

1 Simultaneous encapsulation of *trans*-resveratrol and vitamin D₃ in 2 highly concentrated double emulsions

3 Rocío DÍAZ-RUIZ^{a,b} (uo219049@uniovi.es), Irene VALDEÓN^a (UO237168@uniovi.es), José
4 Ramón ÁLVAREZ^a (jras@uniovi.es), María MATOS^{a,b} (matosmaria@uniovi.es) Gemma
5 GUTIÉRREZ^{1,a,b} (gutierrezgemma@uniovi.es),

6 ^aDepartment of Chemical and Environmental Engineering, University of Oviedo, Julián Clavería 8,
7 33006 Oviedo, Spain.

8 ^bInstituto Universitario de Biotecnología de Asturias, University of Oviedo, 33006, Spain

9 Tel: +34 985103509; Fax: +34 985103434

10 **Abstract**

11 *Background*

12 Biocompounds encapsulation is essential in order to protect them from environmental factors that
13 could enhance their oxidation or make them loss their beneficial properties due to extreme
14 photosensitivity, among others. The main goal of this work was to study the feasibility of preparing
15 concentrated double emulsions with a high loading capacity containing simultaneously *trans*-
16 *Resveratrol* (RSV) and vitamin D₃ (VitD₃). Such emulsions could be used for food fortification or
17 pharmaceutical formulations or as vehicles for targeted controlled release.

18 *Results*

19 In order to achieve large concentrations of the encapsulated compounds all the double emulsions
20 were formulated using a W₁/O in W₂ ratio of 80/20, while the ratios tested for W₁ in O where 20/80
21 and 30/70. All the emulsions were characterized by droplet size, morphology, colloidal stability and
22 encapsulation efficiency (EE) during six weeks. VitD₃ and RSV concentration were determined by a

¹Corresponding author.

E-mail address:

23 technique based on RP-HPLC (Reverse Phase High Performance Liquid Chromatography). The
24 viability of preparing concentrated $W_1/O/W_2$ emulsions containing both biocompounds has been
25 demonstrated with satisfactory results. Initial RSV concentrations in the concentrated double
26 emulsions formulated varied from 5.0 to 8.3 mg/L while for VitD₃ values of 28-32 mg/L were obtained.

27 The presence of Vit D₃ retarded RSV release in the formulated emulsions. It was observed that
28 during the first week of storage it varied considerably between 10-50%, depending on the W_1/O ratio
29 used. However, after two weeks of storage RSV release has been slightly retarded by 1-5 % by the
30 presence of Vit D₃.

31 *Conclusion*

32 Simultaneous encapsulation of RSV and VitD₃ was possible in high internal phase emulsions. The
33 emulsions presented high colloidal stability being suitable for food fortification applications.

34

35 **Keywords**

36 *Trans-Resveratrol*, Vitamin D₃, Simultaneously encapsulation, High concentrated double
37 emulsions, Biocompounds release, Double emulsion stability.

38 **1. Introduction**

39 A functional product consists on a food matrix enriched with any potential micronutrients, such as
40 vitamins or antioxidant compounds, which provides health benefits reducing risk disease. However,
41 frequently protection of these micronutrients is essential in order to avoid their oxidation or lost of
42 their beneficial properties. For these purpose, several colloidal systems, such as emulsions,
43 polymeric capsules or vesicles, had been used among others. However, in the food industry
44 emulsion-based systems are commonly used to encapsulate hydrophobic bioactive compounds due
45 to their relative ease of manufacturing using existing food processing operations^{1,2}.

46 Double emulsions are a potential vehicle for both molecules at the same time since $W/O/W$
47 emulsions present a hydrophobic oil compartment/layer between two hydrophilic aqueous phases.
48 Therefore, this colloidal delivery system would allow to encapsulate and protect both hydrophilic

49 RSV^{3,4} and VitD₃ simultaneously⁵⁻⁷. Emulsions could also produce a controlled release of the
50 encapsulated compounds. Moreover, the use of high concentrated emulsions, that is, emulsions with
51 a high proportion of the internal phase, provides greater stability allowing at the same time to contain
52 larger amount of entrapped active biocompound^{8,9}.

53 Resveratrol (RSV) is a polyphenol with a slightly hydrophilic¹⁰ character that has interesting and
54 beneficial health properties. It produces beneficial effects on human health due to its antioxidant
55 activity and anti-aging effect, which has greatly increased its applications in pharmaceutical or
56 cosmetics industries¹¹⁻¹⁵. RSV has a great number of potential health benefits against several
57 diseases such as cancer¹⁶, diabetes, neurodegeneration, cardiovascular disorders, inflammation,
58 and other age-related pathologies¹⁷⁻²². In previous studies, with in vitro and in vivo experiments, it
59 has been reported that RSV presents more evident effects when administrated in combination with
60 Vitamin D₃ (VitD₃) caused by a biphasic cooperative effect in ovarian cells²³. It was also stated that
61 supplementation of RSV and VitD₃ could reduce colon cancer risk²⁴. RSV has high instability (cis-
62 isomerization²⁵ and very photosensitive character²⁶) so its encapsulation is necessary before
63 administration²⁶⁻²⁸.

64 Vitamin D is a hydrophobic micronutrient that is required in the human diet to maintain good health
65 and well-being²⁹. Calcitriol (25-dihydroxy vitamin D₃) is the biological form of VitD₃, which plays a
66 critical role in osteoporosis prevention^{30,31}. VitD₃ deficiency often occurs in people who are not
67 exposed to sufficient sunlight and with metabolic or gastrointestinal disorders.

68 VitD₃ is highly sensitive to environmental factors, such as exposure to light, heat and oxygen. It can
69 therefore be easily oxidized, leading to loss of functionality and physiological benefits²⁹. In addition,
70 it is a molecule that generally has low water solubility and low oral bioavailability. For these reasons,
71 VitD₃ is often encapsulated within colloidal delivery systems, such as nanoparticles or microparticles
72 assembled from food grade biopolymers^{5,30} or lipids^{1,32}.

73 For these exposed reasons, it would be interesting to formulate a system that allows the
74 incorporation of both active biocompounds (RSV and VitD₃), while protecting them could also control
75 their release^{33,34}.

76 Highly internal phase concentration are referred to those emulsions which its external phase do not
77 overpass the critical Oswald number (26 % v/v), those emulsions describe a clear advantage versus
78 creaming instability respect more diluted emulsions since the high presence of droplets surrounding
79 each individual droplet neighbour creates hindrance droplets movements⁸. Hence internal fraction is
80 considered for the droplet movement models prediction³⁵. However, in the case of W/O/W double
81 emulsions two internal fractions are combined simultaneously. Taking into account the density
82 difference between phases, the presence of highly concentration of W/O droplets in the external
83 aqueous phase will retard droplets creaming, since the lower density of the oil respect to the water
84 is the main responsible for this phenomena, as it was reported in previous works³⁶. However, it is
85 also important to take into account that the inner water droplets will increase oil droplets viscosity
86 producing a retard on oil droplets movement.

87 In previous works RSV has been satisfactory encapsulated in high concentrated double emulsions³⁶
88 with high stability. Moreover, it has also been reported the effect of concentrating both the primary
89 W/O (increase of the content of water in the primary emulsion) and secondary W/O/W emulsions
90 (increase of the content of W/O droplets in the double emulsion) simultaneously in order to increase
91 the amount of RSV encapsulated using concentrated double water-in-oil-in-water emulsions
92 (W/O/W) optimizing the resulting colloidal stability and encapsulation efficiency (EE)³⁷.

93 Therefore, the aim of this work was to study the viability of encapsulating simultaneously RSV and
94 VitD₃ in concentrated double emulsions with high loading capacity, regarding the synergistic effect
95 of both encapsulated compounds on final emulsion properties. For this purpose, emulsions
96 containing individual and combined biocompounds were synthesized at two internal emulsion (W/O)
97 ratios: 20/80 and 30/70. All formulated emulsions were characterized in terms of droplet size
98 distribution, zeta potential, rheology, EE and colloidal stability.

99 **2. Materials and methods**

100 **2.1. Materials**

101 RSV (C₁₄H₁₂O₃), VitD₃, absolute ethanol and Tween 20 were purchased from Sigma–Aldrich (USA).
102 Miglyol® 812 (density 945 kg/m³ at 20°C), which is a neutral oil formed by esters of caprylic and
103 capric acids with glycerol, was supplied by Sasol GmbH (Germany). Polyglycerol Polycicineoleate

104 (PGPR, C₂₁H₄₂O₆) was supplied by Brenntag AG (Germany). Sodium chloride was purchased from
105 Panreac (Spain). Distilled water was used for the preparation of both aqueous phases. HPLC-grade
106 methanol, acetonitrile, 2-propanol, and acetic acid were purchased from Sigma Aldrich (USA).

107 **2.2. Methods**

108 Four types of emulsions were prepared: (i) blank (no encapsulated compounds), (ii) only RSV, (iii)
109 only VitD₃ (iv) both RSV and VitD₃. Emulsions containing separate compounds (types (ii) and (iii))
110 were prepared in order to study the effect produced by RSV and VitD₃ and the possible interactions
111 between both biocompounds.

112 For the preparation of the primary W₁/O emulsions two different internal ratios of W₁ in O (also
113 represented as W₁/O) were used being 20/80 and 30/70. All the concentrated double W₁/O/W₂
114 emulsions were formulated with the same external ratio of W₁/O in W₂ (also represented as W₁O/W₂)
115 of 80/20.

116 RSV was added to the internal aqueous phase W₁, in contrast, VitD₃ for its hydrophobic nature was
117 added to the oily phase O.

118 2.2.1. Water-in-oil (W₁/O) emulsion preparation

119 RSV is barely soluble in water, so an alcohol was added to water as solubilizing agent. The solubility
120 of RSV in alcohol decreases as the carbon number of the alcohol increases³⁸. Thus, a 20% ethanol
121 (v/v) solution was used as the dispersed phase containing 50 mg/l of RSV. 0.1 M NaCl was added
122 to the inner aqueous phase in all double emulsions to ensure W₁ droplet stability balancing the
123 osmotic pressure in the system^{39–41}.

124 Miglyol 812 was used as the oil phase containing the hydrophobic emulsifier (PGPR) previously
125 dissolved by magnetic stirring for 5 min. PGPR is commonly used in food formulation and has been
126 demonstrated to be highly effective at stabilizing W₁/O emulsions^{40,42,43}.

127 The amount of PGPR to be dissolved is very important for the preparation of the organic phase in
128 order to make the initial emulsion as stable as possible. Since two W₁/O ratios were used for the
129 preparation of the primary W₁/O emulsion, the surfactant/internal aqueous phase ratio was kept
130 constant, to ensure that the surface of the dispersed droplets was always covered by the stabilizer,

131 maintaining the amount of mass per unit area constant. The concentration used in previous
132 studies^{4,36} for a simple W_1/O emulsion ratio of 20/80 was 20% (w/w) of PGPR in oil phase dispersed
133 in the emulsion. However, previous experiments show considerable instability in maintaining the
134 same concentration of surfactant for higher internal phase proportions, so the proportion used was
135 40% (w/w) for simple W_1/O emulsions preparation using the ratio of 30/70³⁶.

136 In the case of the emulsions that contained VitD₃, the vitamin was added to Miglyol 812 to a
137 concentration of 50 mg/l.

138 Two W_1/O ratios were used 20/80 and 30/70. One hundred grams of emulsion were prepared for
139 each test. Both phases were emulsified in glass vessels by high shear mixing (SilentCruser M
140 Homogenizer, Heidolph, Germany) using a 6 mm dispersing tool at 2012 g for 5 min at room
141 temperature.

142 2.2.2. Water-in-oil-in-water ($W_1/O/W_2$) double emulsions preparation

143 Thirty grams of the $W_1/O/W_2$ double emulsions were prepared by dispersing the W_1/O primary
144 emulsion into the external aqueous phase (W_2) at a volumetric ratio of 80/20, with the intention of
145 obtaining more concentrated double emulsions taking into account articles previously published^{36,37}.

