobiphenyl); 256, 326, 328 (2,3',4,4',5- Pentachlorobiphenyl);
246 290,360,362 (2,2',3,4,4',5'- Hexachlorobiphenyl); 290, 360, 362 (2,2',4,4',5,5'-
247 Hexachlorobiphenyl); and 324, 394, 396 (2,2',3,4,4',5,5'- Heptachlorobiphenyl).

248

249 *Dioxins and Furans*

250 To quantify tetra- through to octa-chlorinated dioxins and furans (PCDD and PCDF),
251 samples were extracted, cleaned up, fractionated and then analyzed by isotope dilution
252 and HRGC-HRMS following EPA Method 1613 requirements. PCDDs and PCDFs
253 were identified with a Trace GC Ultra coupled to a DFS high resolution magnetic sector
254 mass spectrometer (GC-HRMS) from Thermo Scientific. The capillary column was a
255 TR-DIOXIN-5 ms 60 m × 0.25 mm i.d. × 0.25 μm film (Thermo Scientific). The initial
256 oven temperature was 140°C (held for 1 min) and was ramped to 200°C at 20 °C/min
257 and held for 3 min, then raised to 310°C at 3 °C/min and held for 8 min, and finally
258 raised to 325°C and held for 5 min. The carrier gas was helium at a column flow of 1
259 mL/min. The HRMS detector was operated in MID (Multiple Ion Detection) mode. The
260 identification and quantification of congeners and the calculation of the toxicity
261 equivalents were done by isotope dilution analysis following EPA Method 1613 and
262 using the software TargetQuan by Thermo Scientific.

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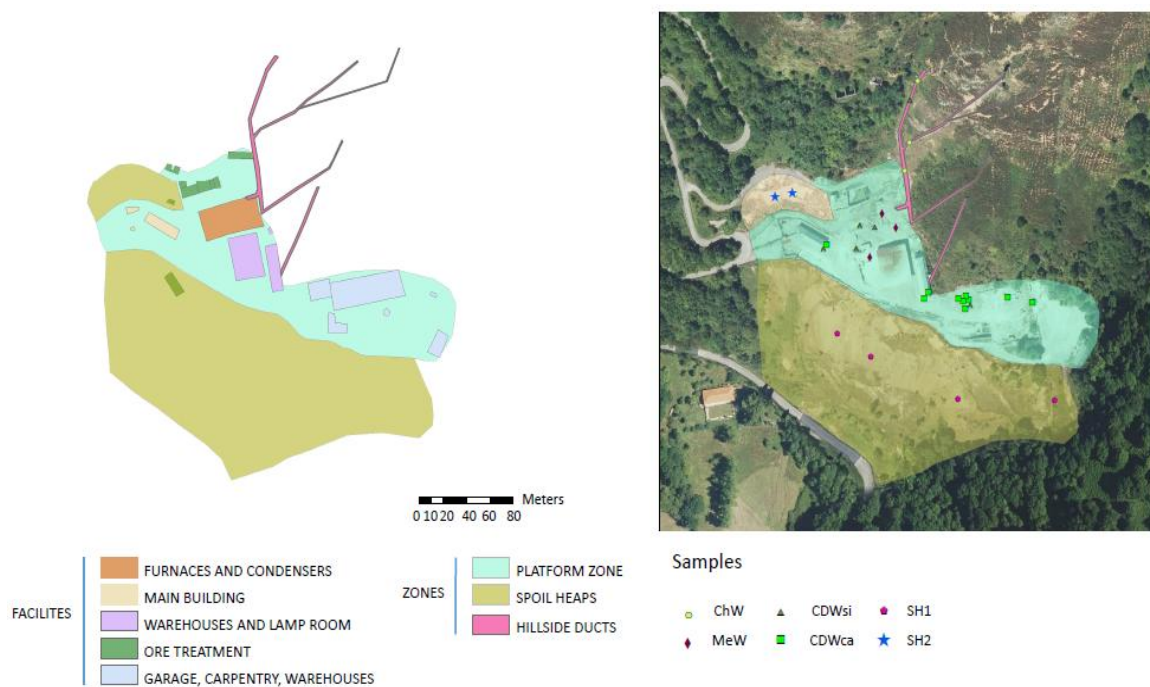
277 **3. Results**

