

1 **Microbial mat contribution to the formation of an evaporitic**  
2 **environment in a temperate-latitude ecosystem**

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21 **Abstract**

22 An evaporitic environment is characterized by having high salinity, climatic, and hydrological  
23 factors that promote a negative water balance; however, biological factors may also influence  
24 their development. Modern coastal flat Paso Seco (40°33'S; 62°14'W) is located in a semi-arid  
25 region with low precipitation and dry winds coming mainly from the NW. The site is an old tidal  
26 channel, which nowadays behaves like a shallow coastal saline-like basin, separated from the sea  
27 by a sand barrier, which the sea periodically overcomes, flooding the flat with eventual water  
28 evaporation. Microbial mats of up to 1 cm thick colonize the sandy sediments of this evaporitic  
29 environment. Water samples were taken during five field trips (2017-2018) from interstitial  
30 water of the flat, a tidal creek that crosses the flat, and two shallow tidal depressions (TDs)  
31 within the flat with different degrees of evaporation. In comparison to the sea, the maximum  
32 salinity values measured in Austral spring (September 2017) in the tidal creek were doubled,  
33 tripled in interstitial water, and 5.9 to 8 times higher in TDs. Ionic concentration denotes that  
34 evaporite chemical divides are followed as water evaporates, corresponding to the presence of  
35 CaCO<sub>3</sub>, gypsum and halite found in TDs. On-site permeability of microbial mat-covered surfaces  
36 presented semi-pervious properties. Microbial mat presence is condition for CaCO<sub>3</sub>, gypsum, and  
37 halite precipitation as they allow for water retention and its consequent evaporation due to the  
38 impermeability they confer to the sedimentary surface. Thus, microbial mats are a biological  
39 factor affecting the development of an evaporitic environment.

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41 Keywords: Saline basin; Evaporite; Calcium carbonate; Water retention; Salinity

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43           **1. Introduction**

44           Evaporites in a broad sense are a “group of sedimentary deposits whose origin is largely  
45 due to evaporation” (Twenhofel, 1950). In the case of marine evaporites, seawater molecules  
46 evaporate following the Usiglio sequence for salt precipitation, resulting first in aragonite and  
47 calcite precipitation, then in gypsum and later halite precipitation (Dean, 2013; Usiglio, 1849).  
48 Present-day marginal marine evaporite deposits are mainly located within two subtropical high-  
49 pressure belts (between 15° and 35° latitude from the equator), where there is limited rainfall and  
50 elevated evaporation (Babel and Schreiber, 2014). Examples of these environments are the  
51 sabkhas of the Red Sea coastal tidal flats of Saudi Arabia and Abu Dhabi, and coastal saline  
52 basins in Southern Australia, among others (Aref and Taj, 2017; Butler et al., 1987; Orti, 2010;  
53 Shearman, 1978; Strohmenger et al., 2010; Warren, 2016a, 2016b).

54           Marginal marine evaporites are being deposited nowadays in coastal salinas (sub-sea level  
55 depressions separated from the sea by a barrier and supplied by seawater, in which subaqueous  
56 evaporites are formed; Warren, 2016a) and coastal sabkhas (supratidal flats in which intra-  
57 sedimentary evaporites are formed; Warren, 2016). A negative water balance is a condition for  
58 the formation of these evaporite environments, where outflow of water should be higher than the  
59 inflow of any kind of water – marine or otherwise (Babel and Schreiber, 2014).

60           Temperature, salinity, and pH are the main physicochemical parameters that characterize  
61 evaporite environments, and these parameters must vary in a way that they promote  
62 supersaturation and precipitation of the ions in solution (Babel and Schreiber, 2014). Low  
63 relative humidity, for example, is needed for halite precipitation, while in a restricted evaporite  
64 setting, the presence of gypsum suggests evaporation at temperatures lower than 30 °C  
65 (Kinsman, 1976; Krumbein and Garrels, 1952). However, beyond climatic and purely physical

66 hydrological features, biological factors could also influence the hydrology of a site, indirectly  
67 contributing to the conditions needed for the formation of evaporite environments.

68         Microbial mats are photosynthetic associations of prokaryotes, archaea and eukaryotes,  
69 dominated mainly by cyanobacterial species, which commonly colonize supratidal sediments,  
70 flooded intermittently. Cyanobacteria are the primary producers of extracellular polymeric  
71 substances (EPS) that help trap, bind and precipitate sediments and minerals (Stolz, 2000). This  
72 physicochemical matrix is composed of secreted polysaccharides, proteins, lipids, DNA, and  
73 vitamins (Flemming and Wingender, 2001), offering several uses for bacteria, such as chelation  
74 of toxic ions and sheltering from harmful UV radiation (Decho, 2011; Or et al., 2007). EPS has  
75 indirectly been shown to protect from desiccation for its high water retention capacity, as  
76 mutations in EPS biosynthesis genes decrease desiccation tolerance and such water conditions  
77 result in increased EPS production (Schnider-Keel et al., 2001; Alvarez et al., 2004; Or et al.,  
78 2007; Decho, 2011). The combination of bacteria, EPS and trapped sediments that form  
79 microbial mats creates a poorly permeable sediment surface (Babel, 2007), slowing water  
80 percolating from the surface over periods of time. That fact is reflected in the flat of Paso Seco  
81 by the formation of special microbially sedimentary structures, such as gas domes (Noffke,  
82 1999), which prevent the escape of gases (Cuadrado et al., 2013).

83         However, although in many evaporitic environments around the world microbial mats are  
84 being studied (e.g., in sabkhas; Aref and Taj, 2017, 2018), their presence is only considered in  
85 the light that they are present as a consequence of the hydrodynamics of the site, and how they  
86 chemically affect the environment (Babel, 2004; Long et al., 2009), but do not incorporate into  
87 their studies the possible effect they have in modifying the hydrodynamics of the site as well.

88 In evaporite environments, water retention is vital for precipitation of evaporite minerals.  
89 Without water retention, the dissolved ions needed for evaporite formation would percolate into  
90 the sediment with water. Thus, in this work, we propose that the presence of microbial mats in  
91 the temperate coastal flat Paso Seco (Argentina) is essential for the formation of the evaporite  
92 deposits found in the site. Although the local climatic factors are enough for evaporite formation,  
93 especially the constant and powerful drying winds, without microbial mats there would not be  
94 enough water retention to allow for the concentration of ions and their precipitation through  
95 evaporation. These results document the biogeochemical control in hydrology in a  
96 geomorphological restricted basin and how biological factors could potentially change the  
97 hydrological conditions of an evaporitic environment.

## 98 **2. Material and methods**

### 99 *2.1. Study site and water sampling period*

100 Water sampling was performed to characterize the environment on May, September and  
101 October 2017, and March and June 2018 at two stations, 3 and 4, and from a tidal creek located  
102 close to St. 3 ('St. 3' and 'St. 4', and 'St. 3 Tidal Creek', respectively from now on; Fig. 1), in a  
103 temperate coastal flat at Paso Seco, Argentina (40°33'S; 62°14'W; Fig 1), located in northern  
104 Patagonia. The climatic region where Paso Seco is situated is considered as semiarid, according  
105 to the recent climatic classification of the Argentinian Pampas proposed by Aliaga et al. (2017).  
106 The average maximum high temperature is 30 °C in summer and 12 °C in winter, while average  
107 low temperatures are 14 and 2 °C, respectively ('Servicio Meteorologico Nacional, ' n.d.). The  
108 average annual rainfall is 300 mm but is characterized by large space-time variation, raining for a  
109 total of 80 days throughout the year in average (Ferrelli et al., 2012, 'Servicio Meteorologico  
110 Nacional, ' n.d.). Potential evapotranspiration in the region exceeds precipitation, with aridity

111 strong enough to limit vegetation development. The SW Atlantic Ocean moderates the  
112 differences between summer and winter, preventing extreme hot and cold conditions. Strong NW  
113 winds prevail in the area, averaging a maximum of  $44 \text{ km h}^{-1}$  with frequent gusts up to  $100 \text{ km}$   
114  $\text{h}^{-1}$  in the last five years (unpublished data). Geomorphologically, the study area comprises a  
115 sizeable closed basin (about  $2.5 \times 0.3 \text{ km}$ ), colonized by microbial mats and choked by a  $1.8 \text{ km}$ -  
116 wide sand spit formed by NE longshore sediment transport. Whenever strong winds combine  
117 with the effect of the spring tides or storms, seawater is pushed through the sand spit to enter the  
118 water into the basin, resulting in the flooding of the study area (Cuadrado et al., 2015). Sediment  
119 of the flat is predominantly sandy, and most of the surface of the flat is covered by microbial  
120 mats. The force of the flooding usually results in a large number of elongated tidal depressions  
121 (TDs, Fig. 2a), as classified by Perillo (2018), re-colonized by microbial mats. Other microbial  
122 sedimentary structures such as erosional pockets (Noffke, 1999) where the sediments have little  
123 colonization and can be found covered by ripples (Fig. 2b). The study site was found flooded  
124 both in May 2017 and June 2018 during field trips. Weather data (air temperature and wind  
125 speed) were taken from the Global Forecast System model provided by the website Windguru at  
126 the closest weather station to the site, located at El Condor, Río Negro, Argentina ( $40^\circ 55' 43.32''$   
127 S,  $64^\circ 23' 31.92''$  W).