146 The external aqueous phase W_2 was formulated using a 2% (w/v) Tween 20 solution and 0.1 M NaCl
147 in order to match the osmotic pressure between W_1 and W_2 , in all emulsions.

148 Emulsification was carried out by mixing the continuous and dispersed phases with the before
149 mentioned Homogenizer at 224 g for 2 min.

150 The conditions of agitation are considerably milder than in the case of simple emulsion because
151 applying too much energy could break the initial emulsion and, thus, the internal aqueous phase
152 would migrate towards the external one, obtaining a simple O/W emulsion.

153 Emulsions were stored at controlled temperature of 20°C for six weeks for further characterization.

154 2.2.3. Emulsion characterization

155 2.2.3.1. Droplet size distribution

156 Emulsion droplet size distributions were obtained by the laser light scattering in a Mastersizer S long
157 bench apparatus (Malvern Instruments, Ltd., UK).

158 For single W_1/O emulsion measurements, the samples were dispersed in paraffin oil, whereas
159 $W_1/O/W_2$ double emulsion samples were diluted with deionized water. The refractive index used for
160 measuring W_1/O emulsions droplet size were 1.3300 for the aqueous internal phase and 1.4500 for
161 the oil external phase, while for measuring $W_1/O/W_2$ emulsions droplet size 1.4500 was used for the
162 W_1/O drops and 1.3300 for the aqueous external phase.

163 Three replicates were performed for each emulsion and the results were reported as droplet size
164 distribution in μm .

165 The size results are expressed in terms of equivalent spherical diameter (the diameter of a sphere
166 of the same volume as the measured particle) although for the emulsions the spherical shape can
167 be assumed.

168 2.2.3.2. Visual inspection

169 Micrographs of the emulsions were obtained with an Olympus BX50 light microscope (Olympus,
170 Japan) with 10–100 \times magnification using UV–vis and fluorescence lamps. Micrographs were used
171 for emulsions visual inspection and, with the proper scale, to confirm the droplet size obtained by
172 laser light scattering.

173 2.2.3.3. Colloidal stability

174 Emulsion stability at 30°C was analysed by measuring backscattering (BS) and transmission (TS)
175 profiles in a Turbiscan apparatus (Formulaction, France). Twenty ml were placed on the equipment
176 cells, transmitted and backscattered light was monitored as a function of time and cell height for six
177 weeks. The optical reading head scans the sample in the cell, providing TS and BS data every 40
178 μm in percentage relative to the standard (suspension of monodisperse spheres and silicone oil) as
179 a function of the sample height (in mm), being the total height occupied by the sample 40 mm. These
180 profiles provide useful information about changes in droplet size distribution, appearance of a
181 creaming layer or a clarification front with time^{44–46}.

182 The Turbiscan Stability Index (TSI) is the sum of all the variations detected in the samples in terms
183 of size and/or concentration, and is defined by the following equation, where H is the total height of
184 the cell at i interval time and is defined by equation 1.

$$TSI = \sum_i \frac{\sum_i |scan_i - scan_{i-1}|}{H} \quad \text{Equation 1}$$

185 2.2.3.4. Zeta potential

186 A ZetasizerNanoZS (Malvern Instruments Ltd., UK) was utilized for zeta potential (ζ) measurements
187 of the W₁/O/W₂ double emulsions. Two replicates were conducted for each sample at room
188 temperature.

189 2.2.3.5. Interfacial tension

190 Surface and Interfacial tension (γ) were measured at room temperature following the Du Noüy's
191 platinum ring method using a Sigma 700 tensiometer (KSV Instruments Ltd., Finland).

192 2.2.3.5. Rheology

193 The rheological tests were carried out with a MARS II rotational rheometer (Haake). All the analyses
194 were carried out at room temperature and a plate/plate measuring system (PP35) with a gap of 1
195 mm was employed. Samples rested for at least 5 min previous to any measurement, allowing the
196 stresses induced during sample load to relax. All measurements were replicated twice and the
197 measured data were processed by the HaakeRheowin 4.0 Software.

198 Steady-state flow measurements were carried out from 0.01 to 500 s⁻¹ Pa in 500 s at 25°C, shear
199 stress versus shear rate data were recorded.

200 Oscillatory measurements were carried out from 0.1 to 10 Hz at a constant shear stress of 1 Pa and
201 at 25°C. The storage modulus (G') and loss modulus (G'') were recorded versus frequency (rad/s).

202 2.2.4. Determination of the initial encapsulation efficiency (EE) by RP-HPLC analysis

203 The biocompounds (RSV and VitD₃) content in the external aqueous phases was determined by
204 chromatography (HP series 1100 chromatograph, Hewlett Packard, USA). The system was

205 equipped with a UV-vis absorbance detector HP G1315A or a fluorescence detector 1260 Infinity A
206 (Agilent Technologies, USA).

207 The separation was performed with a Zorbax Eclipse Plus C18 reversed phase column, with a
208 particle size of 5 μm and 4.6 mm \times 150 mm (Agilent Technologies, USA).

209 The mobile phase consisted of a mixture of (A) 100% milliQ-water and (B) 100% methanol with
210 gradient elution at a flow rate of 0.8 mL/min. The step gradient started with 80% mobile phase (A)
211 running 100% of mobile phase (B) in minute 5 for 10 min. The mobile phase (B) was run for 2 min
212 after each injection to prepare the column for the next run. Separation was carried out at room
213 temperature. A wavelength of 305 nm was used by the UV-vis detector while the fluorescence
214 detector was used at $\lambda_{\text{excitation}}/\lambda_{\text{emission}}$ of 310/410 nm. The column was cleaned after each analysis by
215 first running the mobile phase (A) for 20 min and a mobile phase (C) consisting of 50% acetonitrile,
216 25% milliQ-water, 25% 2-propanol, and 0.01% acid acetic for 40 min at a flow rate of 0.25 mL/min.
217 Finally, the column was rinsed with 50% of the mobile phase (A) and 50% of the mobile phase (B)
218 for another 20 min.

219 The emulsions external aqueous phases injected in the RP-HPLC were previously recovered by
220 centrifugation of the double emulsion at 8944 g for 10 min. The recovered samples was diluted with
221 methanol in volumetric ration 1:1 in order to increase RSV solubility in the phase, then the mixture
222 was filtrated with a 0.22 μm polyvinylidene difluoride syringe filter to remove all the oil phase that
223 could be still present.

224 The recovery yield (R_y) of RSV after the centrifugation and filtration stages was used to determine
225 the amount of biocompound lost during separation processes, since other authors reported on the
226 importance of taking into account to determine the EE values^{4,47}. For this purpose, a standard
227 emulsion, where all W_1 has migrated to W_2 (with a 0 % EE) was prepared. Therefore, an oil-in-water
228 (O/W₂) emulsion was prepared following the procedure previously described, and then W_1 phase
229 was added, simulating that all RSV transferred to W_2 . The theoretical RSV concentration for this
230 case is denominated the maximum concentration expected (C_0) in equations 2 and 3. The RSV
231 concentration measured in the separated aqueous phase from these simple emulsions prepared and

232 determined by RP-HPLC was denominated blanch concentration (C_{Blanch}). The recovery yield, R_y , of
233 the process was calculated by equation 2:

$$.R_y(\%) = \frac{C_{\text{blanch}}}{C_0} \cdot 100 \quad \text{Equation 2}$$

234 The encapsulation efficiency (EE) of these double emulsions was defined as the percentage of
235 encapsulated biocompound that remained in the W_1/O primary emulsion after the second
236 emulsification step. It was calculated by equation 3.

$$.EE(\%) = 100 - \frac{C_{\text{recovered}} \cdot 100}{C_0 \cdot R_y} \quad \text{Equation 3}$$

237 where $C_{\text{recovered}}$ is the RSV concentration measured in W_2 by RP-HPLC after centrifugation of each
238 individual double emulsion.

239 **3. Results and discussion**

240 The emulsions were characterized in terms of droplet size distribution, morphology, stability and
241 encapsulation efficiency. These parameters were evaluated weekly for six weeks.

242 **3.1. Droplet size distribution**

243 First of all, the droplet size distribution of the primary (W_1/O) emulsions was analysed (Figure 1).

244 It can be seen that there is a clear difference in the size distribution due to the different W_1/O ratios
245 used. The emulsion prepared using a 20/80 ratio had a narrower droplet size distribution, with a
246 single peak near the value of $3\mu\text{m}$, while the emulsion prepared using a 30/70 ratio showed wider
247 size distribution that goes from 1.5 to $13\mu\text{m}$, what could be to for the larger internal aqueous volume
248 what could produce less efficient agitation.

249 **Figure1.Droplet size distribution of W_1/O emulsions.**

250 The results of the droplet size distribution for the concentrated $W_1/O/W_2$ double emulsions prepared
251 using each type of the primary emulsions are shown in Figure 2A (internal W_1/O ratio of 20/80) and
252 Figure 2B (internal ratio of 30/70).

253 **Figure 2. Droplet size distribution of the double $W_1/O/W_2$ emulsions formulated with an**
254 **internal phase of W_1/O emulsion in fixed ratio 20/80 (A) or 30/70 (B) varying the**
255 **encapsulated biocompound.**

256 It has been clearly observed that the presence of an encapsulated biocompound did not affect the
257 droplet size of the final concentrated double emulsion. Zeta potential of all formulated emulsions with
258 and without encapsulated biocompound is presented in Table 1. It was observed that the presence
259 of biocompound does not affect the emulsion zeta potential and hence its stability, according to
260 electrostatic repulsion. For emulsions formulated with both internal ratios containing only RSV
261 encapsulated presented the lower zeta potential values. In all cases zeta potential had negative
262 values and lower than -30 mV what is an indication that the formulated emulsions presented
263 electrostatic stability despite the use of non-ionic surfactant in the formulations.

264 **Table 1. Zeta potential and Span values of formulated double emulsions**

265 However, there was an evident difference in their droplet size distributions regarding the different
266 internal phase ratios. When the 20/80 W_1/O internal phase ratio was used two clearly differentiated
267 peaks were observed in Figure 2A, one (between 0.1 and 1 μm) probably due to the formation of
268 excess surfactant agglomerates in the external aqueous phase W_2 and another due to the droplets
269 present in the double emulsion around 13 μm . This bimodal distribution has been observed in
270 previous studies where double $W_1/O/W_2$ emulsions were prepared using the same ratio for the primary
271 emulsion^{3,4,36,47}. However, when the concentrated double emulsions were prepared with the 30/70
272 W_1/O ratio, two peaks were also observed in Figure 2B but the proximity of the smaller peak to the
273 larger one, indicated that both of them correspond to a wider oil drops size distribution. However,
274 the larger peak has slightly lower main value (9 μm) than the one obtained with emulsions 20/80
275 W_1/O ratio. Span of all formulated emulsions is presented in Table 1, smaller values registered for
276 emulsions prepared with internal ratio of 30/70 corroborate the proximity of the two peaks observed
277 on the droplet size distributions of these emulsions. Span values varied from 1.5 to 1.8 for fresh
278 emulsions prepared using an internal ratio of 20/80 while values from 0.9 to 1.0 were obtained for
279 fresh emulsions formulated with an internal ratio of 30/70.

280 Figures 3 and 4 present the optical images of the fresh concentrated $W_1/O/W_2$ emulsions prepared
281 with the two W_1/O ratios used (20/80 and 30/70). Optical microscope corroborated the presence of
282 double emulsions. Both Figures showed very small water droplets (W_1) inside the bigger oil droplets
283 (O) dispersed in the external aqueous phase (W_2). It has also been observed that when the 30/70
284 internal emulsion ratio was used the double emulsions had smaller droplet size in all cases.