278

279 *3.1 Site reconnaissance and selection of samples*

280 This section provides a summary of the background information about the activities
281 undertaken in the plant, and the reconnaissance and cartography data obtained (Fig.
282 2A). The site has different sub-areas according to the activities that were developed; i.e.
283 mining activity, ore processing and metallurgy, auxiliary processes, etc. After several
284 decades of abandonment, the site is fully derelict and only some of the structures
285 (partial ruins) are still standing. In this context, the main elements of the site can be
286 classified into three main groups (see detailed photographs in Fig. S1, supplementary
287 material): (i) platform zone where the mine and metallurgical activities were carried out,
288 including 20 derelict buildings (only 5 still standing although damaged). C&D and
289 metallurgical waste are distributed unevenly and the remains of the old ovens and
290 condensation systems can be observed; (ii) mine waste spoil heaps covering a total area
291 of 2.9 ha divided in two separate dumps, one in the upper part (northern spoil heap,
292 smaller size, labelled SH2 in Fig. 2B) of the platform and the other in the lower part
293 (southern spoil hear, larger size, labelled SH1 in Fig. 2B); and (iii) the ruins of five
294 hillside ducts (between 150 and 350 m long) running along the southern slope of the hill
295 up to vertical chimneys (up to 13 m high). Note that gases from the furnace were
296 released into the atmosphere through these hillside chimneys of distinct diameters.

297



298

299 Figure 2: A) Main facilities and subdivision of the site into zones. B) Sample locations
 300 (SH: Spoil heaps; MeW: Metallurgy waste; CDW: Construction & Demolition Waste;
 301 ChW: Chimney Waste –Flue Dust-).

302

303 Samples were collected within the areas described above with the following distribution
 304 and sorting (Fig. 2B):

- 305 ✓ Mining waste: Four representative samples were collected in the larger spoil heap
 306 (southern, SH1) and two in the smaller one (northern, SH2).
- 307 ✓ Metallurgical waste: Following the previous description abovementioned (Gallego
 308 et al., 2015), 3 representative samples of stupp/soot waste (MeW in Fig. 2B) and 3
 309 samples of flue dust (ChW in Fig. 2B) were taken. In brief, stupp/soot is a mixture
 310 of As-enriched dust recovered in the first stage of the Hg purification/condensation
 311 system, and Hg-rich soot is a dark residue consisting of Hg-soot and Hg-ore dust.
 312 On the other hand, flue dust is a mixture of the slurry formed during the cleaning of
 313 the chimneys and some As- and Hg-rich soot and dusts that were not retained in the
 314 purification/condensation systems.
- 315 ✓ Construction & demolition waste: Up to 15 different types of debris and wall
 316 material were sampled in the remaining buildings and rubble. An initial XRF
 317 screening of these samples suggested a division between limestone-gypsum type (9

318 samples, labelled CDW_{Ca} in Fig. 2B) and ceramic-metallic type (6 samples,
319 labelled CDW_{Si} in Fig. 2B).

320 ✓ Organic residues: Spilled fuel oil from the underground storage tanks used in the
321 ovens and a blackish residue found beneath the zone housing the former
322 transformers were also sampled.

323

324 3.2 Waste total concentration, leachability, speciation and volumes

325 Samples classified as mining, metallurgy or C&D waste were studied by ICP-MS
326 measurements of total concentrations and the two types of leachates described above.

327 The main results are summarized in [Table 1](#).

328 Table 1: Total concentration of As and Hg, leachates and chemical speciation.

Waste type (sample label, see Fig 2B), number of samples		Average total ± dev. (g/t)		Leachates (%)				HPLC-MS speciation (%)	
				UNE-EN 12457 (H ₂ O)		US-EPA 1311 (TCLP)			
		As	Hg	As	Hg	As	Hg	As (III)	Methyl-Hg
Mining waste	Spoil heap 1 (SH1), n=4	15,000±2,210	890±170	0.2	<0.1	3.2	<0.1	2.1	0
	Spoil heap 2 (SH2), n=2	23,000±250	6,000±1,610	0.2	<0.1	0.4	<0.1	0.3	0
C&D Waste	Limestone-gypsum (CDW _{Ca}), n=9	2,120±980	30±20	3.0	0.1	12.0	<0.1	0	0
	Ceramic-metallic (CDW _{Si}), n=6	64,830±11,600	14,680±10,320	1.9	0.2	5.4	<0.1	0	0
Metallurgical waste	Stupp/soot (MeW), n=3	95,900±51,620	107,100±24,640	2.2	<0.1	12.0	<0.1	15.0	0
	Flue dust (ChW), n=3	138,200±25,800	33,000±6,900	14.2	<0.1	14.8	<0.1	64.6	0