## 128 *2.2. Water level and permeability*

129 Water-level fluctuations in the groundwater and also by tidal inundation of the study area  
130 have been measured for five-years in St. 3 and St. 4 (Fig. 1), with a HOBO water level logger  
131 (model U20), placed  $40 \text{ cm}$  below the flat surface. The sensor recorded water level and  
132 temperature every 10 minutes. The HOBO sensor at St. 4 had been stolen before the March 2018  
133 field trip, and thus, we lack data from October 2017-March 2018.

134 Permeability was measured both *in situ* on August 2018 at St. 4 and in the lab. Field  
135 measurements consisted on inserting a PVC cylindrical corer (19 cm long, inner diameter = 4  
136 cm) into a water saturated surface covered with microbial mats with sand layers underneath until  
137 a 5.6 cm depth. Both in the field and lab, the core was filled to the top with seawater and the rate  
138 of infiltration of the water was measured in duplicate. This measurement was also repeated with  
139 a larger PVC cylindrical corer (1 m long, inner diameter = 11 cm), in two locations: where there  
140 was clear evidence of microbial mat presence and in an erosional pocket where there was little  
141 microbial mat colonization (Fig. 2b). For the larger PVC corer, we considered the time for the  
142 first 20 cm of the water column to permeate to perform our calculations, assuming a constant air-  
143 pressure. In the lab, unsaturated samples with and without the superficial microbial mat layer  
144 were tested. The size of lab samples was 11 x 7 cm with a thickness of 5 cm, and the same  
145 smaller field PVC core was inserted into the samples. Surficial microbial mats had a thickness of  
146 7 mm.

### 147 2.3. Water and petrographic analysis

148 Water samples were routinely taken from St. 3, St. 4 and St. 3 Tidal Creek. Additionally,  
149 two TDs with at least 4 cm of water were chosen in St. 4 (St. 4 TD), with and without the  
150 presence of halite, and water samples from these TDs were also taken. Water sample salinity was  
151 determined gravimetrically according to Strickland and Parsons (1972). Main seawater ion  
152 concentrations (i.e.  $K^+$ ,  $Ca^{2+}$ ,  $Mg^{2+}$ ,  $HCO_3^-$ , and  $SO_4^{2-}$ ) were determined following standard  
153 methods from APHA (2005). A 10 % HCl solution was pipetted on site over the sediments, and  
154 carbonate presence was observed whenever effervescence occurred as a qualitative measure.  
155 Sediment cores were taken from the mat in cylindrical PVC tubes (7 cm long, inner diameter = 3  
156 cm), which were cut longitudinally, and then, several layers were obtained at different depths to

157 prepare petrographic thin sections and for scanning electron microscopy with Energy-dispersive  
158 X-ray spectroscopy (SEM with EDS; JEOL35 CF 8, Tokyo). Samples for SEM were fixed and  
159 washed in a mixture of 2.5 % glutaraldehyde in Sorensen phosphate buffer. Afterwards, samples  
160 were dehydrated in an acetone series (10 to 80 %), and last, they were dried by critical point and  
161 coated with gold. Petrographic analyses were carried out by the use of analyzed under a NIKON  
162 SMZ 1500 polarization microscope and a Nikon Eclipse 80i fluorescence optical microscope.  
163 The nature of carbonate was determined by carbonate staining method described by Dickson  
164 (1966), which consisted in dyeing the thin sections with alizarin red solution, hydrochloric acid  
165 and  $K_4Fe(CN)_6$ . Halite crystals were determined in the field by their characteristic appearance at  
166 a macroscopic scale. Gypsum precipitation was confirmed by XRD.

### 167 **3. Results**

#### 168 *3.1. Permeability of microbial mats*

169 Two sites with different degree of sediment colonization were chosen to compare the  
170 permeability over the Paso Seco flat. One site has a thick surficial microbial mat (~ 1 cm),  
171 characterized by several old microbial mats beneath the surface, that covers the majority of the  
172 flat; and the other was an erosional pocket, where the sediment is less colonized by microbes.  
173 The average rate of permeation of infiltrated water in the field measured with the 1 m long PVC  
174 tube was  $37.63 \text{ cm}^3 \text{ min}^{-1}$  in the presence of microbial mats (Fig. 2a and c). In contrast, the water  
175 permeated at a rate of  $191.87 \text{ cm}^3 \text{ min}^{-1}$  over an erosional pocket (Fig. 2b and c). The same  
176 experiment was performed in the presence of microbial mats with a smaller PVC tube (19 cm  
177 long), and more than 66 % of the water volume remained in the PVC tube after 78 min had  
178 passed ( $0.46 \text{ cm}^3 \text{ min}^{-1}$ ). For the samples taken to the lab (also measured with the shorter PVC  
179 tube), similar results of different magnitudes were found (Fig. 2c). The time water took to



180 permeate through the sample when the surficial microbial mat was removed (less than 3 minutes  
181 or an average of  $79.1 \text{ cm}^3 \text{ min}^{-1}$ ) was significantly shorter than both the measurements in the  
182 field and the untouched sample in the lab (more than 75 minutes or  $1.92 \text{ cm}^3 \text{ min}^{-1}$ ; Fig. 2c).  
183 Numerical differences between the different PVC tubes could be a result of the higher  
184 hydrostatic pressure in the larger tubes than in the smaller ones (capillarity phenomenon).

### 185 *3.2. Fluctuations in water level and temperature*

186 The characteristic fluctuation of the water level over the flats at Paso Seco and as  
187 groundwater can be seen in figure 3, where the floods and the exposed periods are recognized  
188 along three years. The water temperature in Austral winter was  $< 10^\circ\text{C}$  and rose to above  $20^\circ\text{C}$   
189 in Austral summer. During winter, the groundwater level is close to the flat, and might be  
190 submerged by a sea-water lamina during several weeks after a flooding. In contrast, in summer  
191 (identified with rectangles in Fig. 3) the flat is sometimes flooded more often by seawater, but  
192 the flat is exposed to solar radiation during several days when the groundwater level fluctuates  
193 up to 0.50 m below the flat. In those occasions periods of evaporation (PE) might be possible.

194 Before the September 2017 field trip, the flat as a whole was occasionally covered with  
195 seawater (four times) during the two previous months before sampling, with water height less  
196 than 10 cm (Fig. 4a), and water temperature slowly raising from  $10$  to  $15^\circ\text{C}$  in the days leading  
197 to the sampling date. Water last flooded the flat ten days before sampling, and subsequently the  
198 water level dropped, maintaining most of the flat exposed during eight days (PE, Fig. 4a) and the  
199 remaining seawater was only preserved in several TDs. During September 2017, weather  
200 conditions show rising high daily temperatures with the maximum temperature ranging between  
201  $20$ - $25^\circ\text{C}$  five days before the sampling (Fig. 4b) and moderate winds between  $20$  to  $36 \text{ km h}^{-1}$ ,  
202 predominately coming from the NW (Fig. 4c). These winds in the region are very dry,

203 reinforcing the evaporation process. Thus, these conditions were conducive to a period of  
204 evaporation (PE) on the exposed flat.

#### 205 *3.4. Evaporite presence*

206 During the September 2017 field trip, TDs at St. 4 exposed rings of precipitated salts that  
207 marked where the seawater had been before evaporating (Fig. 5a). Carbonate was detected in the  
208 flat by HCl test (confirmed as calcite by thin sections), and sheets of gypsum (confirmed by  
209 XRD) and halite crystals were found both in TDs and on the microbial mats after complete  
210 evaporation (Fig. 5b-d).

211 SEM with EDS and petrographic analyses of thin sections of the microbial mats were  
212 carried out displaying similar results. EDS analyses demonstrated that repeated thin layers of  
213 cation  $\text{Ca}^{2+}$  are present in a sedimentary corer (Fig. 6). Likewise, petrographic thin section  
214 exhibited 100-200  $\mu\text{m}$ -thick layers of dense micritic calcite within the microbial mats (Fig. 7).