285 **Figure 3. Influence on morphology according to the type of emulsion prepared with 20/80**
286 **ratio of internal phase at 40x: (A) Double emulsion without encapsulated compound; (B)**
287 **Double emulsion containing RSV; (C) Double emulsion containing VitD₃; (D) Double**
288 **emulsion with RSV and VitD₃.**

289 **Figure 4. Influence on morphology according to the type of emulsion prepared with 30/70**
290 **ratio of internal phase at 40x: (A) Double emulsion without encapsulated compound; (B)**
291 **Double emulsion containing RSV, (C) Double emulsion containing VitD₃; (D) Double**
292 **emulsion with RSV and VitD₃.**

293 Values lower than 2 mN/m were registered for all interfacial tension measurements, indicating a low
294 tension and hence easy emulsified capacity for all systems. Moreover, surface tension
295 measurements of all systems were measured and any influence was observed by the presence of
296 biocompounds on them. Values between 40 and 42 were obtained for the internal aqueous phases
297 (W_1), while values between 29-30 mN/m were registered for the oil phases used being 35.5 mN/m
298 the value observed for the external aqueous phase (W_2). The fact that the presence of biocompound
299 do not affect surface tension of the different phases indicate that its presence did not affect the
300 emulsification properties of the systems.

301 Weekly evolution of the droplet size distribution for all the double emulsion formulated is shown in
302 **Figure 5.**

303 Size distribution of the concentrated double emulsions formulated without any compound
304 encapsulated did present significant changes during the 6 weeks studied for emulsions prepared
305 with an internal W_1/O ratio of 20/80 (Figure 5A). However, when the 30/70 ratio was used, slightly
306 wider size distributions with time were observed for the final $W_1/O/W_2$ emulsions, indicating faintly
307 coalescence (Figure 5B). Span values (Table 1) remained constant with time for all emulsions

308 formulated with an internal phase ratio of 20/80 while for emulsions prepared with an internal phase
309 ratio of 30/70 an increase of span value with time was observed for samples formulated with and
310 without biocompounds encapsulated, span values increase from 0.9-1.0 to 1.5-1.7, indicating the
311 presence of a wider droplet size distribution and hence instability.

312 **Figure 5. Influence of time (weekly evolution) on droplet size distribution of double**
313 **emulsions with internal phase of 20/80 ratio (left column) or 30/70 ratio (right column)**
314 **without encapsulated biocompound (A and B), with encapsulated RSV (C and D), with**
315 **encapsulated VitD₃ (E and F) and with encapsulated RSV and VitD₃ (G and H).**

316 No changes on droplet size distribution were appreciated when double emulsions were prepared
317 loaded with RSV and VitD₃ respectively using a 20/80 W₁/O internal ratio for primary emulsions
318 (Figure 5C and 5E). However, a 30/70 W₁/O ratio was used a substantial variation was also observed
319 (Figure 5D and 5F) regarding a wider size distribution indicating oil droplets coalesce.

320 For W₁/O/W₂ emulsions encapsulating both compounds simultaneously (Figure 5G and 5H) droplet
321 size distribution was appreciated to be constant when using a W₁/O internal ratio of 20/80 as was
322 the case when individual biocompounds were encapsulated. Surprisingly, for the double emulsion
323 prepared using 30/70 W₁/O ratio the droplet size distribution seemed to remain constant what was
324 not the case for individual biocompounds encapsulations.

325 Size distribution for double emulsions encapsulating only VitD₃ (Figure 5E and 5F) showed slight
326 variations with time for both W₁/O ratios studied. However, the main droplet size was not affected.

327 Figure 1S of Supplementary material present an optical image of all double emulsions formulated
328 after 6 weeks, for both internal phase ratios. There was no clear evidence of the influence of time on
329 morphology for all emulsion prepared. The images did not show significant changes in distribution
330 or droplet size after 6 weeks. However, it was evidence that the average oil droplet size was clearly
331 smaller for emulsions with a W₁/O ratio of 30/70 versus those of 20/80.

332 **3.2. Colloidal stability**

333 Stability was evaluated through the TSI value evolution during six weeks (Figure 6). The lower the
334 values of TSI the higher the emulsion stability, i.e. it reflects minor changes in the emulsion during
335 ageing⁴⁸.

336 **Figure 6. Evolution of TSI values during 45 days for double emulsions formulated with a**
337 **fixed volumetric ratios of W_1/O dispersed in W_2 (80/20) with different proportion of W_1**
338 **dispersed in O: 20/80 (A) or 30/70 (B) and different encapsulated biocompounds.**

339 Figure 6 showed that concentrated double emulsions formulated using a W_1/O ratio of 20/80 had a
340 greater stability, with TSI values varying from 5 to 10 after 6 weeks (Figure 6A), than those prepared
341 using a 30/70 ratio that showed variations from 40 to 55 after 6 weeks (Figure 6B). This indicates
342 that the W_1/O ratio used for the preparation of the primary emulsion had larger effect on emulsion
343 stability than the encapsulated biocompound. This effect was also observed when the span value
344 was evaluated, since an increase on span value with time was observed for emulsions formulated
345 with an internal emulsion ratio of 30/70 and it was not the case for emulsions prepared with 20/80.

346 The presence of biocompounds in the primary emulsions formulated with a 20/80 internal ratio gave
347 higher stability to the final concentrated double emulsions, up to 20 days. After this point, emulsions
348 with only VitD₃ encapsulated presented the lowest stability (larger TSI values)

349 The presence of biocompounds in the W_1/O emulsions formulated with a 30/70 internal ratio
350 increased also the stability of the double emulsions up to 20 days, although the TSI values were 10
351 times higher than those prepared with a 20/80 internal ratio, indicating significantly higher instability
352 for the ones formulated with 30/70 W_1/O ratio.

353 Measurement of the clarification height for all the formulated double emulsions was carried out using
354 Turbisoft software, results are presented in Figure 7.

355 **Figure 7. Clarification height observed during 45 days for samples with internal phase ratio**
356 **of 20/80 (A) and internal phase ratio of 30/70 (B)**

357 The clarification of all the concentrated double emulsions formulated using a 20/80 internal ratio is
358 negligible (lower than 1 mm). Only the very last days a sudden increase up to 1.2 mm of the
359 clarification height is observed for emulsion containing only VitD₃ encapsulated. For the $W_1/O/W_2$

360 emulsions prepared with a 30/70 internal ratio the clarification height varied with time stabilising at
361 9-10 mm after 20 days. This indicated that TSI values cannot be attributed to the migration of oil
362 droplets to the surface, hence any other destabilization phenomena takes place at the same time.

363 Droplets of the dispersed phase tend to migrate to the top of the cell, separating from the continuous
364 phase in which they are present. This process is known as creaming. Larger particles will tend to
365 migrate to the surface more quickly (according to Stoke's law). On the other hand, a higher density
366 difference between the droplets and the continuous phase would increase the droplets migration
367 velocity.

368 According to the theoretical behaviour described, double emulsions formulated with a 30/70 internal
369 ratio would be more stable against creaming than the ones prepared with 20/80. However,
370 experiments showed the opposite trend regarding a higher clarification front observed at the bottom
371 part of the cell for emulsions prepared with a W_1/O ratio of 30/70, which corroborates that the
372 observed creaming phenomenon is not only due to migration of the drops and there is another
373 instability phenomenon that is taking place simultaneously.

374 Backscattering profiles (Figure 2S and 3S of Supplementary Material) indicate droplet coalesce for
375 emulsions prepared with a 30/70 internal phase ratio, especially for those without biocompound
376 encapsulated (indicated by the backscattering difference observed at the medium part of the cell
377 along time). In addition, in the upper part of emulsions with internal phase ration of 30/70 a decrease
378 in backscattering along time, determined as a second clarification layer. This upper clarification layer
379 was assumed to be to the presence of a small layer of free oil as a consequence of oil drops
380 coalescence and migration.

381 Moreover, clarification height observed at the bottom part of the cell registered for 30/70 internal
382 phase ratio emulsions indicated that part of inner water phase (W_1) has migrated to the external
383 water phase (W_2), since the bottom clarification height was larger than expected if just all W_2 will be
384 at the bottom of the cell. The formation of this water layer at the bottom of the cell was reported in
385 previous works where the stability of W_1/O emulsions was studied using different types of emulsifiers
386 at 5% (w/v) for the formulation of double $W_1/O/W_2$ emulsions containing RSV⁴.

387 On the other hand, double emulsions formulated with a 20/80 W_1/O ratio did not present any
388 significant coalescence phenomenon, since backscattering profiles did not present significant
389 changes on the middle part of the cell and the thickness of the top clarification layer is nearly
390 negligible and any free oil was observed at naked eye.

391 Recently the swelling and deswelling of inner droplets of double emulsions was modelled by other
392 authors by the use of osmotic pressure⁴⁹⁻⁵¹. The use of different salinity in both aqueous phases of
393 double emulsions verified the effect of the swelling and deswelling phenomena in order to equal
394 salinities of both phases⁴⁹. However, in the present study deswelling will be expected when a
395 biocompound is encapsulated while for emulsions with non biocompound encapsulated any swelling
396 or deswelling behaviour will be estimated since salinity was the same in both aqueous phases.

397 Moreover, the same authors validated a model to evaluate inner droplets deswelling rate (W_1 migrate
398 to W_2), using two main forces: Separation force governed by the interfacial tension between phases
399 and resistance force governed by oil viscosity.

400 These two models indicate that for big oil drops the effect of the W_1 drop size had a minor importance.
401 However, for small oil droplets (a few microns) the effect of inner water drop diameter had a major
402 influence. The W_1 deswelling rates sharply increases in double emulsions with small oil droplets, with
403 large W_1 droplets⁴⁹.

404 According to the mentioned models, it can be assumed that in the present study, the W_1 droplets
405 from emulsions prepared with an internal ratio of 30/70 will leave easily oil drops to escape into the
406 external W_2 phase than those emulsions formulated with a W_1/O ratio of 20/80 and this phenomena
407 can be attributed to the highest bottom clarification layer observed on emulsions prepared with
408 internal phase ratio of 30/70 respect the ones with 20/80. Moreover, it is important to point out that
409 the similar droplet size obtained between W_1/O emulsions (range between 1.5 and 13 μm) and O/W_2
410 emulsions (range between 2-22 μm) when an internal ratio of 30/70 was used what could enhance
411 W_1 escape from oil droplets

412 3.3 Rheology behavior

413 Emulsions were characterized in terms of rheology. Curves flows for emulsions with and without
414 encapsulated biocompounds with W_1/O ratio of 20/80 and 30/70 are presented in Figure 8. It was
415 observed that all emulsions presented a pseudoplastic behavior. Emulsions with W_1/O ratio of 20/80
416 presented lower viscosity than emulsions with W_1/O ratio of 30/70 in all cases. The presence of the
417 biocompounds had minor influence on emulsion viscosity, without a clear trend observed for
418 emulsions with W_1/O ratio of 20/80, but regarding a decrease for emulsions with W_1/O ratio of 30/70.

419 **Figure 8. Flow curves of emulsion with W_1/O ratio of 20/80 (A) and 30/70 (B)**

420 Storage modulus or elastic modulus (G') and loss or viscous modulus (G'') were recorded for a
421 frequency range of 0-63 rad/s, for emulsions with W_1/O ratio of 20/80 and 30/70 and presented in
422 Figure 9. For the range studied all emulsions showed larger elastic than viscous behaviour. Similar
423 viscous modulus values were observed for all emulsions formulated but large difference was
424 observed regarding elastic modulus. Emulsions with W_1/O ratio of 20/80 presented values more than
425 twice the registered for emulsions with W_1/O ratio of 30/70. Moreover, for both type of emulsions the
426 presence of biocompound increased viscous modulus values, but it was not found a clear trend
427 between regarding the effect produced by the biocompound encapsulated.