329

330 In general terms, Hg leachability was very low with both methods used. In contrast, As
331 leachability was significant, not correlated with the total concentrations, and higher
332 using the US-EPA 1311 (TCLP, slightly acidic) than the UNE-EN 12457 method (H₂O,
333 neutral). Metallurgical wastes presented the highest percentages of leachability,

334 although C&D waste limestone-gypsum type also showed high values for As. Finally,
335 mining waste presented lower values in the leachate measurements. On the other hand,
336 analysis of As and Hg speciation (also shown in Table 1) revealed the absence of
337 methyl-Hg (more toxic than inorganic Hg), and a heterogenous abundance of As (III)
338 (more toxic than As (V)), which was particularly abundant in the metallurgical waste.

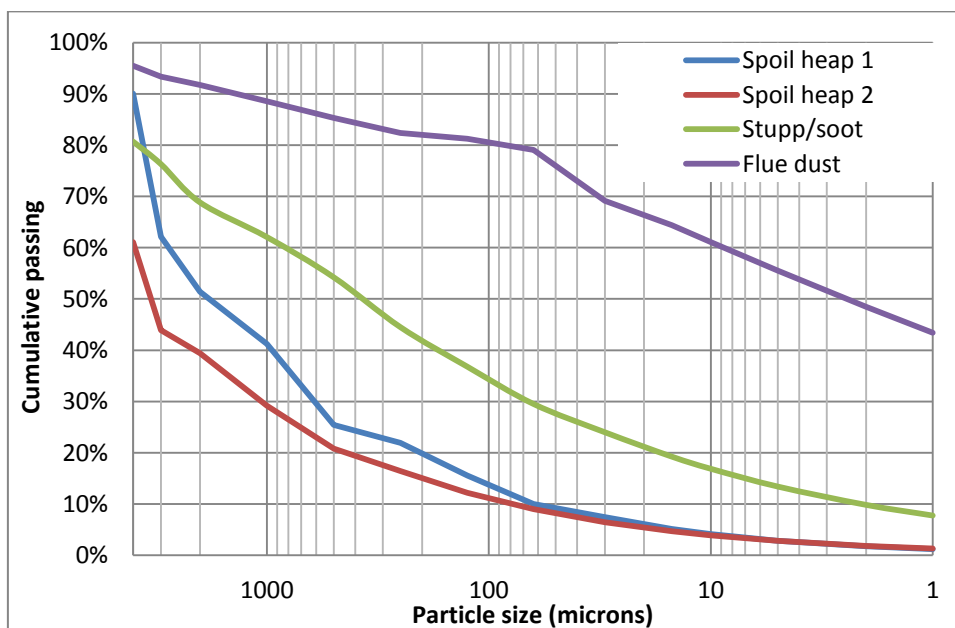
339 Finally, and in addition to the chemical characterization, we calculated the surficial area
340 and volumes of the different waste using a compilation of the cartographic data reported
341 by the drone flight, other previous aerial views, former studies (Loredo et al., 2006;
342 Luque, 1985, and references therein), and the *in situ* reconnaissance carried out in this
343 work. Our results (Table S1) highlighted the vast amount of waste accumulated in the
344 main spoil heap (around 200,000 t), and a remarkable estimated total mass of Hg
345 (hundreds of tons, particularly in the metallurgical waste) and particularly of As
346 (thousands of tons, particularly in the mining waste) still present in the study site.

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349 3.3 Grain-size study of mining and metallurgical waste

350 Only mining and metallurgical wastes (because of their volume and very high
351 contaminant content) were subjected to grain-size studies. Fig. 3 shows the merged
352 results of the wet sieving and laser diffraction grain-size analyses for the representative
353 subsamples examined.



354

355 Figure 3: Grain-size analyses of mining and metallurgical waste (wet sieving + laser
356 diffraction).

357 The grain-size curves indicate a distinct distribution of fractions between mining waste
358 in the spoil heaps (thicker, with d_{50} well-above 1 mm) and metallurgical waste (d_{50}
359 notably lower, especially for flue dust, which was below 10 μm).

360 Regarding the distribution of As and Hg, the grain-size fractions obtained by wet-
361 sieving (63-500, 500-2000, 2000-4000 and $>4000 \mu\text{m}$) and hydrocycloning (10-63 and
362 $<10 \mu\text{m}$) were subjected to ICP-MS analysis. The concentrations of Hg and As (lines in
363 the figure) increased as the grain-size decreased, with the exception of Hg in the
364 metallurgical waste. In contrast, when percentages in weight were evaluated (bars in the
365 figure), As and Hg were predominant in the finer fractions of the flue dust while As was
366 also predominant in stupp/soot (Fig. 4).