#### 215 *3.5. Water salinity and composition*

216 The values for the three sites (St. 3, St. 3 Tidal Creek, and St. 4) varied throughout the  
217 study period (Fig. 8a); however, St. 3 always had the highest salinity values, followed by St. 4,  
218 and lastly, St. 3 Tidal Creek, which still presented salinity values much higher than the expected  
219 seawater value. The lowest value,  $49.4 \text{ g L}^{-1}$ , occurred in St. 4 in May and the highest salinity,  
220  $115 \text{ g L}^{-1}$ , was found in St. 3 in September (Austral spring; Fig. 8a). In June 2018, the salinity of  
221 the water flooding the flat was  $62.8$  and  $52.4 \text{ g L}^{-1}$ , at St. 3 and St. 4, respectively. At two of the  
222 sampling dates (September 2017 and March 2018), water samples were also taken from TDs –  
223 with and without halite crystals –having salinities higher than  $200 \text{ g L}^{-1}$  (Fig. 8b).

224 Water from St. 3, St. 3 Tidal Creek, St. 4 and two TDs (with and without halite  
225 precipitation) was collected in September 2017 and analyzed to obtain its ion concentration

226 (Table 1). The comparison between TDs shows that as salinity increased at the different sites  
227 measured with regards to seawater salinity,  $\text{Ca}^{2+}$  concentrations increased reaching a maximum  
228 value in a clear TD (with no halite precipitated) (Table 1). For salinities higher than  $250 \text{ g L}^{-1}$ ,  
229 where precipitated halite was found,  $\text{Ca}^{2+}$  concentrations dropped to values of less than  $30 \text{ mEq}$   
230  $\text{L}^{-1}$  (Fig. 8c). Conversely, both  $\text{Mg}^{2+}$  and  $\text{SO}_4^{2-}$  concentrations rose as salinity increased and  
231 always remained higher than the one for  $\text{Ca}^{2+}$  (Table 1). Equally, the concentration of  $\text{HCO}_3^-$  was  
232 always lower than the one for  $\text{Ca}^{2+}$ . The ratio of  $\text{Mg}^{2+}/\text{Ca}^{2+}$  also behaved in a similar fashion,  
233 dramatically increasing whenever precipitated halite was found.

#### 234 **4. Discussion**

235 Paso Seco's saline-like behavior during summer is found outside the subtropical high-  
236 pressure belts where most marginal marine evaporitic environments are usually found (Babel and  
237 Schreiber, 2014). Similar to the Mediterranean sabkhas (Warren, 2016a), Paso Seco is a  
238 periodically flooded coastal flat, with little vegetation (Cuadrado et al., 2015). Salinity of the  
239 pore water in St. 3 when the flat is subaerially exposed is within the same range as the pore water  
240 of the saline zone (between  $95$  and  $160 \text{ g L}^{-1}$ ) in sabkhas located between Sulaymaniya lagoon  
241 and Sharm Obhur (Bahafzullah et al., 1993), and the Al-Kharrar sabkha (Aref and Taj, 2017),  
242 among others. St. 3 receives water both from the sea and the St. 3 tidal creek, which lead to salt  
243 precipitation due to water evaporation, and later, as the next flood enters the flat, these salts are  
244 then diluted, increasing the saltwater values. The coastal flat, also in a similar fashion to sabkhas,  
245 is covered with a thick microbial mat layer (up to  $1 \text{ cm}$ ), which shows little grazing as salinity in  
246 most stations and on the flat is higher than what gastropods and other predators can stand  
247 (Cornée et al., 1992). The main difference between the location in the present study and low-  
248 latitudes sabkhas characterized by arid climate mainly revolves around the higher amount of

249 rainfall that the study site receives yearly. This condition would affect the redissolution of the  
250 evaporites, explaining the small amounts of intra-sedimentary gypsum and halite found on site.

251 Weather conditions at the Paso Seco site are also favorable for water evaporation, as the  
252 coastal flat surface is exposed to evaporation because of the rise in air temperature ( $>20\text{ }^{\circ}\text{C}$ ),  
253 reinforced by moderate wind speed ( $20\text{-}36\text{ km h}^{-1}$ ; Fig 6c) during PE. Water evaporation is  
254 evidenced by, first, the evaporation rings in the coastal flat (Fig. 5a); and second, by the visual  
255 presence of both gypsum and halite (Fig. 5b, c and d), as well as  $\text{CaCO}_3$  (Fig. 6), as the Usiglio  
256 sequence predicts (Usiglio, 1849; Warren, 2016c).  $\text{CaCO}_3$  would crystallize when the solute  
257 concentration reaches twice the concentration of seawater, followed by gypsum and later, by  
258 halite precipitation (Usiglio, 1849), as was sustained by the chemical analysis (Table 1).

259 Hardie and Lowenstein (2003) also followed the ion concentrations in seawater as water  
260 evaporates and determined two important divides for the formation of brines of different  
261 composition: the  $\text{CaCO}_3$  and the gypsum divides. As seawater evaporates,  $\text{Ca}^{2+}$  concentrations  
262 increase up to the point where, if the concentration of  $\text{Ca}^{2+}$  is higher than the concentration  
263  $\text{HCO}_3^-$ , all  $\text{HCO}_3^-$  precipitates as  $\text{CaCO}_3$  and the remaining  $\text{Ca}^{2+}$  will start precipitating as  
264 gypsum (Babel and Schreiber, 2014; Hardie and Lowenstein, 2003), as can be seen in the Paso  
265 Seco site (Fig. 9 and Table 1). Gypsum precipitation begins at salinities of  $150\text{ g L}^{-1}$  and  
266 continues until salinity reaches between  $290$  and  $320\text{ g L}^{-1}$  (Logan, 1987; Usiglio, 1849),  
267 salinities that are reached at the different TDs. This precipitation is also favored by the presence  
268 of microbial mats in the surface (Oren, 2010). After gypsum precipitation, when the  
269 concentration of  $\text{SO}_4^{2-}$  is high as can be found in Paso Seco (Table 1), the concentration of  $\text{Ca}^{2+}$   
270 in water falls abruptly (Babel and Schreiber, 2014; Hardie and Lowenstein, 2003) because it  
271 precipitates as  $\text{CaSO}_4$  over the inundated flat (Figs. 5 and 9). At the same time, the ratio of

272  $Mg^{2+}/Ca^{2+}$  in the evaporating water rapidly increases after the precipitation of gypsum, which we  
273 have also seen both in the most evaporated samples (TDs; Table 1 and Fig. 8). That fact was  
274 corroborated in sedimentary cores where thin layers of calcite were recognized by EDS and  
275 petrographic analysis (Figs. 6 and 7).

276 As seen with the presence of evaporites at the site, Paso Seco behaves as a saline-like basin  
277 that has evolved to total dryness during summer months (Fig. 10) (Babel, 2007; Babel and  
278 Schreiber, 2014). From September until April (Austral spring, summer and beginning of fall),  
279 seawater enters the coastal flat as it overcomes the sand barrier during storms. The main outflow  
280 of seawater is through ebbing; however, the presence of microbial mats covering all the surface  
281 of this saline-like flat is able to retain enough seawater for it to be evaporated during periods of  
282 warm temperatures and moderate winds that occur when the flat is exposed. This effect is  
283 mediated through the semi-pervious property that the microbial mats convey to the surface.  
284 Additionally, the alkaline pH microbial mats develop help maintain seawater pH and thus, favors  
285  $CaCO_3$  precipitation (Dupraz et al., 2009). However, the periodic water input from each storm  
286 dissolves the precipitated gypsum and halite in the TDs, of which only few precipitates remain  
287 (Fig. 6), and most probably modifies the thickness of the precipitated  $CaCO_3$  layer (Babel and  
288 Schreiber, 2014; Sanford and Wood, 1991).

289 Microbial mats have poor permeability as can be seen already at the Paso Seco site by gas  
290 domes formed as typical microbial sedimentary structures. These structures are generated from  
291 intra-sedimentary gas pressure producing a flexible deformation microbial mat preventing the  
292 escape of gases (Cuadrado et al., 2015; Gerdes, 2007). Also, with a simple field test, we  
293 determined that with microbial mats covering the surface, the coastal flat behaves as semi-  
294 pervious surface, similar to how unconsolidated very fine sands, silts, and loams behave (Bear,

295 2013). The copious sheath material secreted by *Microcoleus chthonoplastes*, a species found in  
296 Paso Seco's microbial mats (Cuadrado and Pan, 2018), provides cohesiveness to the sediment  
297 and helps retain water (Stal, 2000; Stolz, 2000) where microbial mats are present. In contrast, the  
298 erosional pockets, structures characterized by bottom ripples formed by tidal currents due to less  
299 colonized sand (Noffke, 1999), water permeability is higher, determining that permeability is  
300 related to sediment colonization. Moreover, we also obtained a similar result in the lab as in the  
301 field, and when the microbial mat was removed, the remaining layer permitted water permeation  
302 at a significantly higher rate, which leads to conclude that without the microbial mat, the  
303 behavior of the coastal flat would resemble a pervious surface.