428 **Figure 9. Elastic modulus (G') and viscous modulus (G'') versus frequency for double**
429 **emulsions $W_1/O/W_2$ formulated with an internal phase of W_1/O of 20/80 (A) and 30/70 (B)**

430 **3.4 Encapsulation efficiency (EE)**

431 The double emulsions that contained VitD₃ did not show any evidence of the compound recovery,
432 implying EE values of 100%. This was the expected result considering the low solubility of VitD₃ in
433 water⁵².

434 The EE values obtained of all the concentrated double emulsions formulated were calculated and
435 are shown in Figure 10 (A) and the RSV encapsulated concentration on the emulsions with time is
436 presented in Figure 10 (B).

437 **Figure 10. (A) Encapsulation efficiency and (B) Resveratrol loaded concentration for**
438 **emulsions with an internal phase concentration W_1/O of 20/80 or 30/70.**

439 RSV EE decreased with time in all cases. Initial RSV EE varies from 62.9 to 69.2 % without a clear
440 influence of the internal phase ratio used (20/80 or 30/70). Moreover, the presence of VitD₃ does not
441 offer a clear influence on the initial RSV EE of the formulated emulsions. However, the encapsulation
442 of VitD₃ presents an advantage on the RSV release with time. RSV release was retarded when VitD₃
443 was simultaneously encapsulated in double emulsions, especially for the case of the emulsions
444 formulated with W_1/O ratio of 20/80 for the six weeks studied. Significant retarded release was
445 observed after one week of storage, from them on differences on RSV EE between double emulsions
446 with only RSV and double emulsions with RSV and VitD₃ were less significant. After one week of
447 storage, RSV encapsulated increased from 4.0 to 4.4 mg RSV/L for emulsions formulated with a
448 W_1/O ratio of 20/80, while an increase from 4.6 to 8.7 mg/L was obtained for the emulsions
449 formulated with W_1/O ratio of 30/70 when VitD₃ and RSV were encapsulated simultaneously respect
450 when just RSV was used. From them on, the presence of VitD₃ on the emulsion increases RSV
451 emulsion loading capacity, the differences observed are lower (between 0.3-2.0 mg/L)

452 Despite the similar initial EE values obtained for all formulations, the emulsions formulated with a
453 W_1/O ratio of 30/70 presented higher loading capacity (Figure 10B) due to the higher amount of W_1 ,
454 with values in the range 7.5-8.4 mg/L, while for emulsion with W_1/O ratio of 20/70 the loading capacity
455 obtained varied from 5.0 to 5.5 mg/L, being in good agreement with concentrations obtained in
456 previous studies^{36,37}. After six weeks of storage emulsions presented a loading capacity of 2.6-3.2
457 mg/L for emulsions with W_1/O ratio of 20/80 and 4.1-4.6 mg/L for emulsions formulated with a W_1/O
458 ratio of 30/70.

459 There was no release of VitD₃ observed during all the period studied as expected due to its high
460 hydrophobicity. As consequence, the concentration was constant with values of 28 and 32 mg/L
461 VitD₃ for emulsions formulated with a W_1/O ratio of 20/80 and 30/70, respectively.

462 **4. Conclusions**

463 The viability of preparing concentrated $W_1/O/W_2$ double emulsions containing either RSV or VitD₃
464 and both simultaneously with an external W_1/O in W_2 ratio of 80/20 has been demonstrated when
465 using primary W_1/O emulsions formulated with internal ratios of W_1 in O of 20/80 and 30/70.

466 The colloidal stability of the formulated double emulsions was higher for those prepared using an
467 internal W_1/O ratio of concentration 20/80 than for those with primary emulsion W_1/O 30/70.

468 Initial RSV EE of the concentrated double emulsions increases with increasing the internal W_1/O
469 ratio. However, these emulsions with higher internal phase showed higher instability and therefore
470 led to lower RSV EE values after one week. The initial RSV EE values were between 62.9 and 69.2%
471 and it decreased asymptotically down to 32-40%. Initial RSV concentrations in the concentrated
472 double emulsions formulated varied from 4.6 to 8.2 mg/L while for VitD₃ values of 32 and 28 mg/L
473 were obtained.

474 The presence of VitD₃ in emulsions do not significantly increased the initial RSV EE values but offers
475 a clear advantage on the retarded RSV release, especially from one to three weeks of storage.
476 Moreover, according to the stability results of the concentrated double emulsions formulated, it was
477 evidenced that the presence of VitD₃ in the oily phase could significantly improve the colloidal stability
478 in comparison to those emulsions that only contained RSV. Therefore, the $W_1/O/W_2$ emulsions
479 formulated in this work could be suitable to be incorporated into food or cosmetic products in order
480 to supplement in a control manner a considerable amount of both RSV and VitD₃ increasing the
481 health benefits on the human body.

482 **5. Acknowledgments**

483 This work was supported by the Consejería de Economía y Empleo del Principado de Asturias [Grant
484 IDI/2018/000185, Plan de Ciencia, Tecnología e Innovación 2013-2017], and the Ministry of Science,
485 Education and Universities [grant MAT2017-84959-C2-1-R].

486 **6. References**

487 1 Ozturk B, Argin S, Ozilgen M, and McClements D. Formation and Stabilization of
488 Nanoemulsion-Based Vitamin E Delivery Systems using Natural Biopolymers: Whey Protein
489 Isolate and Gum Arabic. *Food Chem* **188**:256–263 (2015).

- 490 2 Park S, Mun S, and Kim YR. Effect of xanthan gum on lipid digestion and bioaccessibility of
491 β -carotene-loaded rice starch-based filled hydrogels. *Food Res Int Elsevier*; **105**:440–445
492 (2018).
- 493 3 Hemar Y, Cheng LJ, Oliver CM, Sanguansri L, and Augustin M. Encapsulation of resveratrol
494 using Water-in-Oil-in-Water double emulsions. *Food Biophys* **5**:120–127 (2010).
- 495 4 Matos M, Gutiérrez G, Coca J, and Pazos C. Preparation of water-in-oil-in-water (W1/O/W2)
496 double emulsions containing trans-resveratrol. *Colloids Surfaces A Physicochem Eng Asp*
497 Elsevier B.V.; **442**:69–79 (2014).
- 498 5 Teng Z, Luo Y, and Wang Q. Carboxymethyl chitosan – soy protein complex nanoparticles
499 for the encapsulation and controlled release of vitamin D3. *Food Chem Elsevier Ltd*;
500 **141**:524–532 (2013).
- 501 6 Gonnet M, Lethuaut L, and Boury F. New trends in encapsulation of liposoluble vitamins. *J*
502 *Control Release Elsevier B.V.*; **146**:276–290 (2010).
- 503 7 Winuprasith T, Khomein P, and Mitbumrung W. Encapsulation of vitamin D3 in pickering
504 emulsions stabilized by nanofibrillated mangosteen cellulose: Impact on in vitro digestion
505 and bioaccessibility. *Food Hydrocoll Elsevier Ltd*; **83**:153–164 (2018).
- 506 8 Gutiérrez G, Matos M, Benito JM, Coca J, and Pazos C. Preparation of HIPEs with
507 controlled droplet size containing lutein. *Colloids Surfaces A Physicochem Eng Asp Elsevier*
508 B.V.; **442**:111–122 (2014).
- 509 9 Matos M, Gutiérrez G, Iglesias O, Coca J, and Pazos C. Characterization, stability and
510 rheology of highly concentrated monodisperse emulsions containing lutein. *Food Hydrocoll*
511 **49**:156–163 (2015).
- 512 10 Atanackovi MT, Gojkovi LC, and Cveji JM. Improving the low solubility of resveratrol. *BMC*
513 *Pharmacol Toxicol* **13**:6511 (2012).
- 514 11 Vang O, Ahmad N, Baile CA, Baur JA, Brown K, Csiszar A, Das DK, Delmas D, Gottfried C,
515 Lin HY, Ma QY, Mukhopadhyay P, Nalin N, Pezzuto JM, Richard T, Shukla Y, Surh YJ,
516 Szekeres T, Szkudelski T, Walle T, and Wu JM. What is new for an old molecule?

- 517 Systematic review and recommendations on the use of resveratrol. *PLoS One* **6** (2011).
- 518 12 Yang T, Wang L, Zhu M, Zhang L, and Yan L. Properties and molecular mechanisms of
519 resveratrol: a review. *Die Pharm - An Int J Pharm Sci* **70**:501–506 (2015).
- 520 13 Betz J. Resveratrol and its Effects on Human Health and Longevity - Myth or Miracle? Truth
521 Publ. Int. Taichung, Taiwan; 2011.
- 522 14 Baxter RA. Anti-aging properties of resveratrol: Review and report of a potent new
523 antioxidant skin care formulation. *J Cosmet Dermatol* **7**:2–7 (2008).
- 524 15 Serravallo M, Jagdeo J, Glick SA, Siegel DM, and Brody NI. Sirtuins in dermatology:
525 Applications for future research and therapeutics. *Arch Dermatol Res* **305**:269–282 (2013).
- 526 16 Aluyen JK, Ton QN, Tran T, Yang AE, Gottlieb HB, and Bellanger RA. Resveratrol: Potential
527 as Anticancer Agent Resveratrol. *J Diet Suppl* **9**:45–46 (2012).
- 528 17 Diaz-Gerevini GT, Repossì G, Dain A, Tarres MC, Das UN, and Eynard AR. Beneficial
529 action of resveratrol: How and why? *Nutrition* **32**:174–178 (2016).
- 530 18 Murtaza G, Latif U, Najam-UI-Haq M, Sajjad A, Karim S, Akhtar M, and Hussain I.
531 Resveratrol: An active natural compound in red wines for health. *J Food Drug Anal* **21**
532 (2013).
- 533 19 Novelle MG, Wahl D, Diéguez C, Bernier M, and Cabo R De. Resveratrol supplementation:
534 Where are we now and where should we go? *Ageing Res Rev Elsevier B.V.*; **21**:1–15
535 (2015).
- 536 20 Saiko P, Szakmary A, Jaeger W, and Szekeres T. Resveratrol and its analogs: Defense
537 against cancer, coronary disease and neurodegenerative maladies or just a fad? *Mutat Res*
538 - *Rev Mutat Res* **658**:68–94 (2008).
- 539 21 Yang X, Li X, and Ren J. From French Paradox to Cancer Treatment: Anti-cancer Activities
540 and Mechanisms of Resveratrol. *Anticancer Agents Med Chem* **14**:806–825 (2014).
- 541 22 Ruivo J, Francisco C, Oliveira R, and Figueiras A. The main potentialities of resveratrol for
542 drug delivery systems. *Brazilian J Pharm Sci* **51**:499–514 (2015).