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368 Figure 4: Distribution of Hg (left graphs) and As (right) in the representative
 369 subsamples of mining and metallurgical waste subjected to grain-size fractionation by
 370 wet sieving, and subsequent ICP-MS determination of elemental contents.

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379 3.4 *Organic pollution*

380 3.4.1 *Spills*

381 The two “pure” organic residues found in some of the former installations of the site
382 (Section 2.6) were analyzed. The first product was found in the vicinity of the
383 underground storage tanks. LC results revealed a mixture of 30% saturates, 38%
384 aromatics, and significant amounts of non-distillable fractions (15% resins and 17%
385 asphaltenes), thereby strongly suggesting a probable match with heavy fuel oil. In
386 addition, the main features of the fingerprint of the product (Fig. S2A) were a prominent
387 UCM (Unresolved Complex Mixture), a very low abundance of n-alkanes consistent
388 with a depleted C18/Phytane ratio, the abundance of hopanes and of S compounds
389 (mainly dibenzothiophenes), and the absence of light aromatics (Fig. S2B). Taken
390 together, these findings support the notion of a very long weathering process (Kruege et
391 al., 2018) and the usual preferential degradation of n-alkanes compared with isoprenoids
392 or hopanes (Gallego et al., 2007, Hagmann et al., 2019). As regards the other spillage
393 identified in the detailed reconnaissance of the study area, PCB congeners were found in
394 notable amounts (up to more than 1,000 ppm), thereby revealing the classical Aroclor
395 fingerprint (Nikonova and Gorshkov, 2011) (Fig. S2C).

396

397 3.4.2 *PAH contents in different waste and fingerprinting of organic pollution*

398 Representative aliquots of the 27 waste samples were initially screened by qualitative
399 full-scan GC-MS. The results revealed the presence of PAHs as the predominant group
400 of organic contaminants in most of the samples irrespective of the type of waste,
401 although especially abundant in mining and metallurgical waste. Other compounds
402 identified in some of the chromatograms were PCBs and aliphatic compounds, both
403 families of compounds probably linked to the dispersal of spilled products, such as
404 those described in the previous section. PAH quantification (Table 2) showed very
405 different data as regards the different type of waste; i.e. notable content in metallurgical
406 waste, moderate or low in mining waste, and negligible or below detection limit in
407 C&D debris (not included in Table 2). In all cases, 3-4-ring compounds predominated.

408

409 Table 2: PAH content in selected samples: 2-ring PAHs (naphthalene); 3-ring PAHs
 410 (fluorene + acenaphthylene + acenaphthene + phenanthrene + anthracene); 4-ring PAHs
 411 (fluoranthene + pyrene + benz[*a*]anthracene + chrysene); 5–6 ring PAHs
 412 (benzo[*b*]fluoranthene + benzo[*k*]fluoranthene + benzo[*a*]pyrene + dibenz[*a,h*]anthracen
 413 e + benzo[*ghi*]perylene + indeno[1,2,3-*cd*]pyrene).