## 304 **5. Conclusion**

305 This paper documents the biological effect that microbial mats have on the hydrology of a  
306 coastal area. In spite that Paso Seco is located at medium-latitudes, it behaves similar to a low-  
307 latitude coastal sabkha during the summer months, with a semi-arid climate favorable for  
308 evaporite formation. Seawater enters the coastal flat during a storm, and after most of the water  
309 has ebbed, what remains is retained on the flat thanks to the increased impermeability that the  
310 microbial mats confer to the surface, providing a significant barrier for water infiltration.

311 Without the biological action of these microbial mats, water would infiltrate through the sandy  
312 sediment instead of remaining on the surface for longer periods of time. This water retention  
313 results in evaporite formation as water evaporates when temperatures rise and drying winds  
314 prevail. Thin calcite layers in the microbial mats record the evolution of the flat to total dryness.

315 Studies made in the last thirty years are demonstrating that microbial mats are very  
316 common in all type of environments (being found also in fossil records), but seldom studied in  
317 relation to how their presence affects the hydrology. This paper confirms that the presence of

318 microbial mats provides a significant barrier for water infiltration deeply affecting results dealing  
319 with water percolation and evaporation processes.

320

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436

**Declaration of interests**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

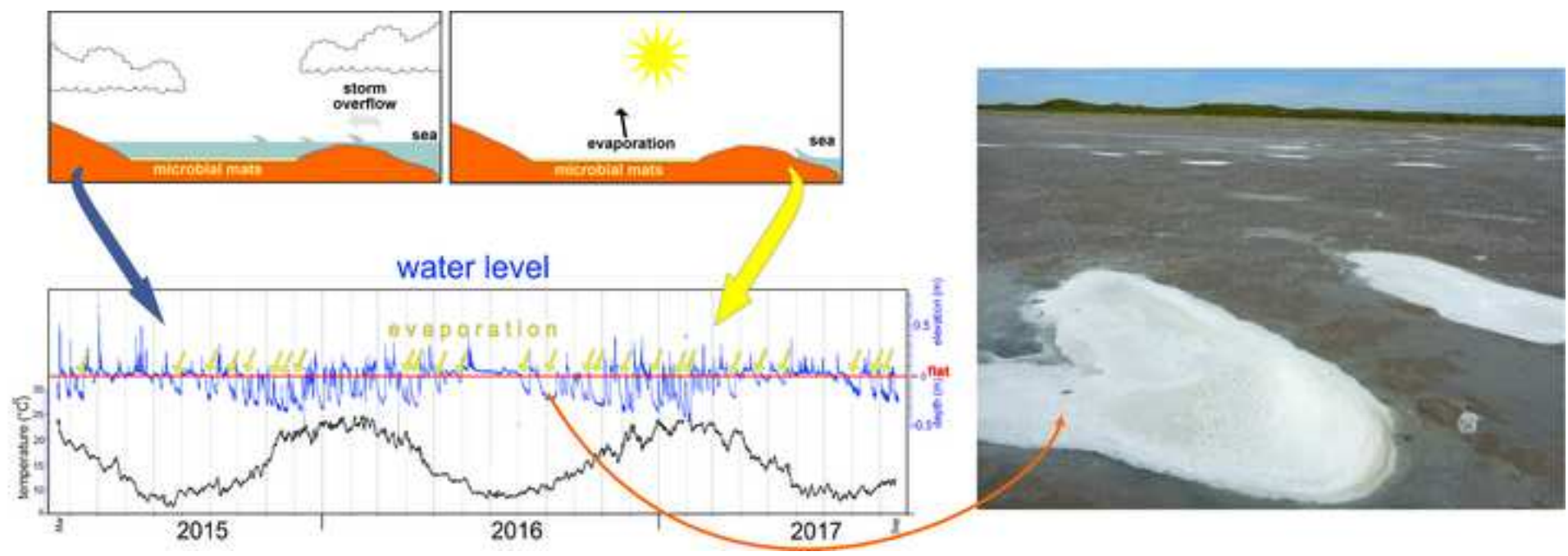


Table 1.

Station	Ion concentrations and $\text{Mg}^{2+}/\text{Ca}^{2+}$ ratio (September 2017)					
	Salinity (g L <sup>-1</sup> )	$\text{SO}_4^{2-}$ (mEq L <sup>-1</sup> )	$\text{Mg}^{2+}$ (mEq L <sup>-1</sup> )	$\text{Ca}^{2+}$ (mEq L <sup>-1</sup> )	$\text{HCO}_3^-$ (mEq L <sup>-1</sup> )	$\text{Mg}^{2+}/\text{Ca}^{2+}$
Sea	34.29	55.2	103.9	20.0	2.3	3.2
St. 3 Tidal Creek	67.95	77.4	143.5	31.5	3.0	2.8
St. 3	115.37	109.81	268.5	42.8	3.2	3.8
St. 4	91.66	113.0	244.5	38.0	5.02	3.9
St. 4 TD	202.51	310.4	604.0	53.5	1.94	6.9
St. 4 TD Halite	279.41	477.3	915.0	22.5	5.81	24.7

1           **Figure Captions**

2

3   **Figure 1. Study site.** Location and sampling stations: St. 3 Tidal creek ( $40^{\circ}38'42.20''\text{S}$ ;  
4    $62^{\circ}13'4.81''\text{W}$ ), St. 3 ( $40^{\circ}38'39.32''\text{S}$ ;  $62^{\circ}13'1.72''\text{W}$ ), and St. 4 ( $40^{\circ}38'3.32''\text{S}$ ;  $62^{\circ}12'24.85''\text{W}$ ).

5

6   **Figure 2. Tidal depressions and erosional pocket surface.** (a) Overview of the coastal flat with  
7   depressions where water remains (red arrow shows the predominant NW wind direction). (b)  
8   Erosional pocket with ripples overlain by halite. (c) Permeated seawater volume through  
9   microbial mats in the field and in the lab, contrasted to permeated volume after mat was removed  
10  (both in field and in the lab, and with different PVC tube sizes).

11

12   **Figure 3. Flooding frequency and periods of exposition.** Flood and evaporation frequency at  
13  sampling station St. 4 starting March 2015 and ending on September 2017. The blue line shows  
14  water level, with flooding events showing over the red line (flat surface), while all events below  
15  the line show the level of the water table (below the surface). The yellow squares show the  
16  austral summer and spring, periods where groundwater levels are lower and the plain is exposed  
17  for longer periods of time. The black line shows water temperature ( $^{\circ}\text{C}$ ).

18

19   **Figure 4. Weather and hydrodynamic data leading to the September 2017 sampling trip.**

20  (a) Flood and evaporation frequency at sampling station St. 4 starting August 2017 and ending  
21  on the sampling date. The green line shows water level, with flooding events showing over the  
22  red line (flat surface), while all events below the line show the level of the water table (below the  
23  surface) and also, periods of evaporation (PE). The blue line shows water temperature ( $^{\circ}\text{C}$ ). (b)



24 Daily air temperature and wind speed during the previous twenty days before sampling in  
25 September 2017. The light blue and orange lines correspond to PE and match the two lines of the  
26 same color show in (a). Black circles highlight the maximum temperature for each day.

27  
28 **Figure 5. Evaporites in field site.** Black arrows show evaporation rings during eight days after  
29 the last flooding of the coastal flat (a). Gypsum sheets (b). Halite crystals were observed both in  
30 water-filled and in completely evaporated TDs (c and d, respectively).

31  
32 **Figure 6. Energy dispersive X-ray spectrometry (EDS) maps of the same fragment.**  
33 Repeated  $\text{Ca}^{2+}$  layers in depth, some co-locating with S, probably in association with gypsum.  
34 Low presence of  $\text{Na}^+$  and  $\text{Cl}^-$  co-location (probably association with halite).

35  
36 **Figure 7. Thin section of the Paso Seco flat.** Microbial mats that colonize the flats are  
37 organized in successive laminae composed of sand, silt, clay and organic matter. Micritic  
38 carbonate laminae (arrows) are found interbedded within the microbial mats. (a) Cross polarized  
39 light. (b) Plane polarized light.

40  
41 **Figure 8. Evaporite precipitation after sea water evaporation.** (a) Salinity concentration ( $\text{g L}^{-1}$ )  
42 in the different stations. (b) Comparison between September 2017 and March 2018 on salinity  
43 in the different stations and tidal depressions. (c) Calcium concentrations ( $\text{mEq L}^{-1}$ ) in the  
44 different stations for September 2017 and March 2018 sampling dates.

45

46 **Figure 9. Evaporite precipitation pathway.** Evaporation rings after sea water evaporation (1).  
47 After  $\text{CaCO}_3$  precipitation (black arrows and dashed white lines), we observed the presence of  
48 gypsum because of the higher  $\text{Ca}^{2+}$  concentration over  $\text{HCO}_3^-$  found in the water (2). Last, halite  
49 crystals were observed because of the higher  $\text{SO}_4^{2-}$  concentration over  $\text{Ca}^{2+}$  in the TDs that had  
50 halite (3). Modified from Hardie and Lowenstein, 2003.