- 543 23 Uberti F, Morsanuto V, Aprile S, Ghirlanda S, Stoppa I, Cochis A, Grosa G, Rimondini L,
544 and Molinari C. Biological effects of combined resveratrol and vitamin D3 on ovarian tissue.
545 *J Ovarian Res Journal of Ovarian Research*; **10**:1–14 (2017).
- 546 24 Pelt V, Msiv CVP, Sabir MS, Galligan M, Jacobs ET, Whitfield GK, Haussler MR, and
547 Jurutka PW. The Interaction of β -catenin , Vitamin D , Resveratrol , and Two Common VDR
548 Polymorphic Variants in Colorectal Carcinogenesis. The University of Arizona; 2016.
- 549 25 Gambini J, López-grueso R, Olaso-gonzález G, Inglés M, Abdelazid K, El M, Bonet-costa V,
550 Borrás C, and Vi J. Resveratrol: distribución, propiedades y perspectivas. *Rev Esp Geriatr*
551 *Gerontol* **48**:79–88 (2013).
- 552 26 Silva CG, Monteiro J, Marques RRN, Silva AMT, Martínez C, Canle L. M, and Faria JL.
553 Photochemical and photocatalytic degradation of trans-resveratrol. *Photochem Photobiol Sci*
554 **12**:638–644 (2013).
- 555 27 Vian M, Tomao V, Gallet S, Coulomb P, and Lacombe J. Simple and rapid method for cis-
556 and trans-resveratrol and piceid isomers determination in wine by high-performance liquid
557 chromatography using Chromolith columns. *J Chromatogr A* **1085**:224–229 (2005).
- 558 28 Soo E, Thakur S, Qu Z, Jambhrunkar S, Parekh HS, and Popat A. Enhancing delivery and
559 cytotoxicity of resveratrol through a dual nanoencapsulation approach. *J Colloid Interface*
560 *Sci Elsevier Inc.*; **462**:368–374 (2016).
- 561 29 Luo Y, Teng Z, and Wang Q. Development of zein nanoparticles coated with carboxymethyl
562 chitosan for encapsulation and controlled release of vitamin D3. *J Agric Food Chem* **60**:836–
563 843 (2012).
- 564 30 Lin Y, Wang YH, Yang XQ, Guo J, and Wang JM. Corn protein hydrolysate as a novel nano-
565 vehicle: Enhanced physicochemical stability and in vitro bioaccessibility of vitamin D3. *LWT*
566 *- Food Sci Technol Elsevier Ltd*; **72**:510–517 (2016).
- 567 31 Liu K, Kong XL, Li QM, Zhang HL, Zha XQ, and Luo JP. Stability and bioavailability of
568 vitamin D3 encapsulated in composite gels of whey protein isolate and lotus root
569 amylopectin. *Carbohydr Polym Elsevier*; **227**:115337 (2020).

- 570 32 Ziani K, Fang Y, and McClements D. Encapsulation of functional lipophilic components in
571 surfactant-based colloidal delivery systems: Vitamin E, vitamin D, and lemon oil. *Food Chem*
572 **134**:1106–1112 (2012).
- 573 33 Mitbumrung W, Supphantharika M, McClements DJ, and Winuprasith T. Encapsulation of
574 Vitamin D3 in Pickering Emulsion Stabilized by Nanofibrillated Mangosteen Cellulose: Effect
575 of Environmental Stresses. *J Food Sci* **84**:3213–3221 (2019).
- 576 34 Davidov-Pardo G and McClements DJ. Resveratrol encapsulation: Designing delivery
577 systems to overcome solubility, stability and bioavailability issues. *Trends Food Sci Technol*
578 Elsevier Ltd; **38**:88–103 (2014).
- 579 35 Snabre P and Mills P. Settling of a suspension of hard particles. *Europhys Lett* **25**:651–656
580 (1994).
- 581 36 Matos M, Gutiérrez G, Martínez-Rey L, Iglesias O, and Pazos C. Encapsulation of
582 resveratrol using food-grade concentrated double emulsions : Emulsion characterization and
583 rheological behaviour. *J Food Eng* **226**:73–81 (2018).
- 584 37 Díaz-Ruiz R, Martínez-Rey L, Laca A, Álvarez JR, Gutiérrez G, and Matos M. Enhancing
585 trans-Resveratrol loading capacity by forcing W1/O/W2 emulsions up to its colloidal stability
586 limit. *Colloids Surfaces B Biointerfaces* **193**:111130 (2020).
- 587 38 Sun X, Peng B, and Yan W. Measurement and correlation of solubility of trans-resveratrol in
588 11 solvents at T = (278.2, 288.2, 298.2, 308.2, and 318.2) K. *J Chem Thermodyn* **40**:735–
589 738 (2008).
- 590 39 Frasc-Melnik S, Spyropoulos F, and Norton IT. W1/O/W2 double emulsions stabilised by
591 fat crystals - Formulation, stability and salt release. *J Colloid Interface Sci* Elsevier Inc.;
592 **350**:178–185 (2010).
- 593 40 Márquez AL, Medrano A, Panizzolo LA, and Wagner JR. Effect of calcium salts and
594 surfactant concentration on the stability of water-in-oil (w / o) emulsions prepared with
595 polyglycerol polyricinoleate. *J Colloid Interface Sci* Elsevier Inc.; **341**:101–108 (2010).
- 596 41 Jiang J, Mei Z, Xu J, and Sun D. Effect of inorganic electrolytes on the formation and the

- 597 stability of water-in-oil (W/O) emulsions. *Colloids Surfaces A Physicochem Eng Asp Elsevier*
598 B.V.; **429**:82–90 (2013).
- 599 42 Wilson R, Schie B Van, and Howes D. Overview of the preparation, use and biological
600 studies on polyglycerol polyricinoleate (PGPR). *Food Chem Toxicol* **36**:711–718 (1998).
- 601 43 Wolf F, Koehler K, and Schuchmann H. Stabilization of water droplets in oil with PGPR for
602 use in oral and dermal applications. *J Food Process Eng* **36**:276–283 (2013).
- 603 44 Do-Yeong K and Weon-Sun S. Roles of Fucoidan, an Anionic Sulfated Polysaccharide on
604 BSA-Stabilized Oil-in-Water Emulsion. *Macromol Res* **17**:128–132 (2009).
- 605 45 Formulation Cosmetics. Stability and end-use properties of personal care products.
606 Application paper: Turbiscan. 2009.
- 607 46 Formulation Food. Stability of various beverage emulsions. Application paper: Turbiscan.
608 2009.
- 609 47 Matos M, Gutiérrez G, Iglesias O, Coca J, and Pazos C. Enhancing encapsulation efficiency
610 of food-grade double emulsions containing resveratrol or vitamin B12 by membrane
611 emulsification. *J Food Eng Elsevier Ltd*; **166**:212–220 (2015).
- 612 48 Matos M, Marefati A, Gutiérrez G, Wahlgren M, and Rayner M. Comparative Emulsifying
613 Properties of Octenyl Succinic Anhydride (OSA) -Modified Starch : Granular Form vs
614 Dissolved State. *PLoS One* **11**:1–16 (2016).
- 615 49 Khadem B and Sheibat-Othman N. Modeling of double emulsions using population balance
616 equations. *Chem Eng J Elsevier*; **366**:587–597 (2019).
- 617 50 Kang Z, Zhu P, Kong T, and Wang L. A Dewetting Model for Double-Emulsion Droplets A
618 Dewetting Model for Double-Emulsion Droplets. *Microfluidics* **7**:196 (2016).
- 619 51 Khadem B, Khellaf M, and Sheibat-othman N. Investigating swelling-breakdown in double
620 emulsions. *Colloids Surfaces A Elsevier*; **585**:124181 (2020).
- 621 52 Almarri F, Haq N, Alanazi FK, Mohsin K, Alsarra IA, Aleanizy FS, and Shakeel F. Solubility
622 and thermodynamic function of vitamin D3 in different mono solvents. *J Mol Liq Elsevier*

623 B.V.; 229:477–481 (2017).

624 **Figure Legends**

625 Figure 1. Droplet size distribution of W_1/O emulsions.

626 Figure 2. Droplet size distribution of the double $W_1/O/W_2$ emulsions formulated with an internal phase
627 of W_1/O emulsion in fixed ratio 20/80 (A) or 30/70 (B) varying the encapsulated biocompound.

628 Figure 3. Influence on morphology according to the type of emulsion prepared with 20/80 ratio of
629 internal phase at 40x: (A) Double emulsion without encapsulated compound; (B) Double emulsion
630 containing RSV; (C) Double emulsion containing VitD₃; (D) Double emulsion with RSV and VitD₃.

631 Figure 4. Influence on morphology according to the type of emulsion prepared with 30/70 ratio of
632 internal phase at 40x: (A) Double emulsion without encapsulated compound; (B) Double emulsion
633 containing RSV, (C) Double emulsion containing VitD₃; (D) Double emulsion with RSV and VitD₃.

634 Figure 5. Influence of time (weekly evolution) on droplet size distribution of double emulsions with
635 internal phase of 20/80 ratio (left column) or 30/70 ratio (right column) without encapsulated
636 biocompound (A and B), with encapsulated RSV (C and D), with encapsulated VitD₃ (E and F) and
637 with encapsulated RSV and VitD₃ (G and H).

638 Figure 6. Evolution of TSI values during 45 days for double emulsions formulated with a fixed
639 volumetric ratios of W_1/O dispersed in W_2 (80/20) with different proportion of W_1 dispersed in O:
640 20/80 (A) or 30/70 (B) and different encapsulated biocompounds.

641 Figure 7. Clarification height observed during 45 days for samples with internal phase ratio of 20/80
642 (A) and internal phase ratio of 30/70 (B)

643 Figure 8. Flow curves of emulsion with W_1/O ratio of 20/80 (A) and 30/70 (B)

644 Figure 9. Elastic modulus (G') and viscous modulus (G'') versus frequency for double emulsions
645 $W_1/O/W_2$ formulated with an internal phase of W_1/O of 20/80 (A) and 30/70 (B)

646 Figure 10. (A) Encapsulation efficiency and (B) resveratrol loaded concentration for emulsions with
647 an internal phase concentration W_1/O of 20/80 or 30/70.

648 Table 1. Zeta potential and Span values of formulated double emulsions.

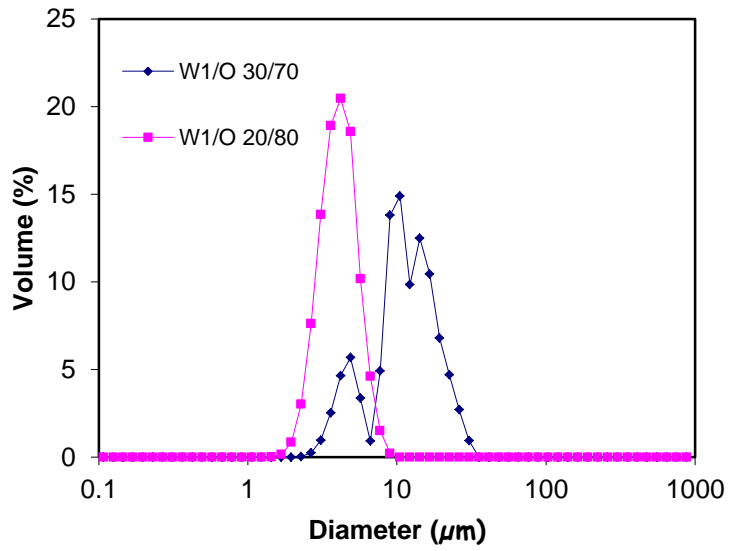


Figure 1

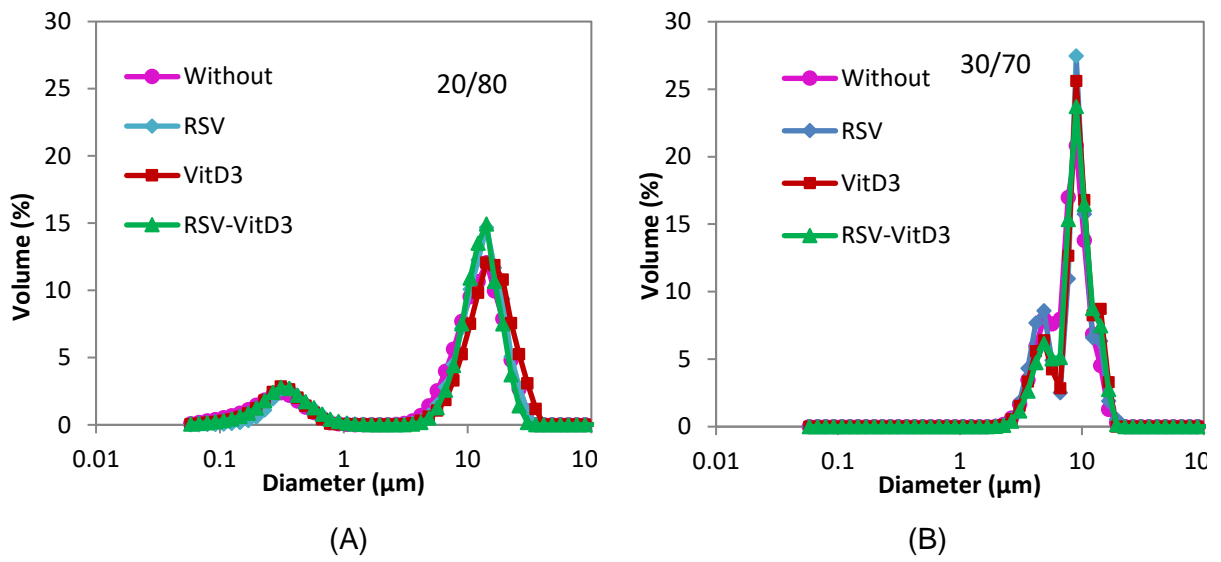
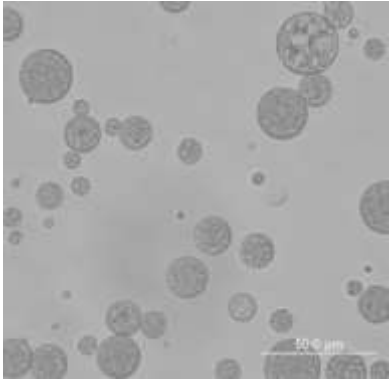
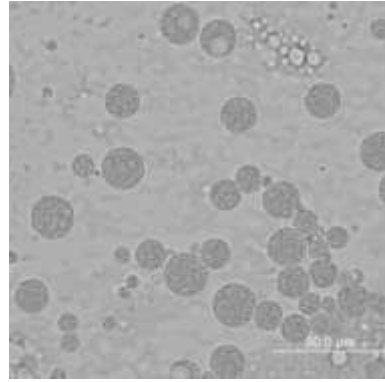