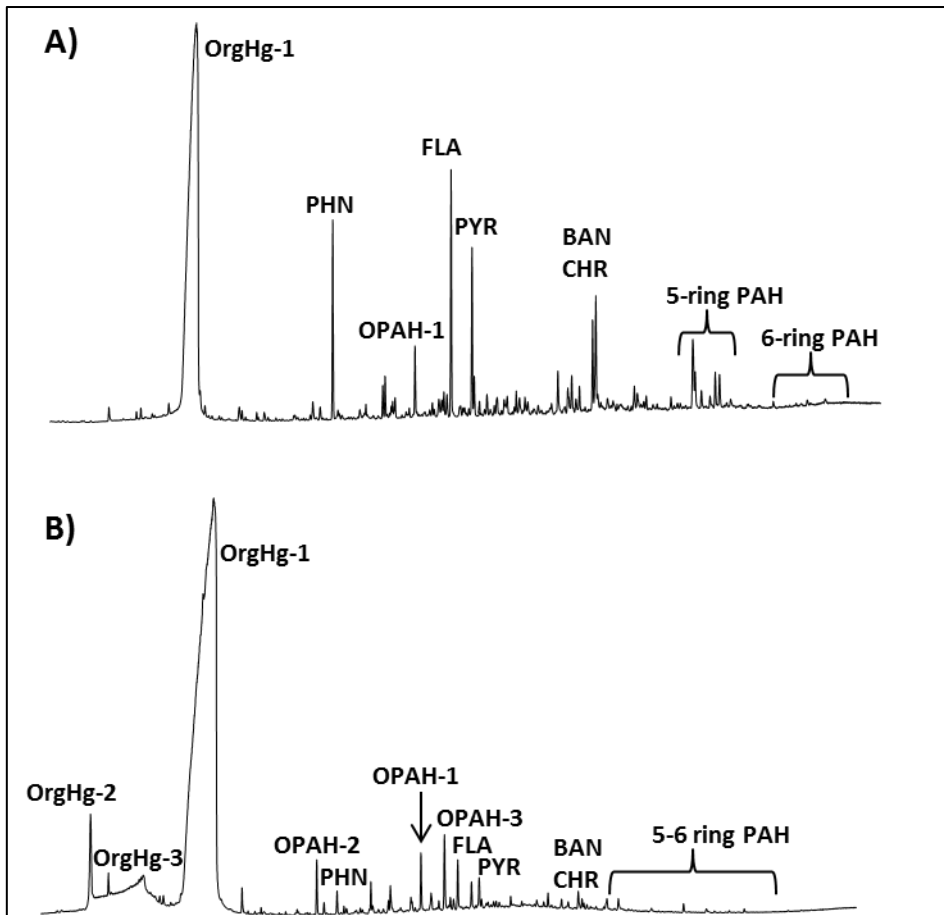
	Waste type	Samples (n)	PAHs (mg/kg)			
			2-ring	3-4 rings	5-6 rings	Total
Mining waste	Spoil heap 1	4	0.05	0.17	0.09	0.30
	Spoil heap 2	2	0.20	0.24	0.10	0.55
Metallurgical waste	Soot –Stupp	3	1.81	6.73	1.43	9.98
	Flue dust	3	0.39	3.50	1.19	5.09

414

415 The distribution of individual PAHs in the four types of metallurgical and mining waste
 416 sampled revealed a relatively homogeneous distribution, thereby suggesting a common
 417 source (Fig. S3). In addition (see Fig. 5), there was a very low abundance of alkyl-PAHs
 418 (typical of crude oil derivatives), and conversely a notable abundance of parent-PAHs,
 419 thus pointing to a pyrogenic origin (Krüge et al., 2018).

420 Finally, a detailed GC-MS fingerprint of the organic extracts of the metallurgical waste
 421 is shown in Fig. 5. Together with the previously described presence of PAHs, of note is
 422 the abundance of Hg⁰, some Hg organometallic compounds, and even Hg (II) chloride
 423 (the mass spectra of these Hg compounds is shown in Fig S4). In addition, oxygenated-
 424 PAHs (OPAHs) were also present and prominent. Several PCB congeners, and sulphur
 425 heterocycles were also identified, but in relative concentrations that were too low to
 426 produce visible peaks.

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Figure 5: A) TIC chromatographic fingerprint of soot-stupp. B) TIC chromatographic fingerprint of flue dust. Note that elemental Hg (monitored using m/z 200 and 202) is present throughout part of the GC run (plateau). (OrgHg-1: Phenylmercury propionate; OrgHg-2: Benzyl(chloro)mercury; OrgHg-3: Mercury (II) chloride; OPAH-1 Anthraquinone; OPAH-2: Fluorenone; OPAH-3: 1,8 Naphthalic anhydride; PHN: phenanthrene; FLA: fluoranthene; PYR: pyrene; BAN: benz[a]anthracene; CHR: chrysene).

446 3.4.4 Dioxins and furans

447 The analyses (Table 3) showed that the polychlorinated dibenzo-p-dioxins (PCDDs)
 448 were more abundant in mining waste than polychlorinated dibenzofurans (PCDFs).
 449 Following the sink/source classification (Dömtörövä et al., 2012; Wagrowski and Hites,
 450 2000), this profile (PCDDs > PCDFs) is classified as “sink”, for which PCDDs are
 451 predominant (a “source” profile is characterized by a PCDF predominance). In
 452 contrast, the opposite results were found for metallurgical waste (PCDFs > PCDDs),
 453 thereby strongly suggesting that dioxins and furans were produced in the
 454 pyrometallurgical process (see discussion).