51

52 **Figure 10. Salina-like behavior in Paso Seco, Argentina.** During the warmer months of the  
53 year (September through April), the flat suffers periodic flooding due to storm surge overflow  
54 with the consequent accumulation of water in TD due to the presence of poorly permeable  
55 microbial mats that cover the otherwise very permeable sands. This water accumulation is  
56 followed by several days of warm weather and moderate winds, allowing for water evaporation  
57 and the subsequent evaporites found precipitated on the flat (photo).

58

Figure 1  
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Figure 2  
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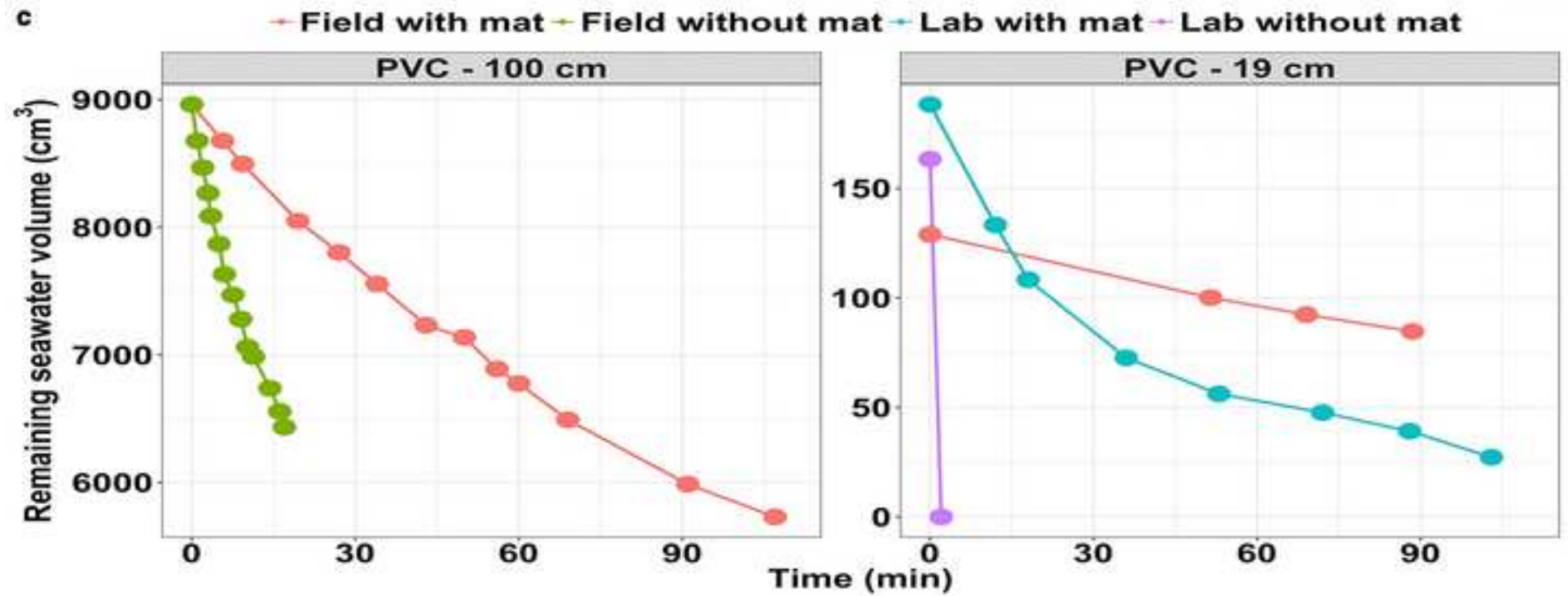
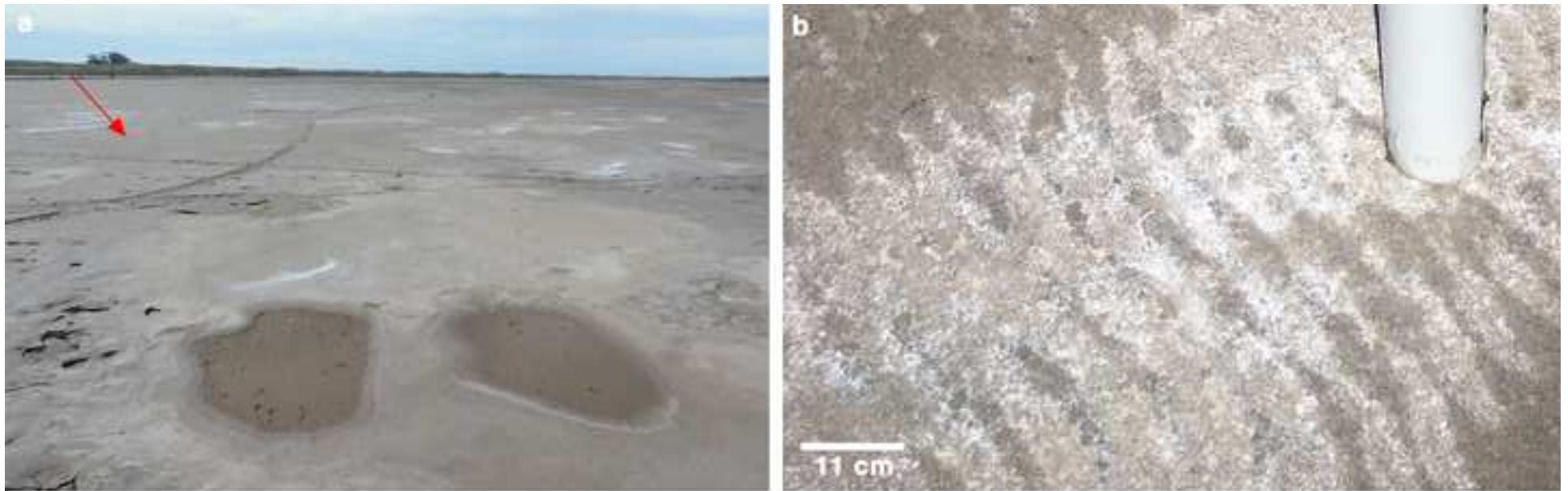