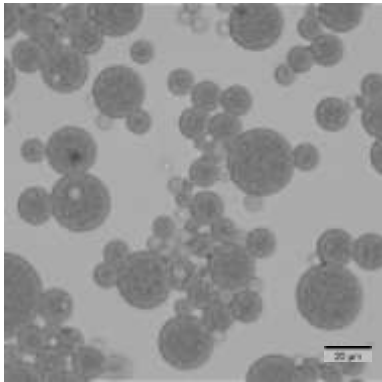
Figure 2



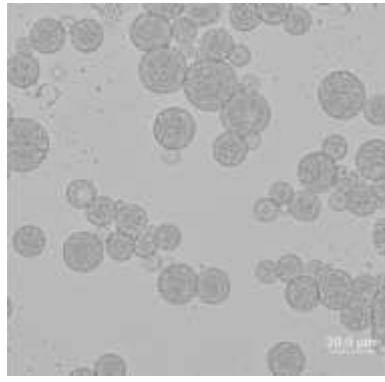
(A)



(B)

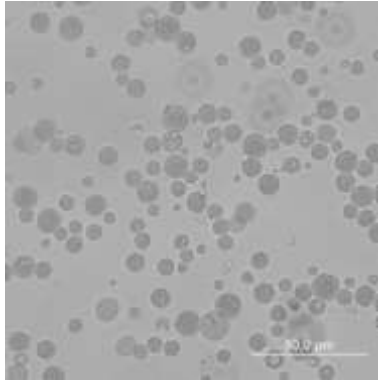


(C)

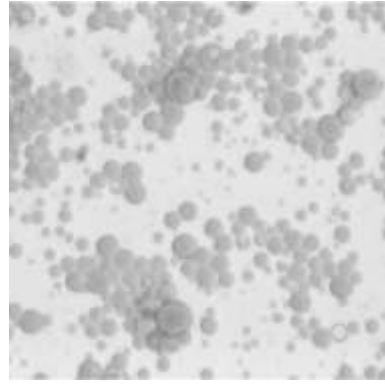


(D)

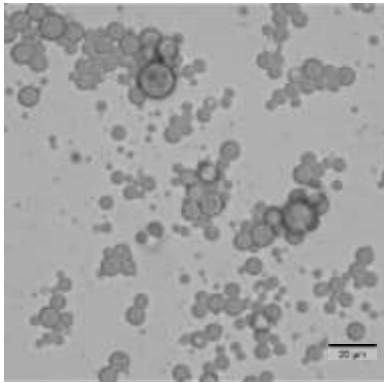
Figure 3



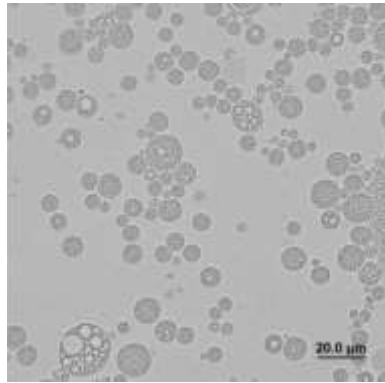
(A)



(B)



(C)



(D)