455 Table 3: PCDF/PCDD congener profiles in the mining and metallurgical waste studied.
 456 (F4 = 2378-TCDF, F5 = 12378-PeCDF + 23478-PeCDF; F6 = 123478-
 457 HxCDF + 123678-HxCDF + 234678-HxCDF + 123789-HxCDF; F7 = 1234678-
 458 HpCDF + 1234789-HpCDF; F8 = OCDF; D4 = 2378-TCDD; D5 = 12378-PeCDD;
 459 D6 = 123478-HxCDD + 123678-HxCDD + 123789-HxCDD; D7 = 1234678-HpCDD;
 460 D8 = OCDD.)

PCDF Congeners	pg/g			
	Spoil heap 1	Spoil heap 2	Soot –Stupp	Flue dust
F4	1.2502	5.6955	37.3716	129.841
F5	0.6053	2.8659	22.7721	43.7394
F6	0.8046	3.5850	13.1185	13.3853
F7	1.2713	4.5573	4.2133	7.0533
F8	1.0610	3.0515	2.0404	6.1498
Total	4.9924	19.7552	79.5159	200.1688

461

PCDD Congeners	pg/g			
	Spoil heap 1	Spoil heap 2	Soot –Stupp	Flue dust
D4	0.0228	0.0632	0.0826	0.2339
D5	0.0522	0.1817	0.1257	0.3083
D6	2.1726	0	0	0
D7	1.7764	4.7866	7.6867	6.1844
D8	6.4191	15.6661	19.0688	11.5689
Total	10.4431	20.6976	26.9638	18.2955

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465 **4. Discussion**

466 The closure of Hg and As facilities in the 1970s was driven worldwide by international
467 health concerns over their use. The abandonment of activity and continuous
468 remobilization of the pollutants in the remaining waste piles make these sites significant
469 environmental hazards (Gray et al., 2002; Millán et al., 2006; Navarro et al., 2009;
470 Rieuwerts et al., 2014). In this context, when the mine and Hg-processing plant were
471 operational at La Soterraña, heterogeneous waste materials in size and nature were
472 disposed of in an unsystematic manner. In addition, the ulterior long-term abandonment
473 of the area and the partial demolition of the buildings have led to an uncontrolled and
474 uneven distribution of supplementary waste. Mining and metallurgical waste made up
475 predominantly of relatively small remnants, as shown in this work (Fig 3.), have
476 exposed an enormous surface area of sulphide/oxide minerals susceptible to reaction
477 with water and/or air. Also, the emissions of polluting steam and particles and the
478 dumping of waste have considerably affected the surrounding areas as a result of
479 mechanical dispersion (wind erosion) of the spoil heaps, together with the
480 abovementioned oxidation and lixiviation (Sierra et al., 2011). On the whole, the effect
481 of the contamination on the environmental compartments is severe (Loredo et al., 2006;
482 Matanzas et al., 2017) and risks associated with these former Hg-mining areas are of
483 concern (Ordóñez et al., 2011). Of note, a recent study of a nearby site with similar
484 characteristics estimated the release of 7 tons of As per year into a river (González-
485 Fernández et al., 2018). Pollutant mobilization gains a further dimension as the spoil
486 heaps are currently covered by incipient soil, which in turn is colonized by pioneer
487 plants and used as pasture for cattle (Matanzas et al., 2017). Furthermore, the relevance
488 of pollutant dispersal on La Soterraña site and other brownfields in the same area has
489 been recently highlighted by studies of geological records located more than 50 km
490 away from the site (Gallego et al., 2019; García-Ordiales et al., 2018). These “long-
491 distance” effects can be clearly understood when considered the total mass of Hg and
492 especially of As still present in the different waste (Table S1).

493 Our findings corroborate the high toxicity of Hg-mining and metallurgical waste
494 (Gallego et al., 2015 and references therein). In this regard, As emerges as the most
495 concerning pollutant because the abovementioned volumes and given that it shows a
496 greater leachability potential, irrespective of the conditions of the tests (Table 1).
497 Consequently, mobility/leachability have recently been the main focus of

498 nanoremediation and adsorption approaches to mitigate the impact of the types of waste
499 studied here (Ayala and Fernandez, 2019; Gil-Díaz et al, 2019) and in other industrial
500 areas (López et al., 2015).