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### water level

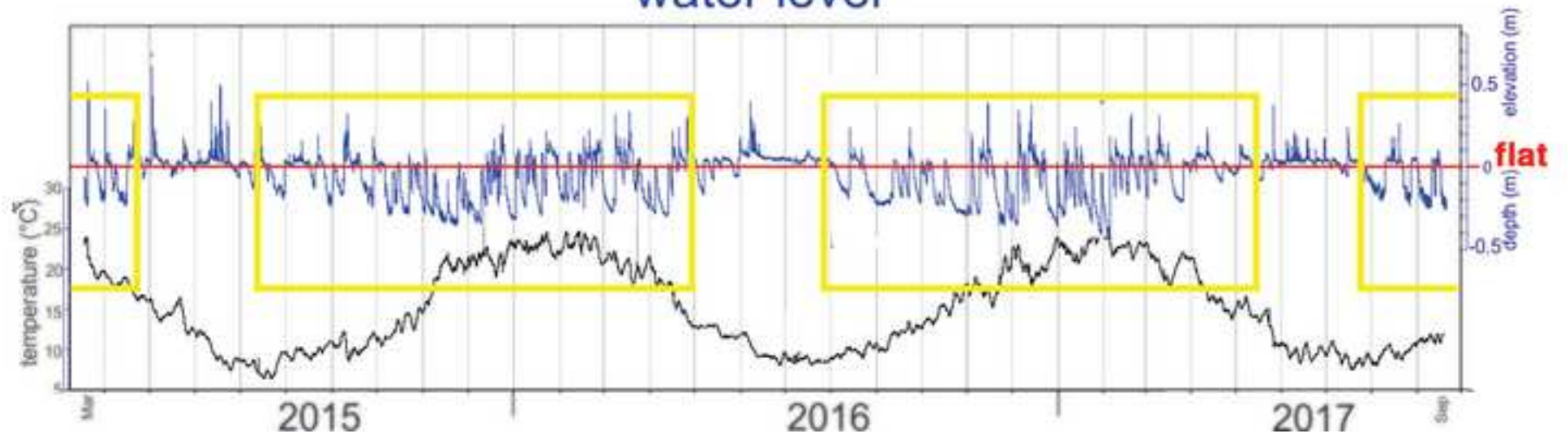


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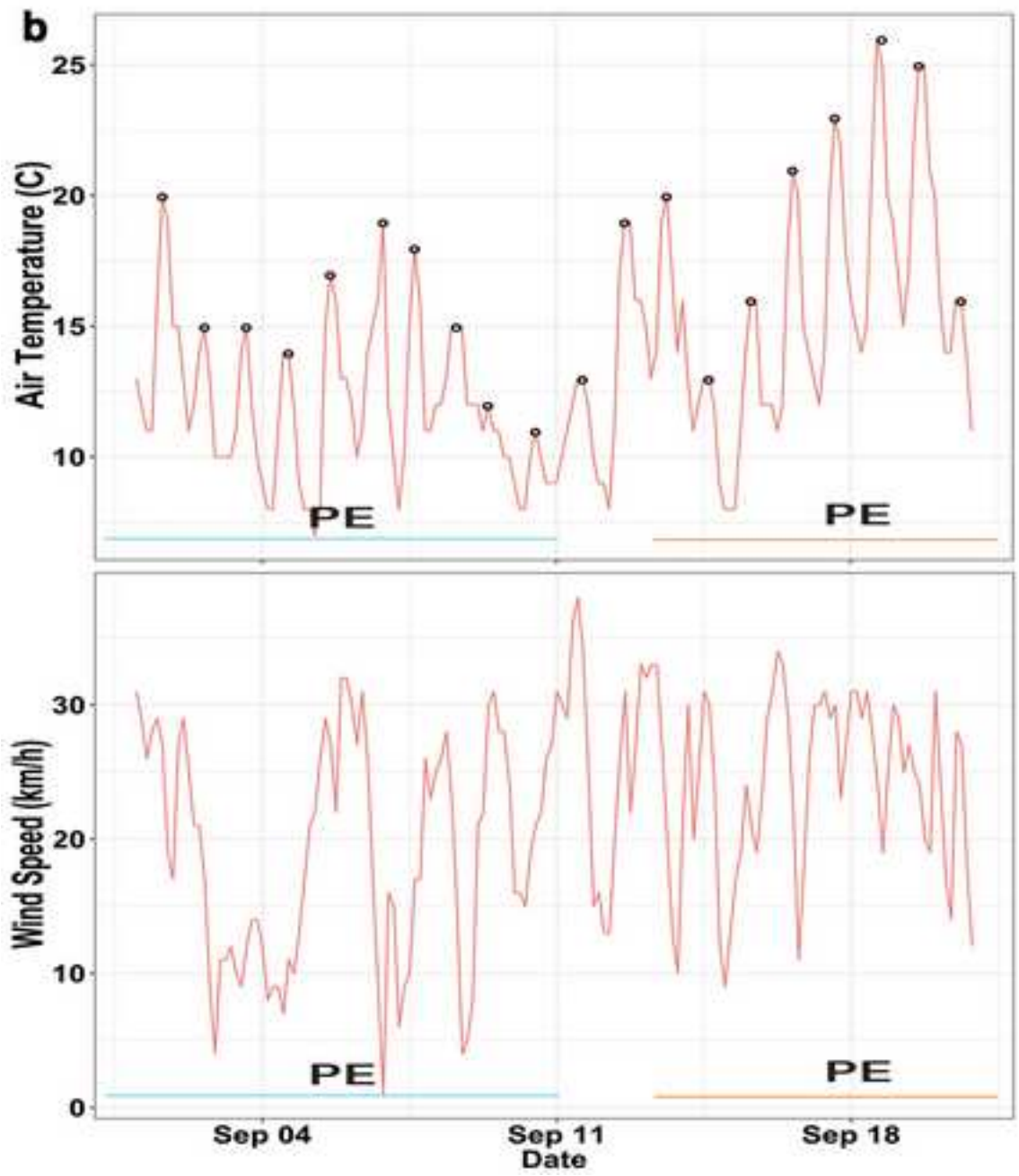
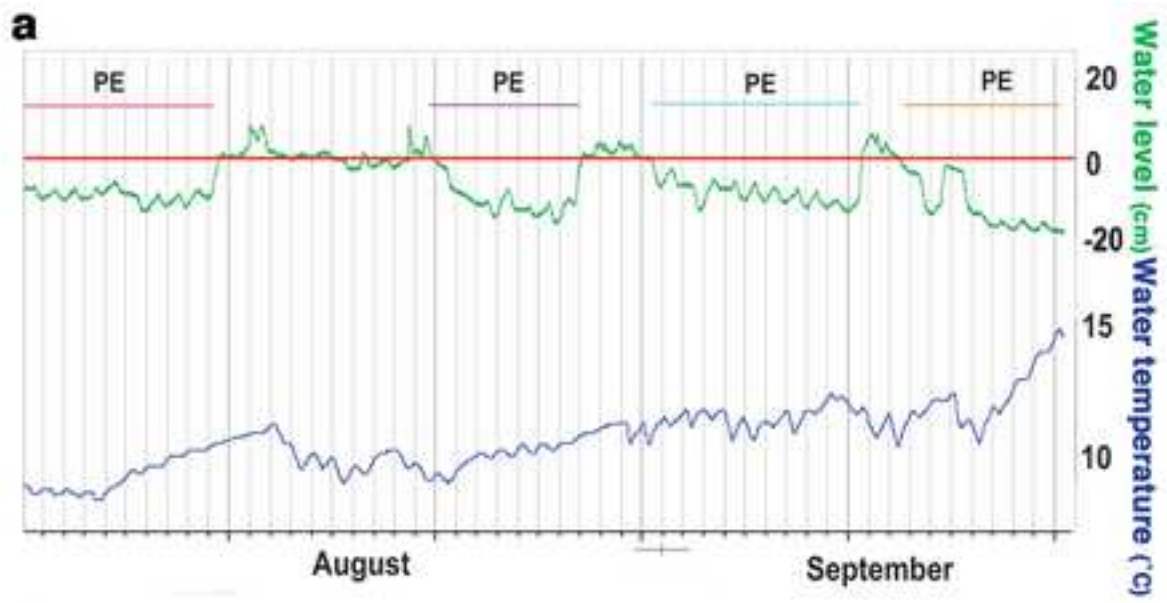
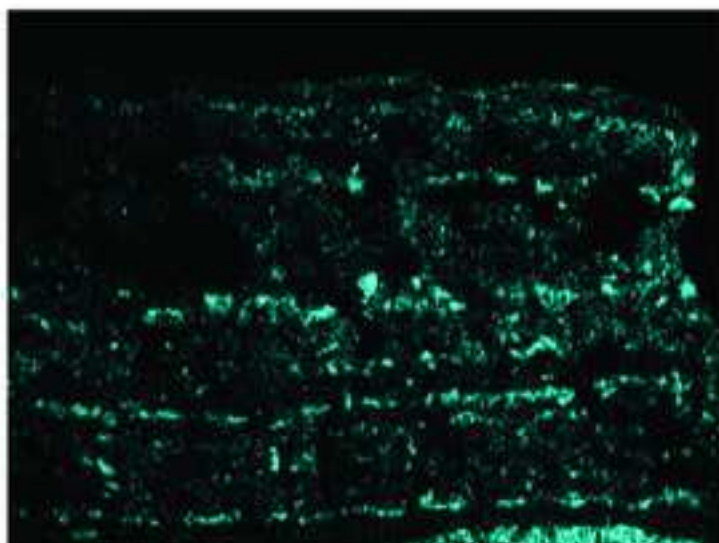


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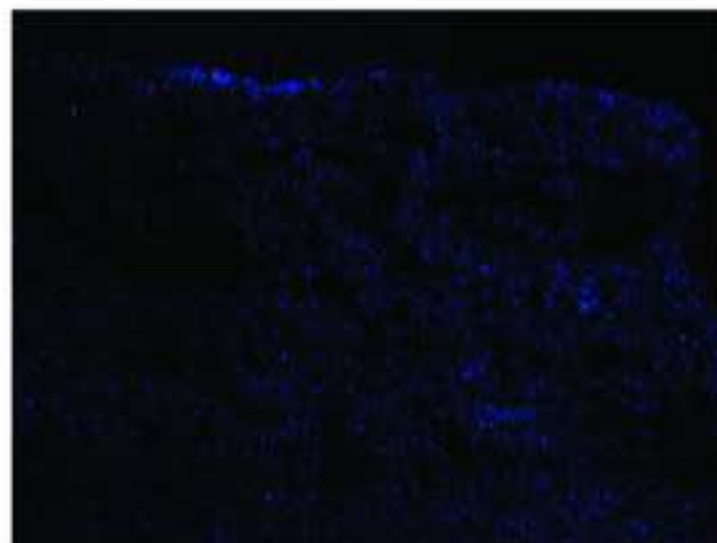
Figure 6  
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Ca K $\alpha$ 1