Figure 4

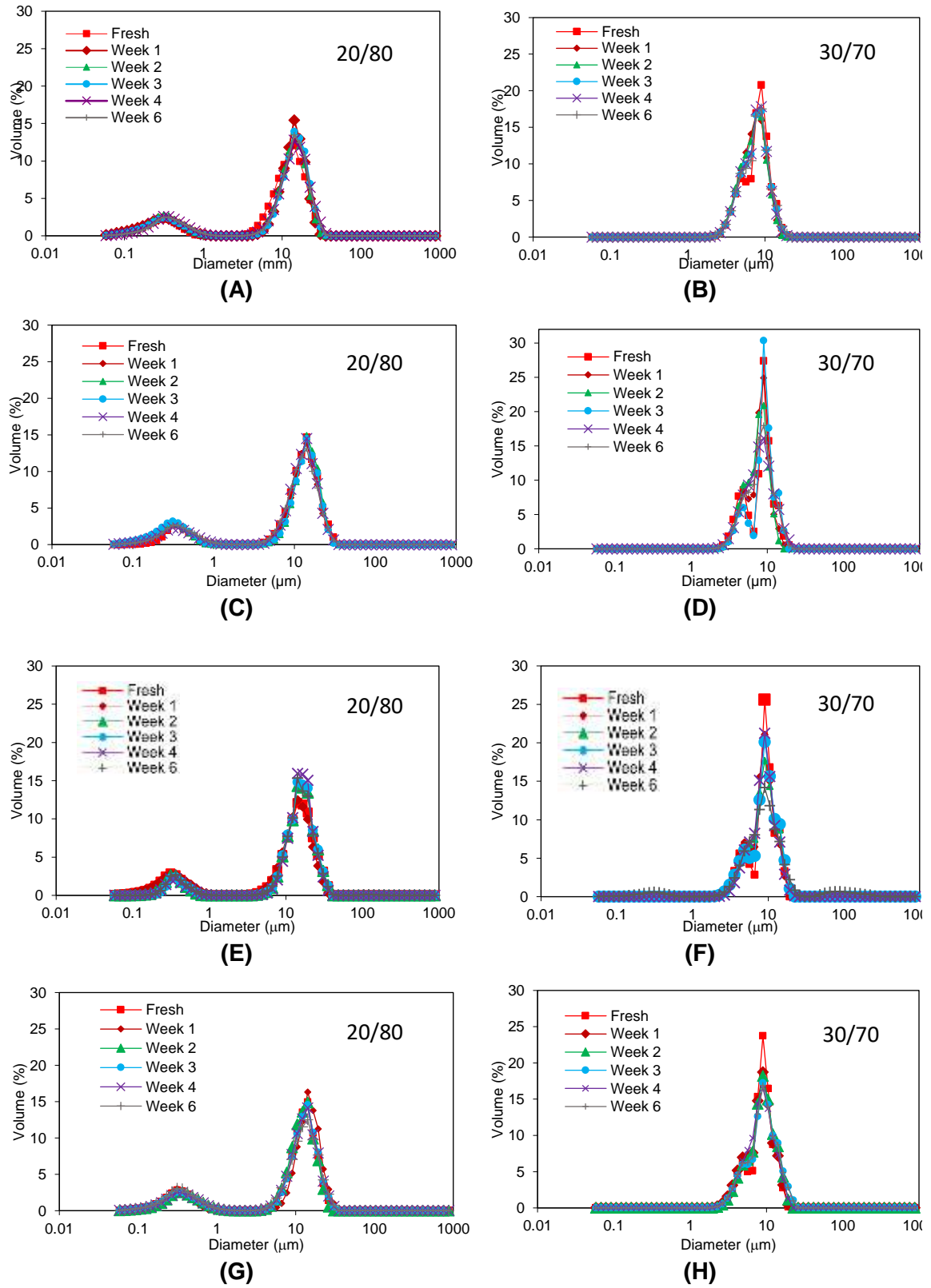


Figure 5

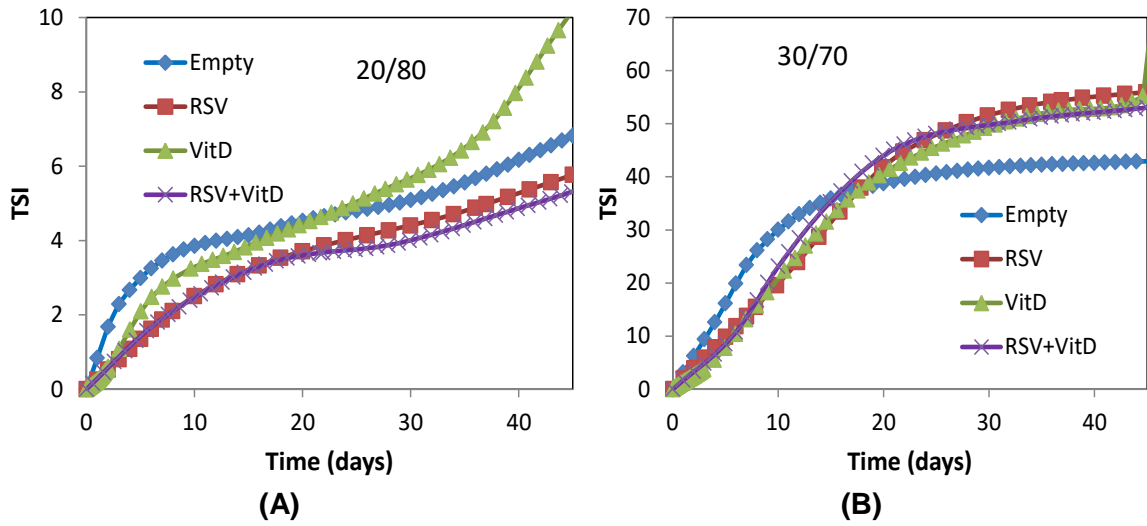


Figure 6

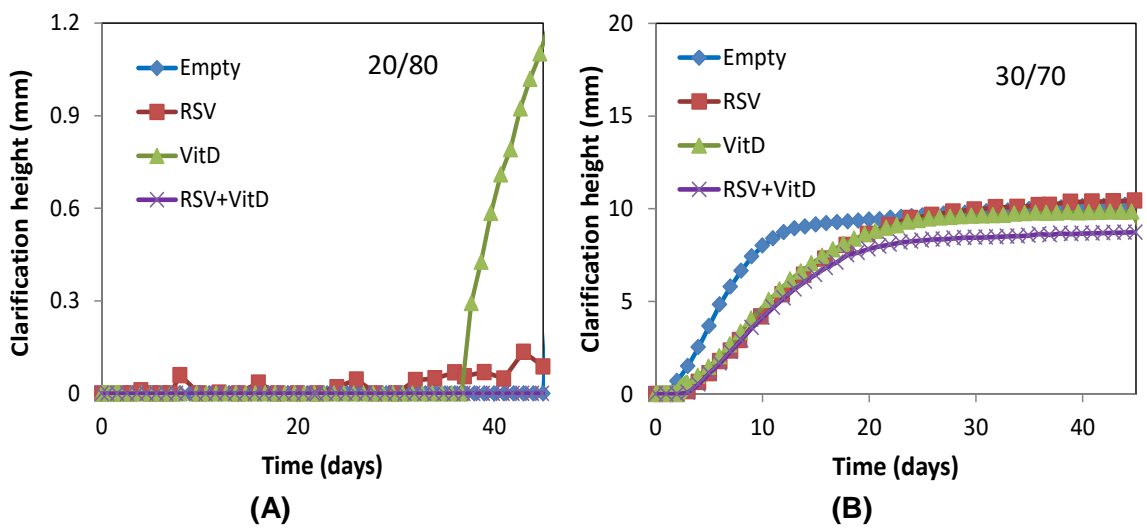


Figure 7

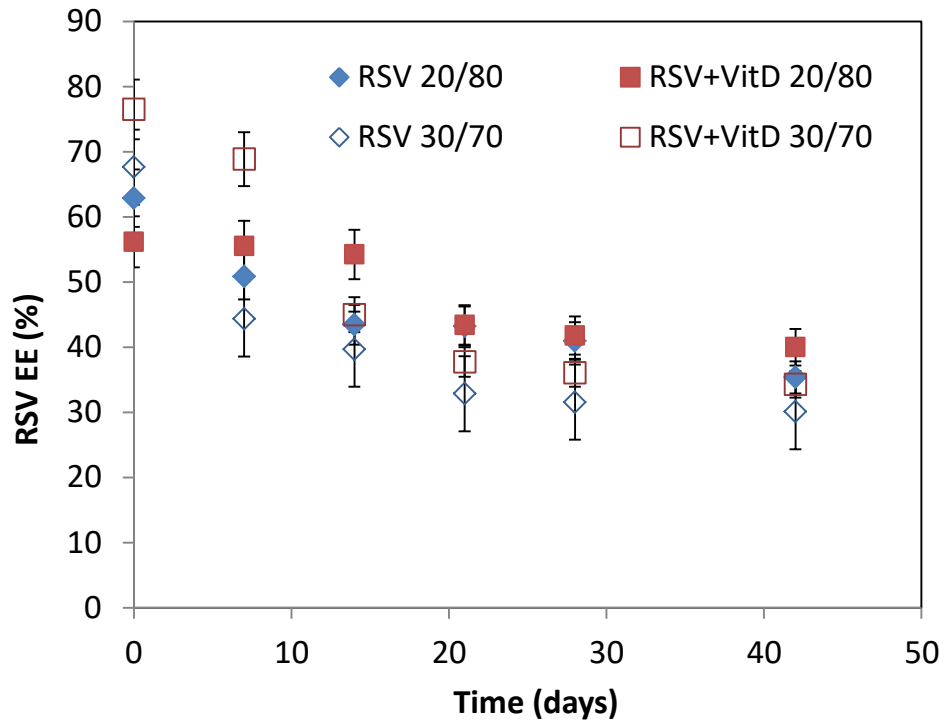


Figure 8

Preparation of water-in-oil-in-water ($W_1/O/W_2$) concentrated double emulsions containing biocompounds of different nature: trans-resveratrol and vitamin D₃

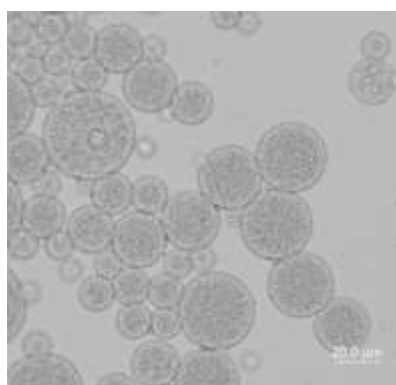
Rocío DÍAZ-RUIZ^a (uo219049@uniovi.es), Irene VALDEÓN^a (UO237168@uniovi.es),

José Ramón ÁLVAREZ^a (iras@uniovi.es), María MATOS^a (matosmaria@uniovi.es)

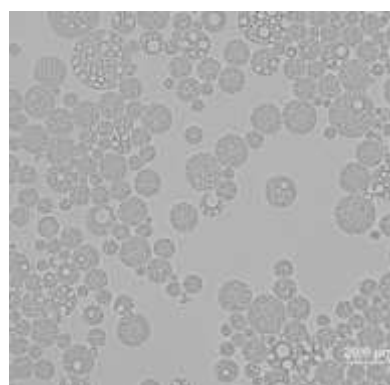
Gemma GUTIÉRREZ^{1,a} (gutierrezgemma@uniovi.es),

^a Department of Chemical and Environmental Engineering, University of Oviedo, Julián Clavería 8, 33006 Oviedo, Spain.

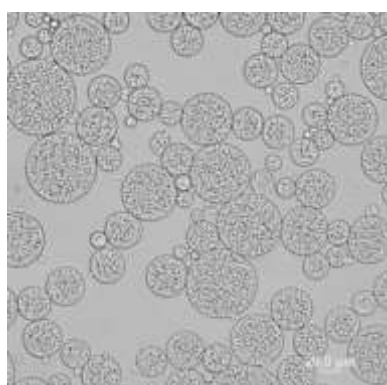
Tel: +34 985103509; Fax: +34 985103434



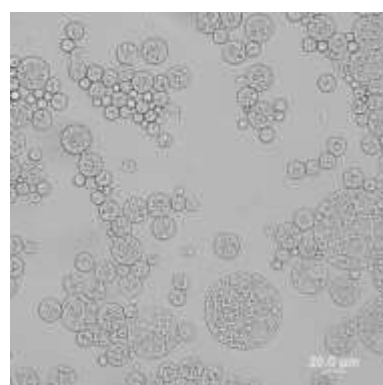
(A)



(B)

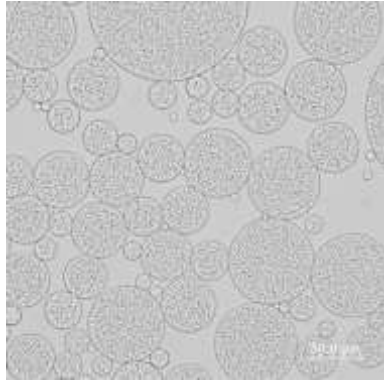


(C)

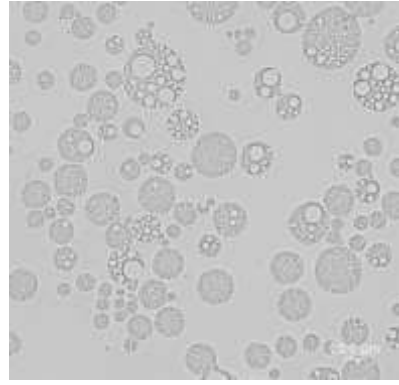


(D)

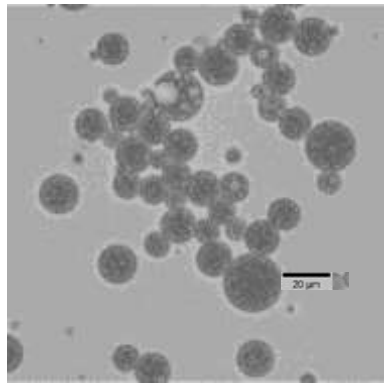
¹ Corresponding author email: gutierrezgemma@uniovi.es



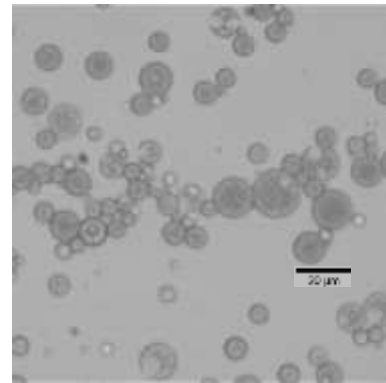
(E)



(F)

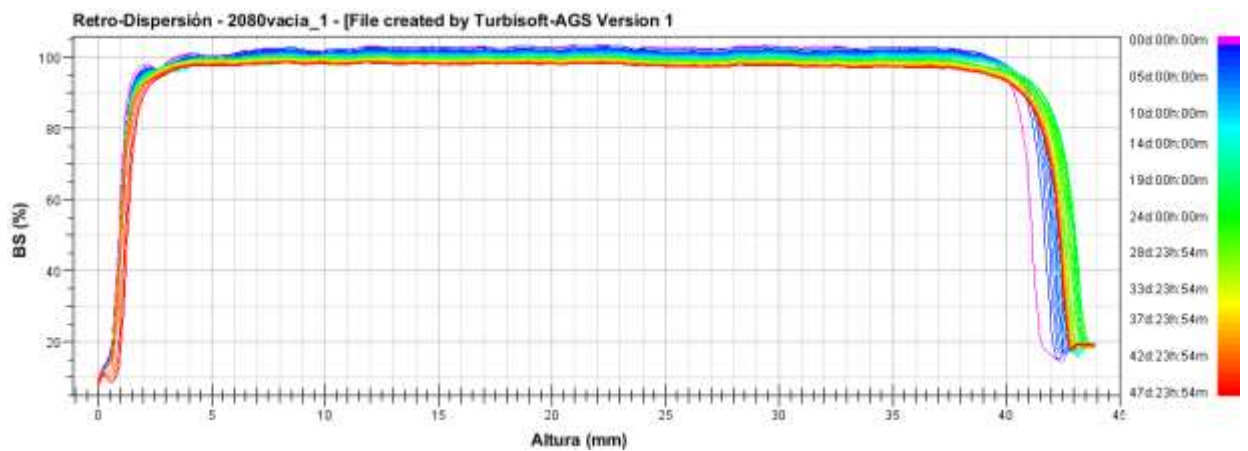


(G)

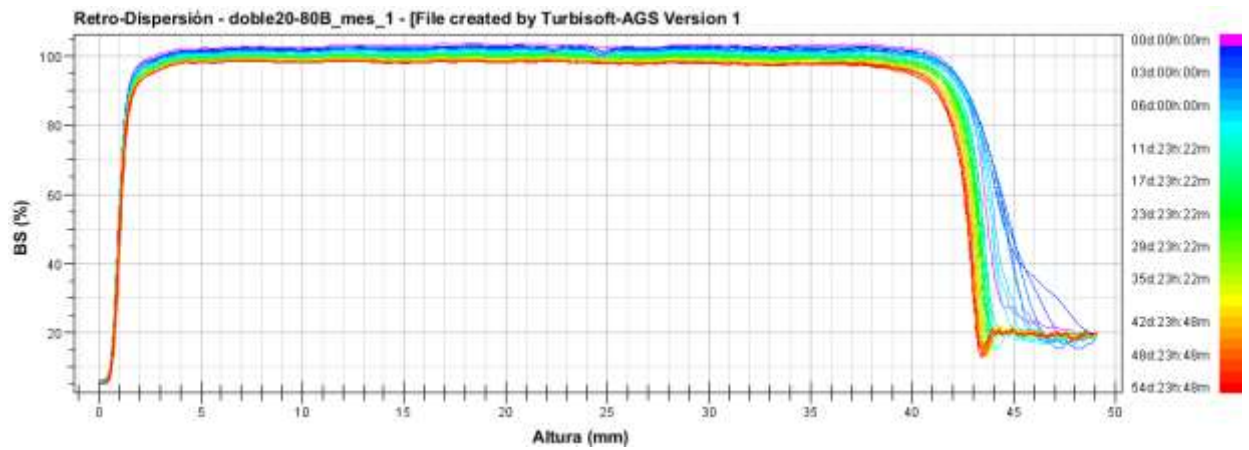


(H)