501 In addition, our results also indicate that consideration should be given to the capacity
502 of C&D waste to contaminate. In this regard, although the average values of C&D
503 waste leachability appeared to be low (Table 2), the extremely high total content of Hg
504 and especially As in this material makes it hazardous. This reflection implies that future
505 site remediation programs must specifically address the treatment of C&D waste. In
506 fact, several studies have reported this type of waste to be a potential source of pollution
507 (Gao et al., 2015; Pecorini et al., 2017; Roussat et al., 2008; Townsend et al., 2004).
508 Indeed, it can be considered a potential repository of hazardous pollutants (Saca et al.,
509 2017), even in highly problematic areas, such as those affected by radioactive waste
510 (Diedenhofen et al., 2006). Several authors have also demonstrated non-negligible
511 release of contaminants from clean and mixed C&D waste on contaminated sites
512 (Butera et al., 2014). In this regard, the adsorption of Hg in several construction
513 materials has been described (Gao et al., 2014). In concordance with our work, Prieto-
514 Taboada et al., 2013 also demonstrated how buildings in an industrial port of dense
515 traffic were “supplied” with both inorganic and organic contaminants. Contrary to other
516 studies (Jang and Townsend, 2001), the presence of organic contaminants in our C&D
517 waste samples was not significant. However, the abundance of PAHs in metallurgical
518 waste stock-piled in the vicinity of the ruined buildings and the presence of fuel spills,
519 and probably of underground storage tanks still not located, suggest that future
520 remediation work should also consider organic compounds in C&D waste.

521 The speciation studies revealed that Hg and As persisted in the environment, mainly in
522 the form of their less toxic chemical species. On first inspection, inorganic Hg appeared
523 to predominate in the samples (Table 1). However, analysis of organic pollutants (see
524 below) revealed unexpected results that disprove this notion. In turn, As (III) had a
525 notable presence in the metallurgical waste. Indeed, a specific study of the nature of this
526 compound (Lara-Gonzalo et al., 2019) confirmed the presence of the oxide arsenolite as
527 the mineral containing As (III), as suggested in previous findings (Gallego et al., 2015;
528 Haffert and Craw, 2009). Our results also revealed that the most problematic residues
529 (metallurgical waste) are found in the fine-grain fraction, in coherence with their former
530 recovery in sedimentation chambers used in the pyrometallurgical process. In fact, for

531 these residues, d_{10} can even exceed 50% (Fig. 2), thereby increasing the likelihood of
532 suspension-deposition and indicating that dust dispersal at the site is a critical issue
533 from a toxicological perspective.

534 Regarding organic pollutants, the highest PAH concentrations were found in flue dust,
535 i.e. the waste produced at the end of the metallurgical process, inasmuch they were
536 carried with the residual gases through the ducts and chimneys and were condensed
537 sequentially as the temperature decreased. In addition, stupp/soot waste also carried
538 significant concentrations of PAHs (Table 2), as the fuel used (fuel oil and coal) in the
539 old kilns was often in physical contact with the mineral during the roasting process. The
540 abovementioned oxide arsenolite (Haffert and Craw, 2009) was the main by-product of
541 this process, whereas Hg was mostly recovered in the condensation systems (Gallego et
542 al., 2015). Nevertheless, organometallic compounds and an unburnt residue of the
543 natural organic matter in the ores were also produced during the process and,
544 consequently, the resulting waste contains both inorganic and organic components.
545 Indeed, the GC-MS approach revealed key molecular features of the foreseen organic
546 pollutant inputs. Several areas where fuel and PCB release/spills had occurred were
547 identified. In addition, partially biodegraded fuel oil and mainly pyrogenic combustion-
548 derived parent PAHs imprinted their signatures in the GC-MS data. Although the spilled
549 fuel was found in a specific area, it was clearly evidenced (absence of alkyl-PAHs,
550 PAHs distribution in Fig. S3, etc.) that the main source of PAHs in the area was
551 pyrogenic. In this regard, the predominance of the pyrogenic component was verified in
552 a complementary way by PAH ratios (Boehm, 2006; Yunker et al., 2002), such as
553 Anthracene/Anthracene + Phenanthrene (0.09 ± 0.03 averaged in the mining and
554 metallurgical waste) and Fluoranthene/Fluoranthene + Pyrene (0.61 ± 0.05), both
555 indicating that PAHs originated from combustion (pyrogenic source). Together with
556 PAHs, the abundance of Oxy-PAHs in the fingerprint of the chromatograms of the
557 metallurgical waste was also noteworthy (Fig. 5). These compounds may be as
558 abundant, persistent and toxic in the environment as their parent PAHs, and they are
559 usually linked to weathering (biodegradation mainly) processes wherever PAHs are
560 abundant (Idowu et al., 2019).