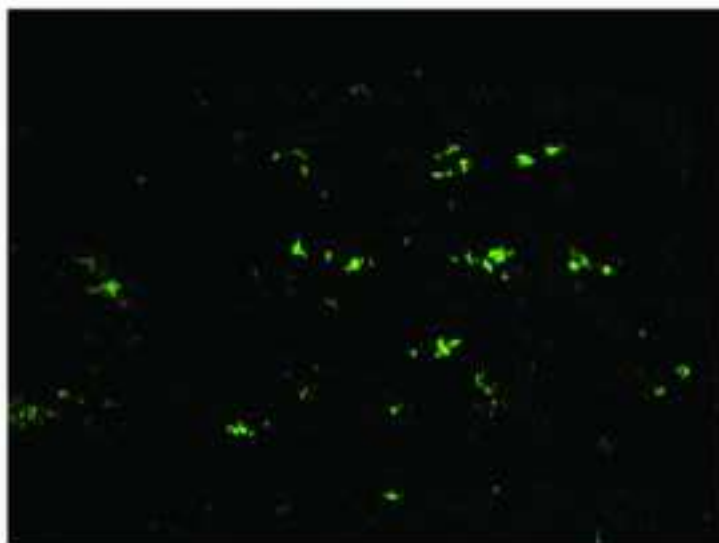
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S K $\alpha$ 1



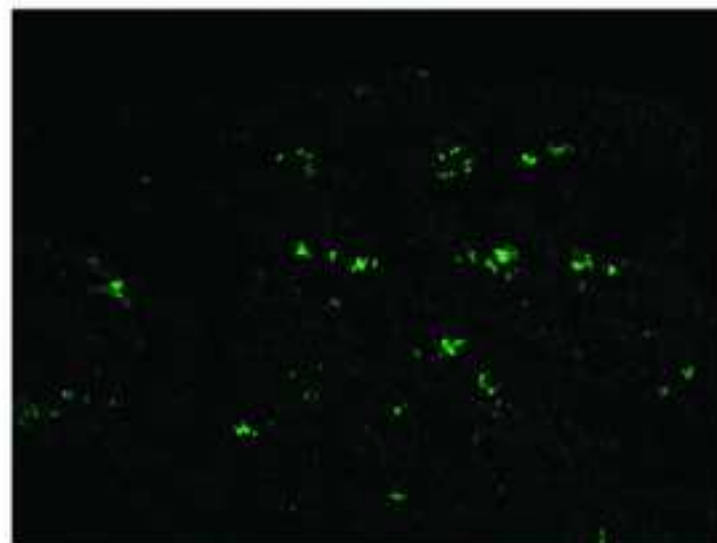
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Cl K $\alpha$ 1



1mm

Na K $\alpha$ 1\_2



1mm



Figure 7  
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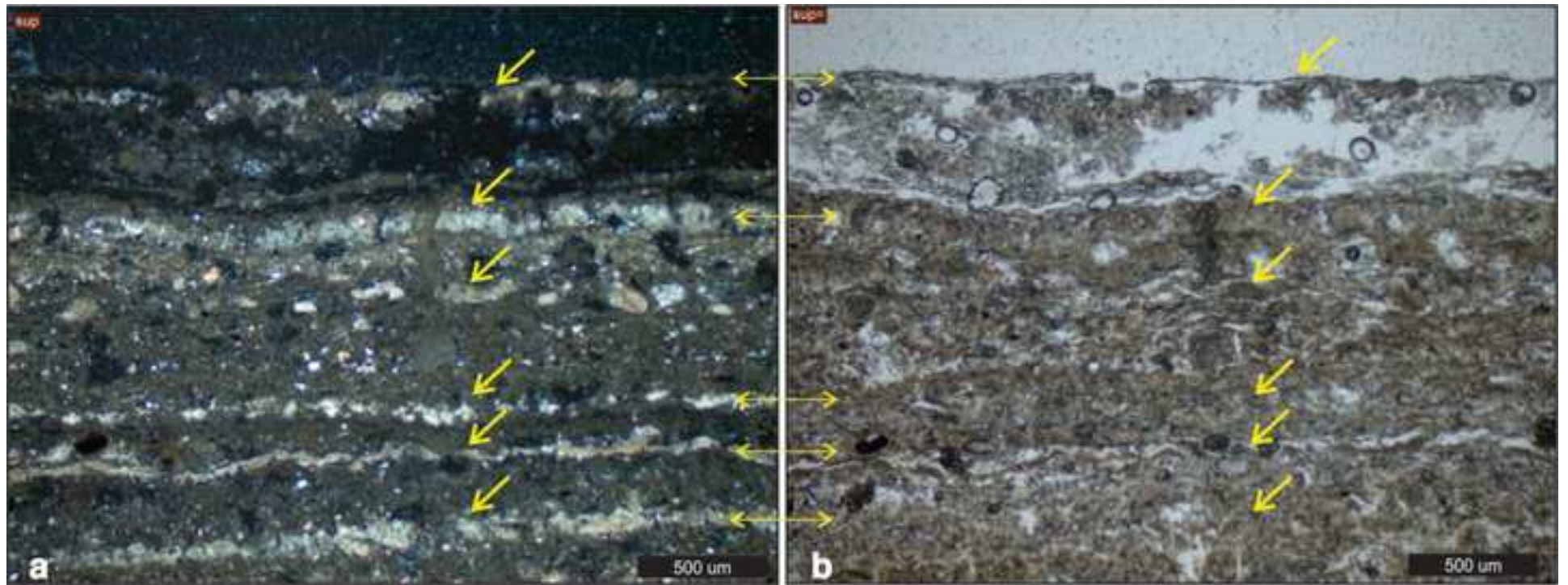


Figure 8  
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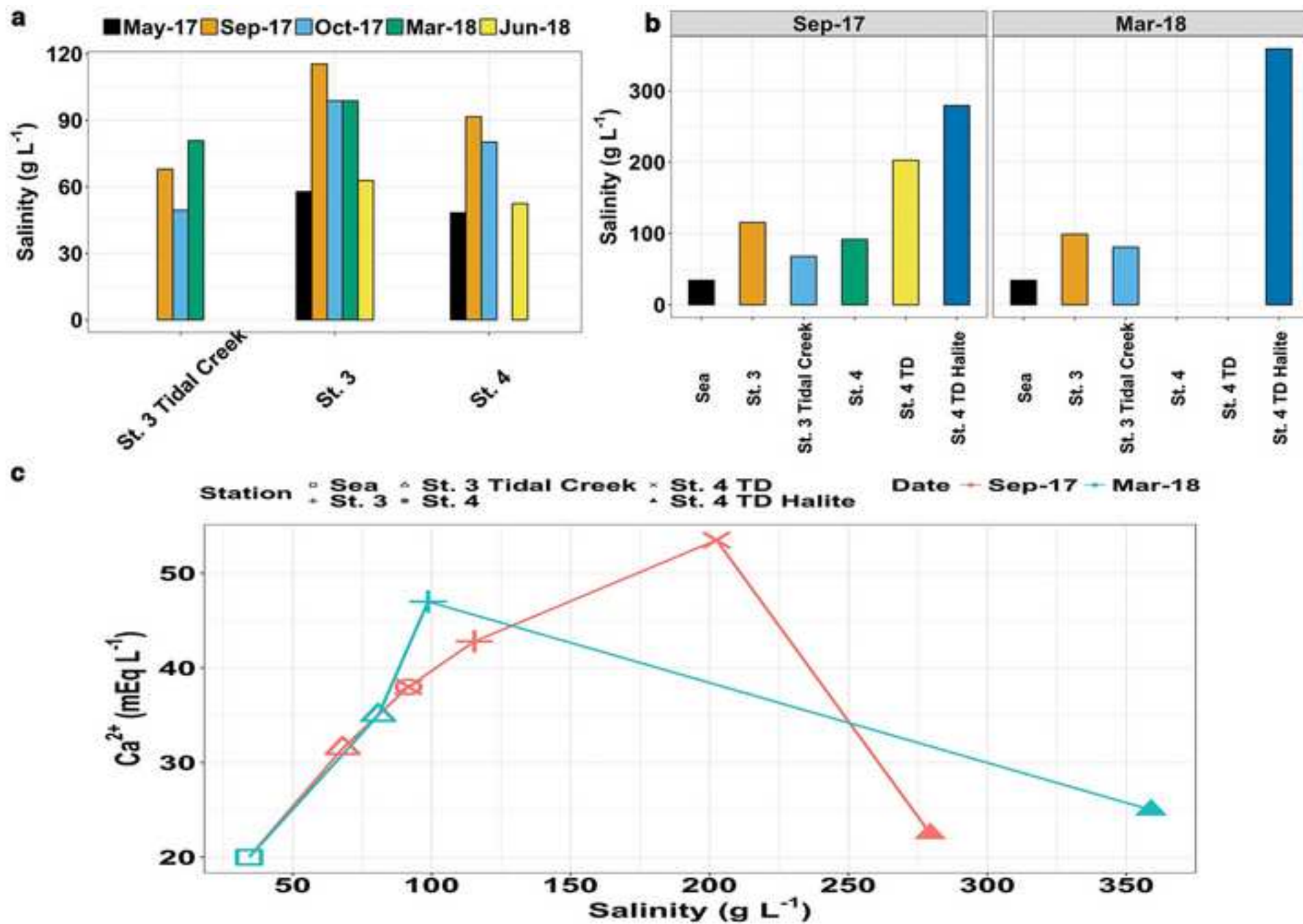


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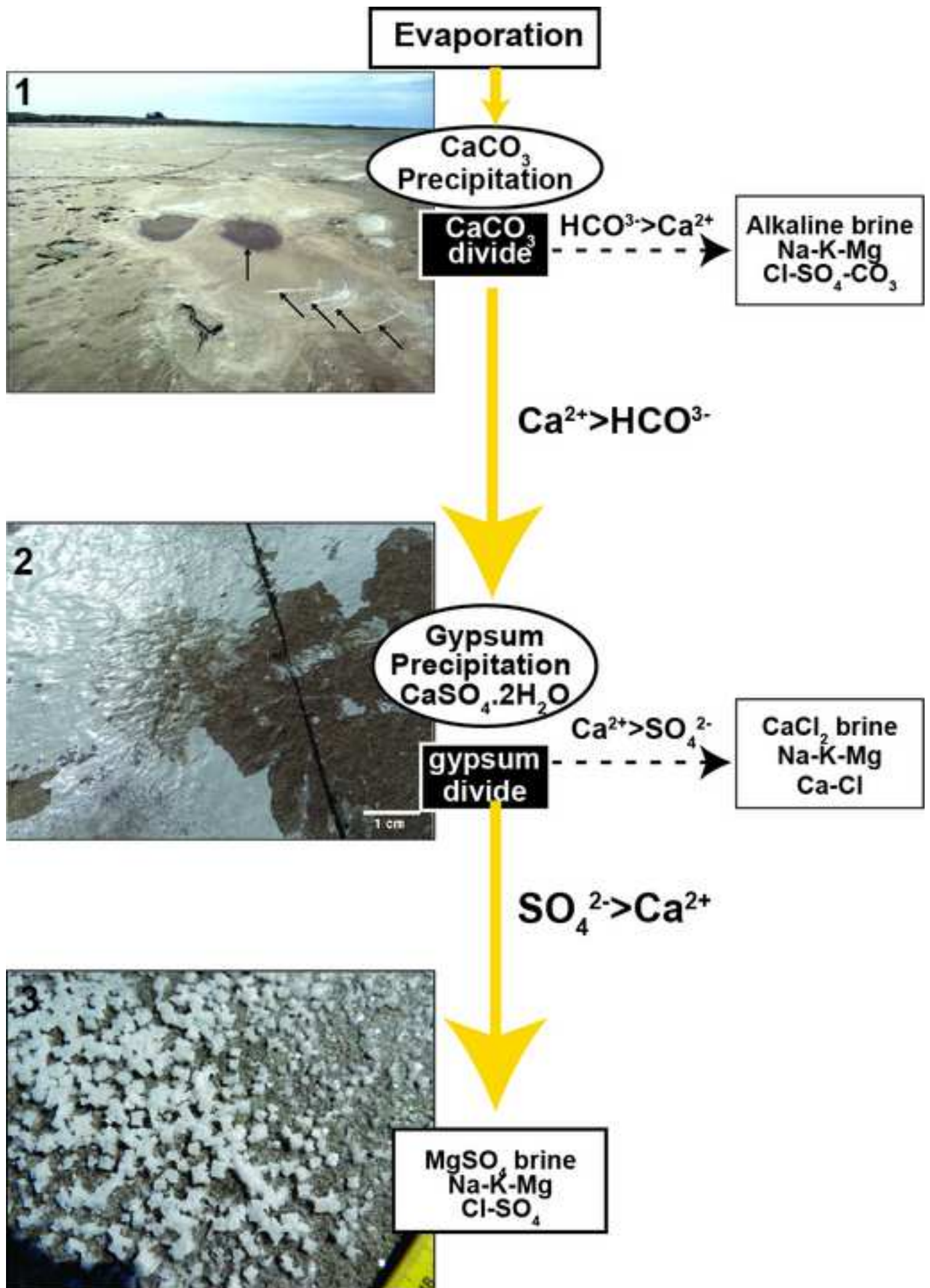
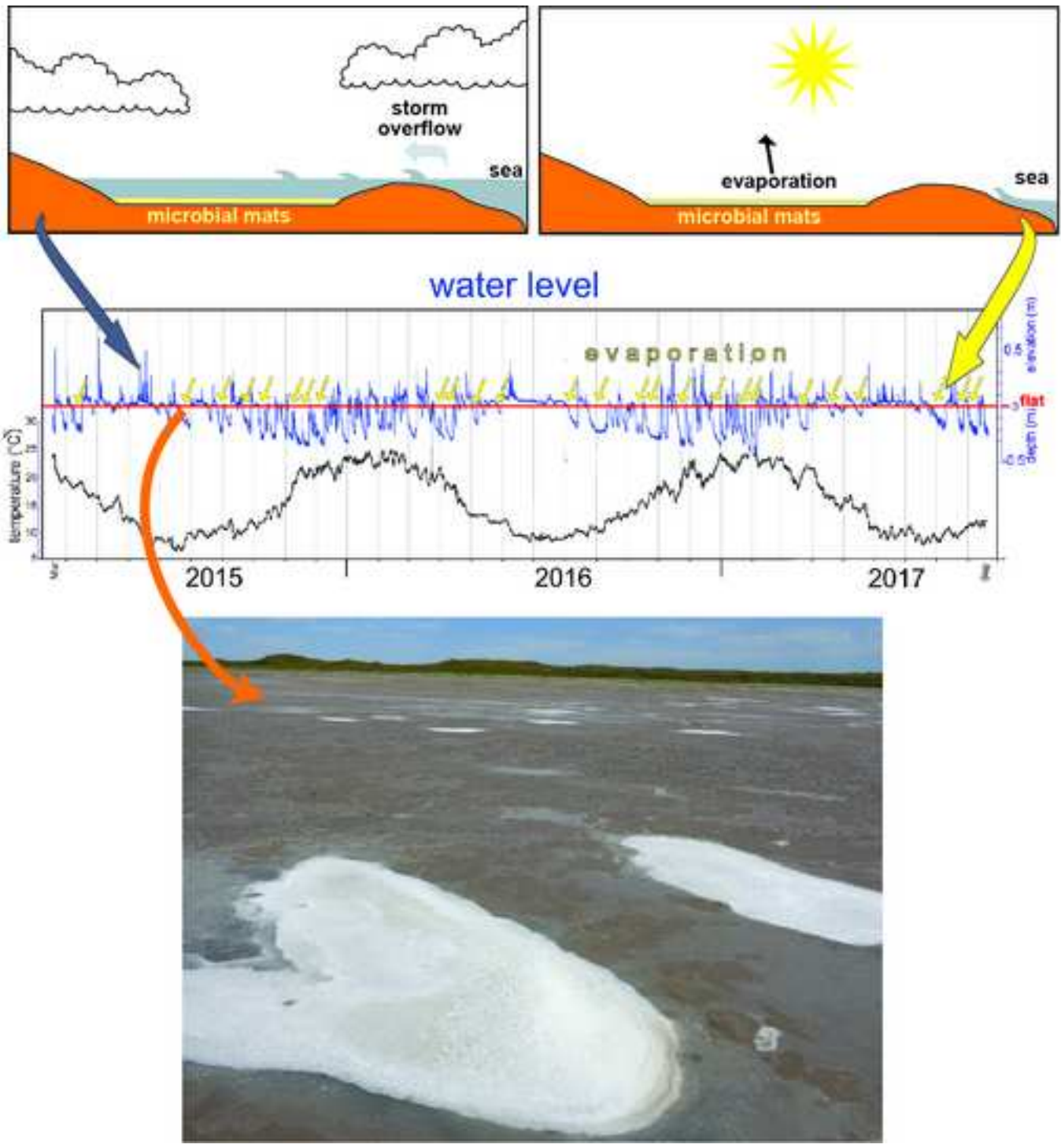


Figure 10  
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## **CRedit author statement**

**Vanesa L. Perillo:** Conceptualization; Data curation; Investigation; Methodology; Resources; Visualization; Writing - original draft; Writing - review & editing. **Lucía Maisano:** Conceptualization; Data curation; Investigation; Visualization; Resources; Writing - review & editing. **Ana M. Martinez:** Investigation; Resources; Writing - review & editing. **I. Emma Quijada:** Conceptualization; Investigation; Visualization; Writing - review & editing. **Diana G. Cuadrado:** Conceptualization; Funding acquisition; Investigation; Methodology; Project administration; Resources; Supervision; Visualization; Writing - review & editing.