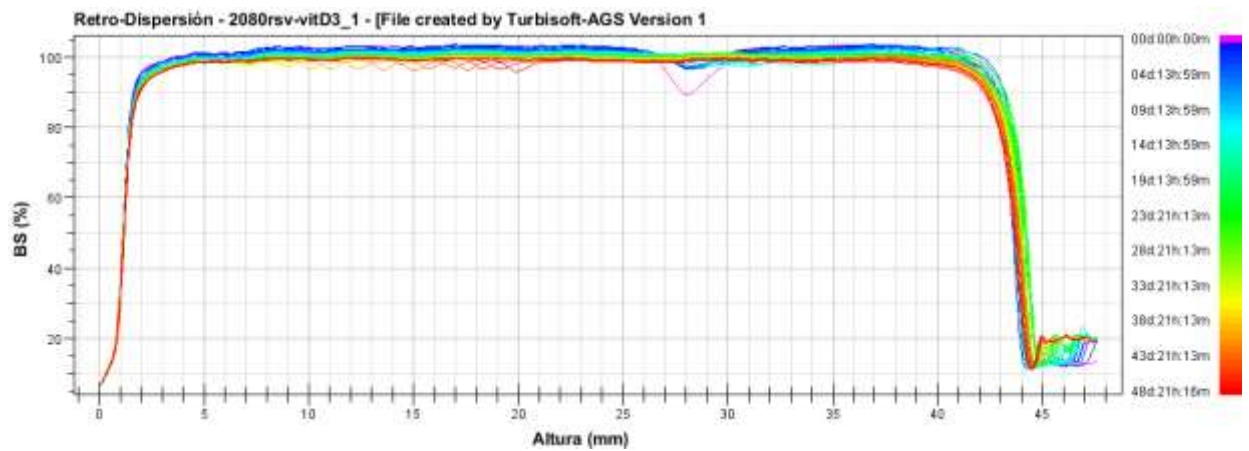
Figure 1S. Optical image of emulsions formulated after 6 weeks, for internal phase ratio 20/80 (left column) and 30/70 (right column) without encapsulated biocompound (A and B), with encapsulated RSV (C and D), with encapsulated VitD₃ (E and F) and with encapsulated RSV and VitD₃ (G and H).



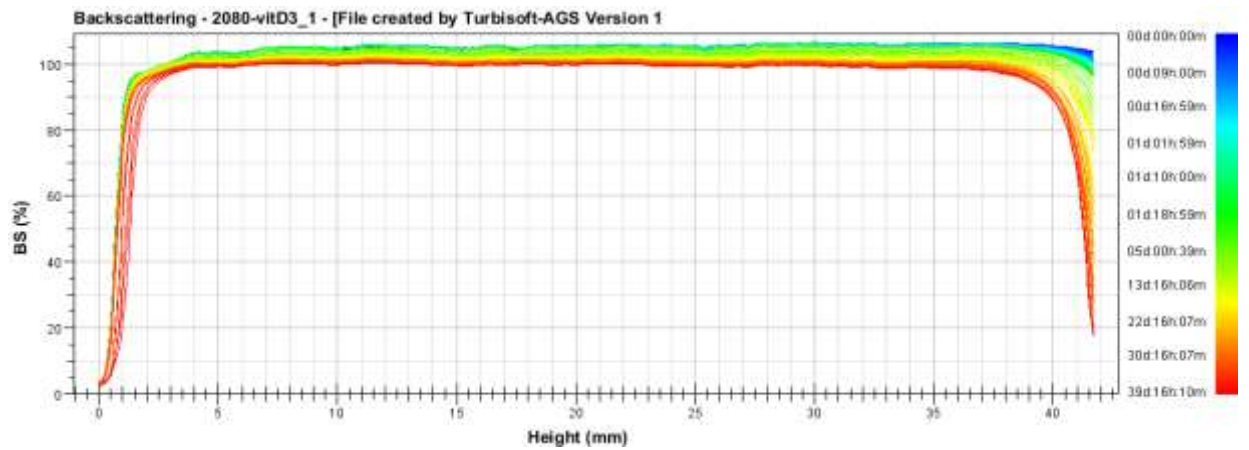
(A)



(B)

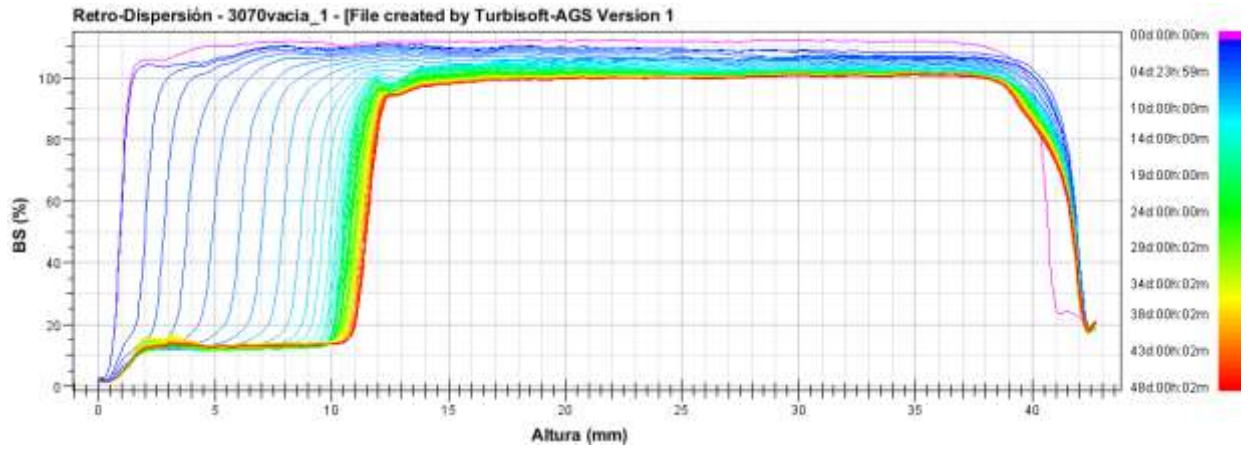


(C)

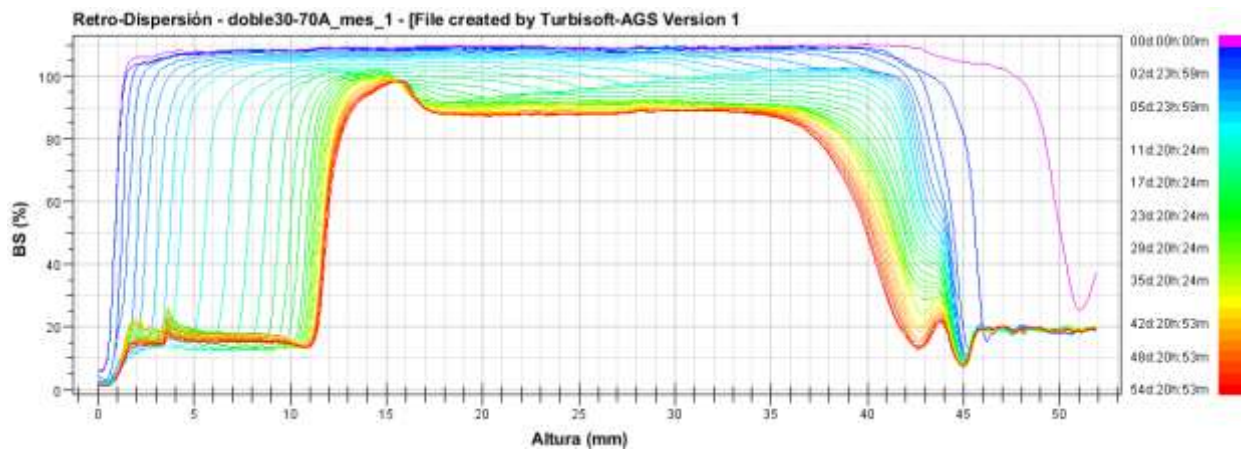


(D)

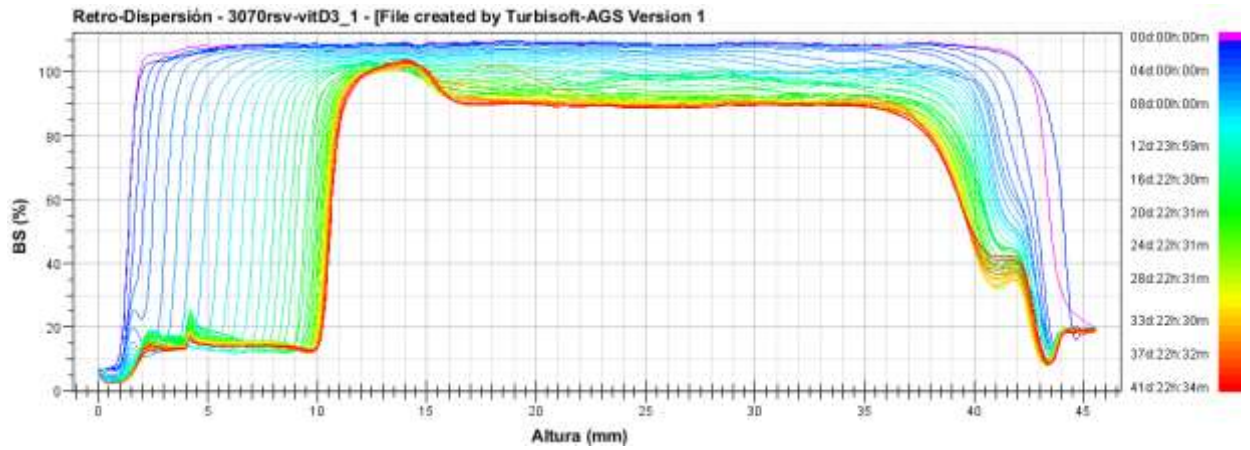
Figure 2S. Stability of the 20/80 W_1 to O ratio emulsions: (A) Double emulsion without encapsulated biocompound; (B) Double emulsion with resveratrol; (C) Double emulsion with resveratrol and vitamin D_3 ; (D) Double emulsion with vitamin D_3 .



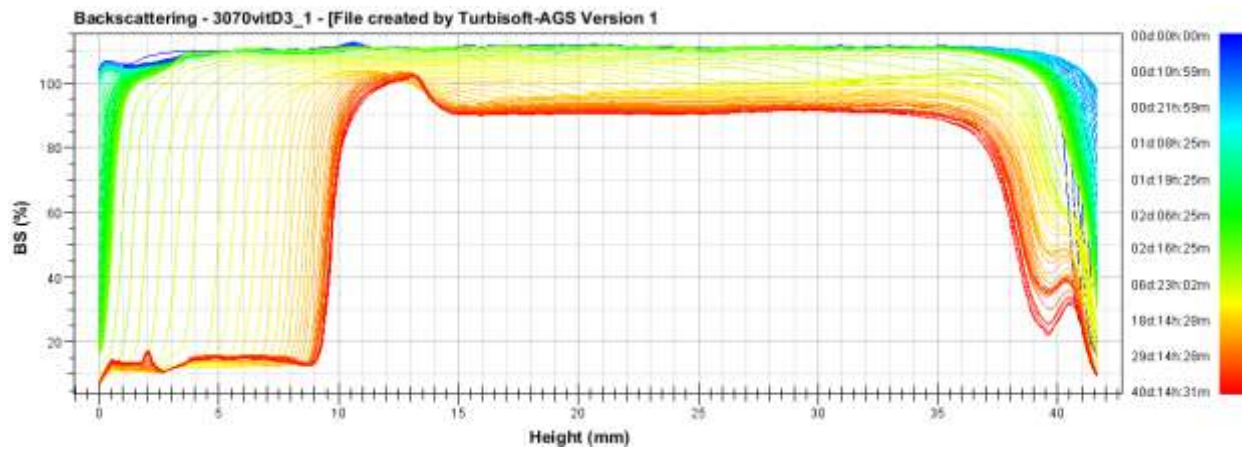
(A)



(B)



(C)



(D)

Figure 3S. Stability of the 30/70 W_1 to O ratio emulsions: (A) Double emulsion without encapsulated biocompound; (B) Double emulsion with resveratrol; (C) Double emulsion with resveratrol and vitamin D_3 ; (D) Double emulsion with vitamin D_3 .