561 Studies addressing the presence of Hg-organometallic compounds (Fernández-Martínez
562 et al., 2014; Rumayor et al., 2017) pointed to the link between Hg and organic matter.
563 Specifically, Hg speciation obtained by thermal desorption of samples of the

564 metallurgical waste studied here revealed the abundance of Hg associated with organic
565 matter, also “Hg-rich” pyrite (Hg-FeS_2), Hg^0 and HgCl_2 (Rumayor et al., 2017). These
566 results are coherent with the presence of volatile Hg species (Hg^0 , HgCl_2) in the GC-MS
567 study (see Fig. 5). However, our GC-MS data also highlighted the presence of
568 organometallic compounds (other than methyl- or ethyl-Hg), namely phenylmercury
569 propionate ($\text{C}_9\text{H}_{10}\text{HgO}_2$) (previously described in Gallego et al., 2015) and
570 benzyl(chloro)mercury ($\text{C}_7\text{H}_7\text{ClHg}$), and their notable abundance, especially in the flue
571 dust (Fig. 5). These highly toxic compounds (ECHA, 2011) may have been formed
572 through the combination of Hg and aromatic hydrocarbons during the cooling of the
573 gas/vapours flowing through the condensing system.

574 The abundance of pyrogenic PAHs in the metallurgical waste and the presence of
575 chloride in several molecules found on the site (PCBs, HgCl_2) further supported our
576 interest in studying the presence of dioxins (PCDD) and furans (PCDF). In fact, other
577 metallurgical processes have been identified as habitual sources of dioxins (Weber et
578 al., 2008). Our results (Table 3) revealed a significant concentration of PCDD and
579 especially PCDF in the metallurgical waste; i.e. the products that passed through the
580 former ducts and chimneys are a notable source of these extremely toxic compounds. In
581 this regard, it should be pointed out that our data (Table 3) imply that the toxic
582 equivalents (WHO PCDD/F-TEQ) for flue dust are 29.032 pg I-TEQ/g. This TEQ value
583 is not only significantly higher than those obtained recently in soil samples from a
584 heavily polluted brownfield located 20 km north of La Soterraña but also greater than
585 the regional backgrounds (values both below 20 pg I-TEQ/g, see Lara-Gonzalo et al.,
586 2015). A comparison of our findings with the concentrations of PCDD and PCDF
587 reported in other industrial/polluted areas (Dömötöróvá et al. 2012, Rada et al. 2015)
588 strongly suggest a notable contaminant load in La Soterraña. In the light of these and
589 other lines of evidence (source profile as indicated in section 3.4.4, levels markedly
590 above the background, etc.), we conclude that PCDD and PCDF in this brownfield were
591 generated during the pyrometallurgy of Hg-ores.

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595 **5. Conclusions**

596 The brownfield of La Soterraña is a paradigmatic example of the combined effect of the
597 abandonment (> 45 years) of mining/industrial facilities, inappropriate waste disposal,
598 and the presence of highly toxic contaminants. Using a multifaceted, environmental
599 forensics study carried out by means of a combination of advanced analytics,
600 cartographic techniques and a historical study, we have identified features commonly
601 overlooked by conventional analyses. The methodology described is likely to find
602 application in other complex brownfields worldwide.

603 Here we focused on the wide range of waste that had been disposed of at La Soterraña
604 over decades. These waste materials are subjected to weathering and thus are the main
605 source of pollution on the site. Our results indicate that a notable volume of C&D waste
606 should be considered hazardous, although metallurgical waste (the highest contents of
607 Hg and As) and mining waste (the highest volume) are of maximum concern. The site
608 revealed a complex speciation of As and Hg, and the presence of unexpected Hg-
609 aromatic compounds (addressed by GC-MS molecular techniques).

610 The acquisition of complementary information from various analyses also revealed the
611 occurrence of PAHs, oxy-PAHs and PCBs, together with the inorganic and organic
612 compounds abovementioned. In this context, this study is the first to identify PCDD and
613 PCDF as by-products of the Hg-metallurgical process. The accumulation of organic and
614 organometallic pollutants strongly suggests that the study site poses an extreme
615 environmental risk, even irrespective of its high concentration of inorganic As and Hg.

616 We consider that once the most hazardous waste has been removed from the site, the
617 effectiveness of remediation strategies will be complicated by the presence of multiple
618 pollutants and their synergistic and antagonistic interactions. In this context, the
619 identification of target analytes through the forensics approach followed here will now
620 facilitate future remediation measures.

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The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: